

MEDDELELSE R OM GRØNLAND

UDGIVNE AF

KOMMISSIONEN FOR VIDENSKABELIGE UNDERSØGELSER I GRØNLAND

B.D. 192 · NR. 5

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GRØNLANDS GEOLOGISKE UNDERSØGELSE

BULLETIN No. 103

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THE GEOLOGY AND PETROLOGY OF  
THE PRECAMBRIAN ROCKS  
TO THE NORTH-EAST OF THE FJORD  
QAGSSIT, FREDERIKSHÅB DISTRICT,  
SOUTH-WEST GREENLAND

BY

GIORGIO RIVALENTI AND ANTONIO ROSSI

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WITH 24 FIGURES AND 16 TABLES IN THE TEXT,  
AND 11 PLATES

KØBENHAVN

C. A. REITZELS FORLAG

BIANCO LUNOS BOGTRYKKERI A/S

1972

### **Abstract**

The area consists of migmatitic gneisses (with related pegmatites) containing small inclusions and larger concordant layers of unmigmatised rocks (mainly represented by amphibolites, metasediments and ultramafics), and a few discordant amphibolite dykes. The petrography of the various lithotypes is described and their petrogenesis discussed. The following metamorphic history is proposed: 1) an increase in metamorphic grade to the biotite-almandine-cordierite subfacies of the low hornblende-granulite facies, shown only by a few relic assemblages, followed by 2) retrogression to within the almandine-amphibolite facies. Migmatisation (quartz dioritisation or granodioritisation) started in the low hornblende-granulite facies and reached its maximum in the upper almandine-amphibolite facies. A general microcline blastesis and formation of microcline-rich mobilisates is ascribed to the medium to low almandine-amphibolite facies. Further retrogression to epidote-amphibolite or greenschist facies is only local and weak.

Several generations of post-orogenic basic dykes intersect the gneisses.

Structurally, five phases of ductile deformation and three phases of brittle deformation are distinguished, and the relationship between deformation and metamorphism tentatively established.

The metamorphic and structural events are mostly pre-Ketilidian.

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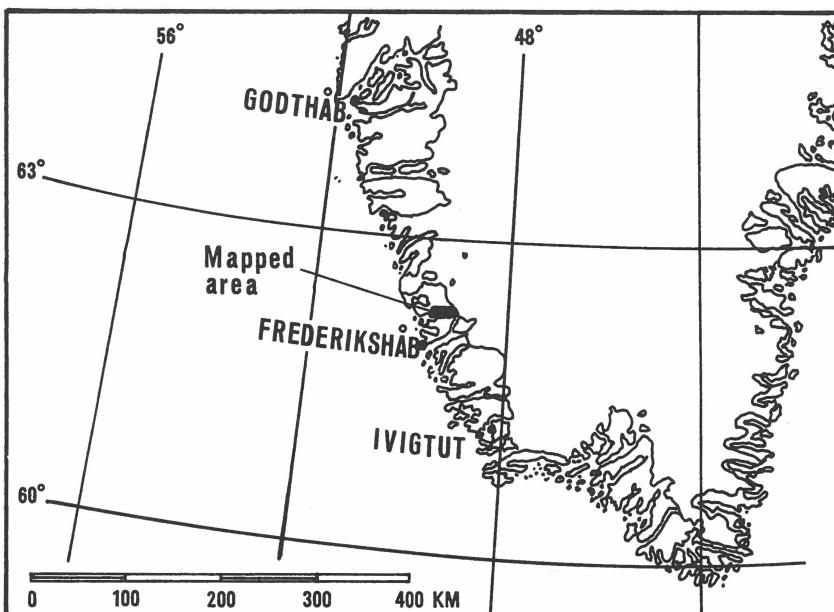


Fig. 1. Map locality in Greenland.

## INTRODUCTION

During the summer months of 1967 and 1968 the authors had the chance of joining the GGU (Grønlands Geologiske Undersøgelse) geological field parties in the Mellembygd area, Frederikshåb district, southwest Greenland. The area mapped lies about 30 km NNE of Frederikshåb and consists of about 300 km<sup>2</sup>, from the fjord Qagssip kangerdluarssua to the ice cap (fig. 1). Laboratory investigations were carried out at the Istituto di Mineralogia e Petrologia of the University of Modena, Italy. In 1967 the two authors mapped the northern part of the Aorngo area together; in 1968 A. Rossi mapped the southern part of the same area and G. RIVALENTI mapped the western area and the peninsula north of Sermilik avangnardleq. The chemical and some of the modal analyses were carried out by A. Rossi, while the microscopic observations, the structural interpretation, the writing and conclusions of this paper are the work of G. RIVALENTI.

The area mapped belongs to the Canadian-Greenlandic Precambrian Shield and has undergone a complicated metamorphic and structural evolution. The chronological division of the Precambrian of South Greenland can be found in BRIDGWATER (1965). The area consists of a migmatitic complex consisting of the following lithotypes: 1) migmatitic

gneisses and pegmatites, 2) concordant layers and inclusions in the gneisses, 3) discordant amphibolite dykes. These rocks are crossed by post-orogenic basic dykes which continue further south where, in the Ivigtut area, they are deformed by Ketilidian structures. Therefore, the metamorphic and structural evolution can be ascribed to the pre-Ketilidian (> 2500 m.y.).

In the following pages the field characteristics, petrography, structure, metamorphism and relationships of the various lithotypes will be described and discussed.

# THE MIGMATITIC GNEISSES AND PEGMATITES

## Introduction

Migmatitic gneisses and related pegmatites are the most widespread lithotypes of the area. In this paper, to avoid any misunderstandings with terminology, the gneiss classification submitted in Copenhagen at the symposium on migmatite nomenclature (BERTHELSEN, 1960) and recommended by GGU is adopted. This classification, which is based only on structural considerations, distinguishes: banded gneiss, streaky gneiss, veined gneiss, small-folded gneiss, nebulitic gneiss, homogeneous gneiss, and agmatitic gneiss. All these types are represented in this area, but the veined type predominates (plate 11). Since there is a close relationship between pegmatites and migmatisation these two subjects are considered together.

## The Gneisses

### Description

#### **Veined gneiss**

Veined gneisses consist of an irregular alternation of thin felsic layers, in which quartz and feldspar predominate, with more basic layers which contain a large amount of mafic minerals (biotite and hornblende). They also contain both single and branching acid veins in any attitude. The veined gneiss passes into the banded type as soon as the alternation of acid and basic layers in more regular and the single layers become more consistent along the strike. No sharp boundary can normally be traced between these two types. In many cases, the veined gneiss grades into streaky gneiss containing schlieren of amphibolite and hornblendite.

In many places, such as the areas NNE of mountain 933 m, NW and NE of mountain 880 m (plate 11), this gneiss has a small-folded appearance (fig. 20). Feldspar porphyroblasts are commonly seen in the acid veins. When these porphyroblasts are the dominant characteristic of the rock, the gneiss has been indicated on the map as 'porphyroblastic gneiss' (see for instance the area of mountain 880 m in the eastern part).



Fig. 2. Migmatitic paleosome embedded in a migmatitic neosome. Border of the Nigerdlip qôrorssua homogeneous gneiss.

In some points this gneiss has a polymigmatitic character (see e.g. figs 2 and 3, which is shown by a migmatitic paleosome embedded in a migmatitic neosome).

Mineralogically, three main lithotypes can be distinguished: biotite veined gneiss, hornblende veined gneiss, and garnetiferous gneiss. Complete transitions occur between these types and in many cases the intermixing is so complete that it has foiled attempts to map them as different lithotypes. The only boundary that has been traced is of garnetiferous gneiss in the area between the fjord Qagssip kangerdluarssua and the valley Nigerdlip qôrorssua and even this is imprecise because it is gradational. In the garnetiferous gneiss layering and veining is very weakly developed and its structure is more homogeneous than that of the veined gneiss. The hornblende gneiss often represents one of the stages in the migmatisation of the amphibolites. In the field it is possible to follow the passage from a normal amphibolite to an agmatite or

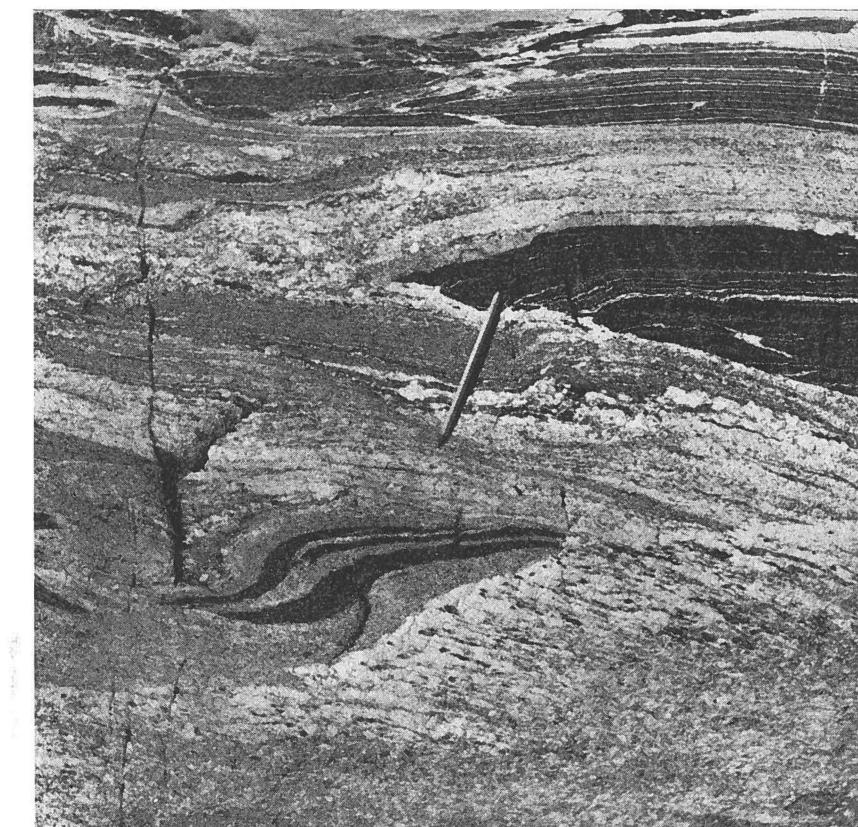


Fig. 3. Migmatitic paleosome, rich in amphibolite inclusions, passing into an inclusion-free, coarse-grained migmatitic neosome. Border of the homogeneous gneiss of the peninsula north of Sermilik avangnardleq.

streaky gneiss, and to a hornblende gneiss in which the amount of hornblende decreases and the felsic material increases; the last product of transformation may be a biotite gneiss in which no or very few relics of hornblende are present.

#### Banded gneiss

In general the banded gneiss consists of a regular alternation of felsic and mafic bands. Two types can, however, be distinguished on the basis of the melanocratic part: A) melanocratic bands which have a mineral assemblage similar to that of the felsic, but are enriched in biotite (and/or hornblende); B) melanocratic bands which are composed of amphibolites. *Type A* is a local, more regularly banded variety of the veined type. It is encountered sporadically all over the region; for instance the whole area between Qagssit, Qingua and mountain 825 m is formed of a gneiss with a fairly regular banded structure. The banding



Fig. 4. Homogeneous gneiss of the peninsula north of Sermilik avangnardleq. The amphibolite inclusions are almost completely transformed. 750 m east of mountain 920 m.

can vary from a few millimetres up to three metres. *Type B* commonly occurs as a transitional rock at the contact of the amphibolite horizons with the surrounding gneisses. The layering varies in thickness from a centimetre to half a metre and consists of amphibolite bands and biotite (hornblende) gneiss. The contacts are macroscopically sharp.

#### Homogeneous gneiss

The main fabric in this type is formed by the planar preferred orientation of mica and hornblende crystals. The small-folding present in the preceding types does not occur here. In the present area the main outcrops are represented by the garnet-bearing gneiss of the Nigerdlip qôrорssua area, by the gneiss west of mountain 780 m and by the gneiss of Sermilik avangnardleq. The first two, however, are cut by a few veins and at certain points show a faint lithological banding.

The garnet-bearing gneiss of the Nigerdlip qôrорssua area is bordered

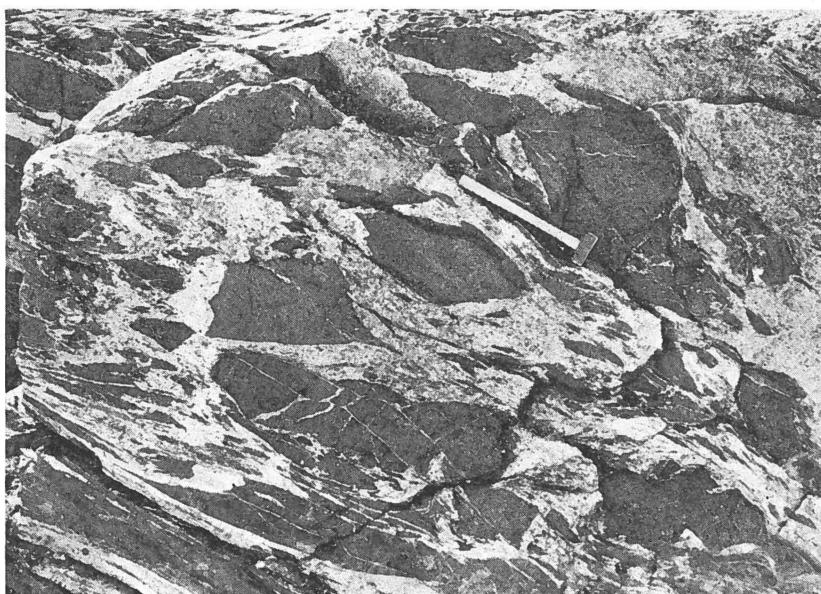


Fig. 5. Agmatitic gneiss. 100 m east of lake 360 m.

towards the north-west by a very fine-grained nebulitic quartz-rich gneiss and towards the east by a veined gneiss. At the contact with the veined gneiss polymigmatisation phenomena have been found (fig. 2). The homogeneous gneiss completely lacks any amphibolite inclusions and major amphibolite horizons are also absent.

The homogeneous gneiss of the area west of mountain 780 m is composed of a biotitic, often hornblendic, but rarely garnet-bearing, medium-grained rock, still rich in schlieren of amphibolite and hornblendite and sporadically intersected by thin pegmatitic bodies. It has gradational contacts with the surrounding veined type.

The homogeneous gneiss of the peninsula north of Sermilik avang-nardleq comprises a well-defined band, up to 500 m thick, which is conformable with the veined type. It is a medium- to coarse-grained rock. In some places, inclusions are almost absent and the rock resembles a homogeneous granite. In the field it is possible to follow the passage from a migmatitic gneiss rich in mafic inclusions (veined-streaky gneiss) to the homogeneous gneiss in which the inclusions are almost completely 'digested', and the hornblende of the veined gneiss is completely replaced by biotite (see figs 3 and 4). The passage can be gradual or abrupt.

#### Agmatitic gneiss

Agmatites are very common all over the area. They represent one of the transitional products formed during the migmatisation of the

amphibolites. They consist of breccia-like rocks in which amphibolitic paleosome blocks are embedded in a gneissic or pegmatitic neosome (fig. 5). The characteristics of the amphibolite remnants are referred to on p. 53. The neosome presents the same characteristics as the gneisses and pegmatites described in this paper, therefore their description will not be repeated here.

### Petrography

Although the various structural types have many points in common, a short description will be given for each of them to point out the differences.

#### *Veined gneiss*

The texture is generally crystalloblastic. It becomes porphyroblastic when big feldspar porphyroblasts are present. When large amounts of garnet are present it may exceptionally be granoblastic. The three main mineral assemblages are as follows.

#### *Biotite gneiss*

The only and most prominent mafic mineral is biotite. The typical assemblage is composed of quartz, plagioclase, biotite, microcline and muscovite as the major minerals with epidote, sphene, apatite, zircon and opaque minerals as accessory minerals. Modal analyses of typical rocks belonging to this group are given in Table 1.

*Quartz* is normally very abundant; it may be in greater or smaller quantity than the feldspar. It consists of at least two generations. The first generation is the most abundant and is replaced by the second generation which shows absorption contacts towards all the other minerals. This is particularly evident in the case of feldspar towards which it may show highly indented rims like those shown in plate 1a. The saw-like textures mainly occur when the polysynthetic feldspar twins form a wide angle to the contact with the secondary quartz. When the angle is low, the contact is usually lobate. Quartz is also present as a reaction product of some minerals (described later on) where it forms small patches in the minerals or assumes vermicular shapes, as in the case of the quartz-epidote symplectites.

*Plagioclase* is normally present in large amounts. Usually it is fresh, but sometimes it is altered to sericite, muscovite, calcite and epidote with the consequent formation of albite. In many instances, when this alteration is very advanced, it is possible to relate it to a local retrogression subsequent to fracturing and folding. The plagioclase compositions vary from acid to intermediate oligoclase (An 20–30 %). Frequently a normal zoning is present, with an irregular, more sodic, peripheral rim. The zoning emphasises the alteration which preferentially affects the core of the crystal. Plagioclase is almost invariably twinned. In the sections examined it has been found that the pericline law predominates over the albite law. Exceptionally, the Carlsbad, Manebach and other more complex twinnings are also present. Plagioclase occurs both as equant crystals, giving the rock an interlobate, seldom amoeboidal, eugranoblastic texture, or as big porphyroblasts (porphyroblastic gneiss)

growing along the acid veins. The size of the porphyroblasts, constantly formed of oligoclase with minor amounts of microcline, may reach a maximum of several centimetres. The biotite is pushed apart and wraps around them.

Table 1. *Chemical and modal analyses of gneisses.*

| GGU No                              | Veined gneiss  |        |       |                |        |                      |       | Banded gneiss             |  |
|-------------------------------------|----------------|--------|-------|----------------|--------|----------------------|-------|---------------------------|--|
|                                     | Biotite gneiss |        |       | Hornbl. gneiss |        | Garnetiferous gneiss |       |                           |  |
|                                     | 57724          | 57873  | 57886 | 57723          | 57737  | 57738                | 57715 |                           |  |
| Oxides wt. %                        |                |        |       |                |        |                      |       |                           |  |
| SiO <sub>2</sub> ....               | 69.72          | 72.05  | 71.65 | 58.43          | 60.56  | 67.40                | 62.25 | 70.82                     |  |
| TiO <sub>2</sub> ....               | 0.33           | 0.14   | 0.34  | 0.96           | 0.85   | 0.51                 | 0.71  | 0.39                      |  |
| Al <sub>2</sub> O <sub>3</sub> .... | 15.95          | 14.78  | 15.41 | 17.36          | 16.75  | 15.45                | 16.45 | 15.41                     |  |
| Fe <sub>2</sub> O <sub>3</sub> .... | 0.77           | 0.90   | 0.79  | 1.70           | 1.89   | 0.38                 | 1.71  | 1.39                      |  |
| FeO ....                            | 1.33           | 0.45   | 1.19  | 5.22           | 4.29   | 6.40                 | 4.67  | 0.78                      |  |
| MnO ....                            | 0.03           | 0.01   | 0.01  | 0.15           | 0.10   | 0.16                 | 0.21  | 0.01                      |  |
| CaO ....                            | 4.06           | 3.15   | 3.08  | 8.27           | 6.03   | 3.79                 | 5.54  | 3.12                      |  |
| MgO ....                            | 1.26           | 0.71   | 0.67  | 3.02           | 4.03   | 1.56                 | 2.87  | 1.16                      |  |
| Na <sub>2</sub> O ....              | 4.40           | 5.92   | 4.26  | 3.20           | 3.51   | 2.52                 | 3.16  | 4.10                      |  |
| K <sub>2</sub> O ....               | 0.83           | 1.48   | 1.24  | 0.29           | 0.96   | 0.25                 | 0.68  | 1.43                      |  |
| P <sub>2</sub> O <sub>5</sub> ....  | 0.03           | 0.10   | 0.03  | 0.07           | 0.13   | 0.03                 | 0.08  | 0.05                      |  |
| H <sub>2</sub> O <sup>+</sup> ....  | 0.55           | 0.68   | 0.57  | 0.85           | 0.90   | 0.76                 | 1.10  | 0.59                      |  |
| H <sub>2</sub> O <sup>-</sup> ....  | 0.01           | 0.03   | 0.06  | 0.02           | 0.02   | 0.01                 | 0.05  | 0.07                      |  |
|                                     | 99.27          | 100.40 | 99.30 | 99.54          | 100.02 | 99.22                | 99.48 | 99.32                     |  |
| Mode vol. %                         |                |        |       |                |        |                      |       |                           |  |
|                                     |                |        |       |                |        |                      |       | melano-<br>felsic<br>band |  |
| quartz ...                          | 11.6           | 28.1   | 26.9  | 6.6            | 16.9   | 40.3                 | 16.0  | 31.4                      |  |
| plagioclase                         | 76.1           | 58.2   | 58.9  | 56.4           | 55.7   | 39.2                 | 59.6  | 37.3                      |  |
| microline .                         | 0.4            | 5.2    | 0.3   | —              | —      | 0.1                  | —     | 27.6                      |  |
| biotite ...                         | 8.5            | 7.4    | 9.4   | 3.1            | 13.4   | 1.5                  | 9.0   | 2.5                       |  |
| garnet....                          | —              | —      | —     | —              | —      | 15.1                 | 4.0   | —                         |  |
| hornblende                          | —              | —      | —     | 30.8           | 12.6   | —                    | 9.1   | —                         |  |
| muscovite                           | 1.0            | 0.9    | 4.3   | —              | —      | 0.7                  | —     | 0.2                       |  |
| chlorite...                         | —              | —      | —     | 0.1            | —      | 2.4                  | 0.5   | —                         |  |
| epidote ...                         | 1.9            | tr     | tr    | 1.4            | 0.3    | —                    | 0.3   | 0.9                       |  |
| sphene ...                          | 0.3            | —      | —     | 1.0            | 0.5    | —                    | —     | —                         |  |
| apatite ...                         | 0.1            | 0.1    | 0.1   | 0.3            | 0.4    | —                    | 0.4   | tr                        |  |
| zircon ....                         | tr             | tr     | tr    | tr             | 0.1    | tr                   | 0.1   | —                         |  |
| calcite ...                         | —              | —      | —     | —              | —      | —                    | 0.7   | —                         |  |
| opaque ...                          | 0.1            | —      | tr    | 0.3            | tr     | 0.6                  | 0.3   | tr                        |  |
| An % in                             |                |        |       |                |        |                      |       |                           |  |
| plag....                            | 29             | 22     | 22    | 38             | 35     | 42                   | 35    | 25                        |  |

57715: CO<sub>2</sub> = 0.66 %.

57879: composed of 60 % melanocratic and 40 % felsic bands, representing the ratio found in the field.

(continued)

Table 1 (continued).

| GGU No.                             | Homogeneous gneiss |       |       |       |       | Polymigmatitic gneiss |       |       |       |
|-------------------------------------|--------------------|-------|-------|-------|-------|-----------------------|-------|-------|-------|
|                                     | 73543              | 57756 | 57757 | 57770 | 57782 | 73816                 | 73817 | 73836 | 73835 |
| Oxides wt. %                        |                    |       |       |       |       |                       |       |       |       |
| SiO <sub>2</sub> ....               | 72.77              | 67.21 | 61.23 | 67.61 | 68.43 | 69.35                 | 63.64 | 70.17 | 67.92 |
| TiO <sub>2</sub> ....               | 0.43               | 0.40  | 0.70  | 0.59  | 0.43  | 0.50                  | 0.62  | 0.36  | 0.53  |
| Al <sub>2</sub> O <sub>3</sub> .... | 14.62              | 15.31 | 18.29 | 16.31 | 16.38 | 15.54                 | 16.37 | 16.05 | 16.56 |
| Fe <sub>2</sub> O <sub>3</sub> .... | 0.29               | 1.02  | 3.01  | 0.88  | 0.91  | 0.62                  | 0.97  | 0.61  | 0.55  |
| FeO ....                            | 2.25               | 2.71  | 1.97  | 1.82  | 1.55  | 2.01                  | 3.86  | 0.92  | 2.33  |
| MnO ....                            | 0.08               | 0.06  | 0.08  | 0.03  | 0.03  | 0.05                  | 0.08  | 0.03  | 0.03  |
| CaO ....                            | 4.42               | 4.35  | 6.33  | 5.82  | 4.20  | 4.56                  | 5.40  | 4.10  | 4.13  |
| MgO ....                            | 1.51               | 2.52  | 1.93  | 0.91  | 1.56  | 1.16                  | 2.57  | 0.60  | 1.51  |
| Na <sub>2</sub> O ....              | 3.06               | 3.80  | 4.40  | 3.80  | 4.52  | 4.12                  | 3.70  | 4.52  | 4.28  |
| K <sub>2</sub> O ....               | 0.28               | 0.96  | 0.41  | 0.69  | 0.50  | 1.03                  | 1.30  | 1.30  | 1.23  |
| P <sub>2</sub> O <sub>5</sub> ....  | 0.07               | 0.06  | 0.05  | 0.04  | 0.04  | 0.04                  | 0.04  | 0.04  | 0.04  |
| H <sub>2</sub> O <sup>+</sup> ....  | 0.27               | 0.78  | 0.82  | 0.80  | 0.58  | 0.55                  | 0.82  | 1.09  | 0.73  |
| H <sub>2</sub> O <sup>-</sup> ....  | 0.02               | 0.11  | 0.06  | 0.08  | 0.08  | 0.07                  | 0.06  | 0.12  | 0.07  |
|                                     | 100.07             | 99.29 | 99.28 | 99.38 | 99.21 | 99.60                 | 99.43 | 99.91 | 99.91 |
| Mode vol. %                         |                    |       |       |       |       |                       |       |       |       |
| quartz....                          | 26.1               | 26.1  | 12.7  | 21.5  | 22.1  | 29.8                  | 27.7  | 28.7  | 24.2  |
| plagioclase                         | 69.1               | 59.3  | 72.7  | 61.9  | 65.8  | 58.2                  | 43.6  | 63.5  | 66.0  |
| microcline                          | —                  | 0.3   | —     | —     | —     | 0.1                   | —     | 0.4   | tr    |
| biotite....                         | 1.6                | 11.8  | 3.2   | 8.3   | 7.2   | 10.7                  | 16.7  | 5.6   | 8.7   |
| garnet....                          | 3.1                | —     | —     | 0.1   | —     | tr                    | —     | tr    | —     |
| hornblende                          | —                  | 2.3   | 7.1   | 6.3   | 2.7   | —                     | 10.0  | —     | —     |
| muscovite.                          | —                  | —     | 1.2   | 0.8   | 0.2   | 0.4                   | —     | —     | —     |
| chlorite...                         | —                  | —     | 0.3   | 0.3   | 0.3   | —                     | —     | 0.8   | —     |
| epidote ...                         | tr                 | tr    | 0.1   | 0.3   | 1.0   | 0.2                   | 1.4   | 0.7   | 0.7   |
| sphene....                          | —                  | —     | 0.1   | —     | —     | tr                    | 0.1   | —     | —     |
| apatite ...                         | tr                 | 0.2   | 0.3   | 0.3   | 0.4   | 0.4                   | 0.4   | 0.2   | 0.3   |
| zircon ....                         | tr                 | tr    | tr    | tr    | tr    | tr                    | tr    | tr    | tr    |
| calcite ....                        | —                  | —     | 0.9   | —     | —     | —                     | —     | —     | —     |
| opaque ...                          | tr                 | tr    | 1.4   | 0.1   | 0.2   | 0.1                   | tr    | —     | —     |
| An % in                             |                    |       |       |       |       |                       |       |       |       |
| plag. ....                          | 34                 | 35    | 34    | 35    | 30    | 29                    | 32    | 30    | 30    |

73543: homogeneous gneiss of Nigerdlip qôrorssua

57756, 57757, 57770, 57782: homogeneous gneiss of the area west of mountain 880 m.

73836: homogeneous gneiss of the peninsula north of Sermilik avangnardleq

73816, 73817: neosome and paleosome respectively of fig. 2

73836, 73835: neosome and paleosome respectively of fig. 3

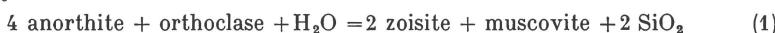
*Microcline* is considered a minor constituent, as it occurs often but never in large amounts. It occurs in two forms: as very thin veins running along the grain boundaries or entering the discontinuity planes of feldspar (cleavages and twinning

planes) and other minerals, or in autonomous crystals clearly showing the typical grid twinning. Microcline shows absorption contacts toward the plagioclase (plate 1b) which is clearly substituted by it, together with the second generation quartz. Sometimes microcline occurs in small patches in the plagioclase. The microcline crystallisation clearly post-dates that of the other minerals.

*Biotite* occurs as thin, subidiomorphic lamellae. Its pleochroism varies from  $\alpha$  = yellow (yellowish green),  $\beta$  =  $\gamma$  = deep green (greenish brown). Biotite is usually fresh, but some alteration products are highly significant from the point of view of its metamorphic history. It may be associated with muscovite, epidote or pass to chlorite. The brown or reddish biotite is fairly uncommon, though sometimes it is met with either as the only mafic mineral, or as small relics in the green biotite. When the green biotite is present, it is accompanied by a net of sagenitic rutile formed with the excess Ti, or by sphene when Ca is available. This is not seen in connection with the brown biotite, which is usually accompanied by ilmenite. This point will be reconsidered in the section concerning the chemistry of the gneisses.

*Epidote* is a common constituent in all the biotite gneisses, though always in small quantities. It is present mainly in two forms: 1) as very small aggregates that, together with sericite, form the usual product of the feldspar alteration; 2) more significantly, as big subidiomorphic porphyroblasts. In the latter case it is clinozoisite and pistacite. Sometimes big crystals of allanite (included with epidote in the modal analyses) are present and usually show a peripheral transformation into normal epidote. Epidote is often, though not necessarily, connected with biotite. When it occurs together, a derivation from hornblende gneiss, by means of a process to be described later on, may be suspected.

*Muscovite* may be present in large crystals, sometimes with a skeletal appearance. It may be formed either from biotite or from the feldspar alteration together with epidote. In the latter case a reaction similar to that suggested by MARMO (1967) seems likely:



Muscovite, though present in the majority of the samples, never attains significant proportions.

The presence of *sphene*, when hornblende is absent, is strictly controlled by the presence of green biotite and by the Ca content of the rock.

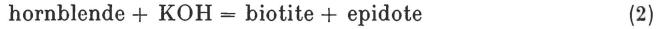
*Apatite* and *zircon* may occur as relatively large crystals.

### *Hornblende gneiss*

All the transitional compositions can be found in the field between this and the biotite gneiss. The main compositional differences between the hornblende and the biotite gneisses, apart from the presence of hornblende, are the more calcic composition (up to An<sub>40</sub>) of the plagioclase in the former and in a general lack of microcline, which seems to be in inverse proportion to the amount of hornblende. In Table 1 two modal analyses of samples belonging to this group are reported.

*Hornblende* is present in xenoblastic, sometimes porphyroblastic (up to 1 cm) crystals. Its common pleochroism is  $\alpha$  = yellow,  $\beta$  = greenish,  $\gamma$  = greenish light blue. The slight difference of the pleochroism in the hornblende of the gneisses and of the amphibolites (Tables 1 & 5) suggests the possibility that they are chemically differ-

ent. This topic will be discussed later on (p. 21). During the migmatitic evolution of these rocks hornblende became unstable. It commonly presents lobate resorption contacts towards quartz and plagioclase. Transformation in a more or less advanced stage into biotite is general: biotite crystallises along the fractures and cleavages of the hornblende or at its border. In this transformation epidote and sphene are formed. The reaction which takes place can be represented in the following way (MARMO, 1967).



The right-hand side of the equation should be stable in lower metamorphic conditions. The equation is only approximate because it does not account for titanium present in the hornblende, which, together with some calcium and silicon forms sphene. When the transformation is very advanced the final product can be a biotite gneiss with no or very few relics of hornblende in it. The association of Ca-silicates (epidote) and Ca-Ti-silicates with biotite in biotite gneisses may indicate an original derivation from a hornblende gneiss.

#### *Garnetiferous gneiss*

The mafic minerals which accompany garnet are biotite with or without hornblende, as is shown by the two modal analyses reported in Table 1.

*Garnet* occurs as pink poikiloblastic crystals up to 5 cm in diameter, which contain inclusions of quartz, feldspar, biotite and opaque grains. Its relationships to the other mafic minerals are not clear: contacts with biotite and hornblende are usually sharp and no transformations seem to have taken place. Possibly garnet grew at the same time, and in equilibrium with the other minerals. Sometimes a brownish red biotite instead of the common greenish type, accompanies the garnet.

*Plagioclase* composition varies from  $\text{An}_{35}$  to  $\text{An}_{42}$ . Normally no significant amounts of microcline and epidote are found.

#### **Banded gneiss**

The two types described at p. 11 have the following petrographical characteristics.

*Type A*: the felsic bands are enriched in quartz, plagioclase, and sometimes very strongly in microcline in respect to the melanocratic bands. (In Table 1 the large amount of microcline found in the felsic band of the examined sample is exceptionally high). The same variability in mineral assemblages described for the veined gneiss is also present in the banded gneiss.

*Type B*: microscopic observation reveals that the passage from the amphibolite bands to the gneissic bands is not as sharp as it appears in the field. The contact is marked by an enrichment in biotite which diminishes towards both the leucocratic bands and the amphibolite, but whereas in the leucocratic bands it remains the prominent mafic mineral, it decreases rapidly in the amphibolites to be a minor constituent. The plagioclase composition does not change much in the mafic and acid

bands: it ranges from around  $An_{30}$  in the latter and is slightly more calcic in the former. Microcline is generally limited to the felsic bands; exceptionally it may appear in small veins in the amphibolites as well. Hornblende may be present in the felsic bands, but undergoes the usual transformation into biotite, epidote and sphene. Garnet, when present in the amphibolite, may also appear in the gneiss.

### Homogeneous gneiss

The garnetiferous gneiss of the Nigerdlip qôrørssua area generally has a granoblastic texture. The mineralogical composition, for the biotite variety, can be seen in Table 1 (73543). No modal determination has been made on the hornblende variety, though some thin sections of it have been studied. *Garnet* in thin section has a pinkish colour. *Biotite* is the usual green, very seldom a dark brown type already described for the veined gneisses. *Microcline* is usually lacking. The plagioclase composition seems to be independent on the presence of hornblende with a consistent composition around  $An_{35}$ . The characters of the other minerals are the same as described for the veined gneiss.

The gneiss of the area west of mountain 780 m is a medium-grained eugranoblastic rock. In Table 1 four modal analyses of samples of this gneiss are reported. The main mineralogical characters and mineral relationships in this gneiss are more or less the same as already described for the veined gneisses with a similar mineral assemblage. A few minor differences are mentioned below.

*Microcline* can co-exist with hornblende (as in sample 57756). It may sometimes form substitution anthiperthites, such as those shown in plate 2a. The anorthite content of the *plagioclase* ranges from 26 % to 35 %, but is most common around 30 %. *Garnet* is seldom present. *Biotite* is represented by a dark brownish type, while the green type is rare.

On the peninsula north of Sermilik avangnardleq the texture is eugranoblastic. The prominent mafic mineral is a greenish or greenish brown biotite, while hornblende is usually completely lacking or is represented by relicts. The *plagioclase* composition is the same as in the preceding gneiss but *microcline* is seldom met with. A modal analysis of one sample of this gneiss is reported in Table 1.

### Chemistry

A few samples for each type of gneiss have been chemically analysed (Table 1). In the same table are also reported the chemical and modal data relative to the two examples of polymigmatisation previously described p. 10 and p. 13). The following observations have been made:

i) all the gneisses have a general quartz dioritic composition, as is shown, for instance, by the triangular plot of fig. 6 regardless of the structural type they belong to.

ii) the  $K_2O$  content is always very low, but, of course, is slightly higher in the biotite (from 0.83 to 1.48) than in the hornblende gneisses (from 0.29 to 0.96). When garnet is present the  $K_2O$  content is a minimum (from 0.25 to 0.69).

iii) the  $Na_2O$  content is fairly constant, though it decreases when hornblende is present.

iv) the  $CaO$ ,  $MgO$  and iron contents are higher in the hornblende than in the biotite gneiss. In the garnet-bearing samples  $FeO$  is higher than in the biotite, and of the same order as in the hornblende gneiss.

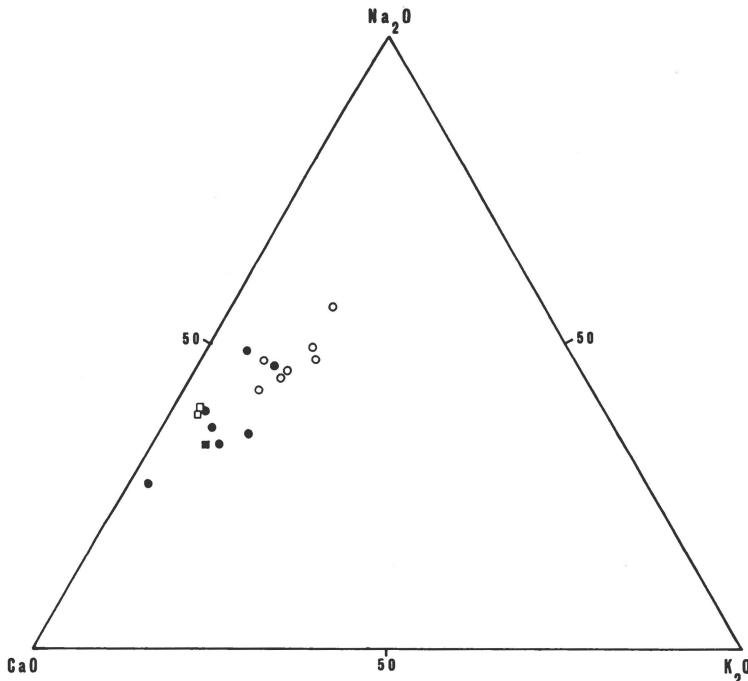


Fig. 6. Triangular plot  $Na_2O$ ,  $CaO$ ,  $K_2O$  (weight %) for the gneisses. Circles = biotitic gneiss; dots = hornblende gneiss, outline squares = garnetiferous biotite gneiss; black squares = garnetiferous hornblende gneiss. Analyses of Table 1.

The variations of the major oxides in relation to  $SiO_2$  for the two polymigmatitic occurrences are shown in fig. 7.

In both the neosome samples  $SiO_2$  increases in relation to the paleosomes. As  $SiO_2$  increases so does  $Na_2O$ , while  $MgO$ ,  $FeO$ ,  $CaO$ ,  $TiO_2$  and  $Al_2O_3$  decrease.  $Fe_2O_3$  exhibits an opposite trend in the two examples.  $K_2O$  does not show any significant variation. The samples of paleosome were

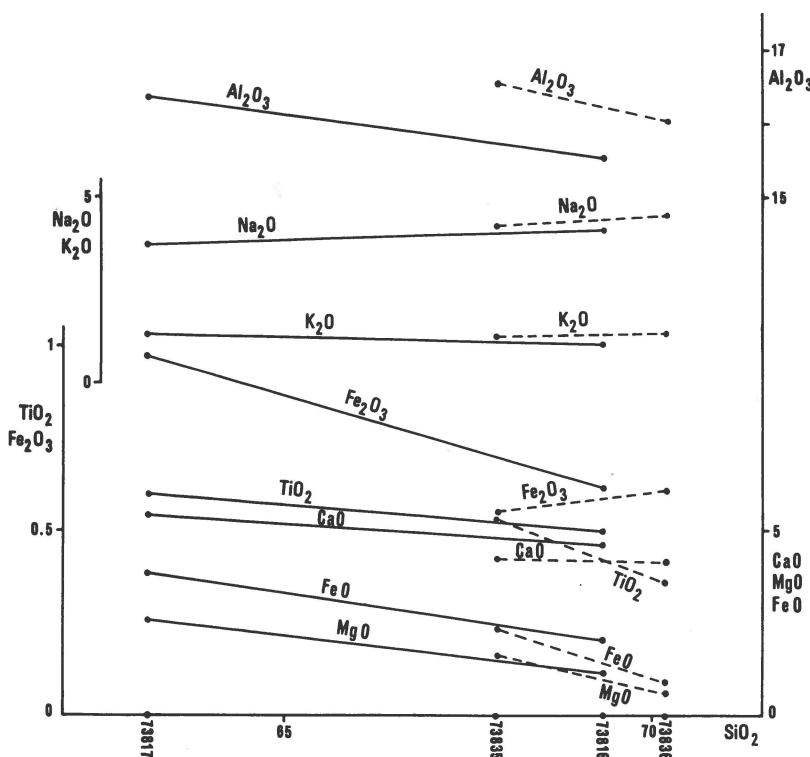


Fig. 7. Chemical variations of the major oxides against  $\text{SiO}_2$  between migmatitic paleosomes and neosomes in the gneisses. (Samples 73816 and 73817 are the neosome and paleosome respectively of fig. 3; samples 73835 and 73836 the paleosome and neosome respectively of fig. 4).

taken from the felsic part and do not account for the amphibolite inclusions present in the gneiss. If the relative abundance of felsic and amphibolitic parts were considered when establishing the average composition of the paleosome, the chemical variation between paleosome and neosome would be much greater than appears here.

In Table 2 the chemical analyses of biotite, hornblende and garnet from the gneisses are reported. The hornblende, in spite of the different pleochroism, is chemically similar to the hornblende of the amphibolite (Table 5), except for a slightly higher total-iron and oxidation ratio in the former.

The release of titanium accompanying the crystallisation of the green biotite, so evident under the microscope as rutile needles (p. 17), cannot be demonstrated chemically, as no significant variation occurs between the green biotites and the others. This is due to the impossibility of separating the rutile from the green biotite.

Table 2. *Chemical analyses of biotite, hornblende and garnet in the gneisses.*

| GGU No.                                     | Biotites             |             |                |            |              | Hornblende   | Garnets |                 |
|---|----------------------|-------------|----------------|------------|--------------|--------------|---------|-----------------|
|   | 57782                | 57724       | 57723          | 73543      | 57738        |              | 57738   | 73543           |
| SiO <sub>2</sub> ...                        | 36.02                | 36.37       | 35.30          | 34.00      | 37.40        | 42.74        | 38.03   | 38.42           |
| TiO <sub>2</sub> ...                        | 2.05                 | 2.36        | 2.60           | 3.50       | 1.99         | 0.89         | 0.30    | 0.33            |
| Al <sub>2</sub> O <sub>3</sub> ...          | 16.88                | 16.58       | 16.56          | 15.29      | 16.18        | 13.12        | 22.10   | 22.93           |
| Fe <sub>2</sub> O <sub>3</sub> ...          | 3.47                 | 5.44        | 6.81           | 9.10       | 6.34         | 4.48         | 0.67    | -               |
| FeO ...                                     | 13.99                | 13.84       | 10.80          | 14.32      | 15.28        | 12.63        | 27.60   | 28.43           |
| MnO ...                                     | 0.18                 | 0.21        | 0.20           | 0.06       | 0.05         | 0.31         | 0.68    | 1.54            |
| CaO ...                                     | 0.81                 | 0.77        | 1.33           | 0.91       | 0.84         | 11.14        | 4.14    | 5.11            |
| MgO ...                                     | 11.74                | 11.03       | 11.39          | 10.03      | 9.32         | 9.80         | 5.95    | 5.74            |
| Na <sub>2</sub> O ...                       | 0.17                 | 0.17        | 0.14           | 0.17       | 0.14         | 1.09         | tr      | tr              |
| K <sub>2</sub> O ...                        | 9.24                 | 8.58        | 7.18           | 8.32       | 6.85         | 0.72         | tr      | tr              |
| P <sub>2</sub> O <sub>5</sub> ...           | -                    | 0.01        | tr             | -          | -            | 0.06         | 0.06    | -               |
| H <sub>2</sub> O <sup>+</sup> ...           | 3.83                 | 3.83        | 5.29           | 3.65       | 5.15         | 2.49         | -       | -               |
| H <sub>2</sub> O <sup>-</sup> ...           | 0.84                 | 0.95        | 1.85           | 0.70       | 1.00         | 0.15         | 0.03    | 0.02            |
|   | 99.22                | 100.14      | 99.45          | 100.05     | 100.54       | 99.62        | 99.56   | 102.52          |
| Number of ions on the basis of 24(0)        |                      |             |                |            |              |              |         |                 |
| Si ....                                     | 5.475                | 5.481       | 5.244          | 5.249      | 5.510        | 6.317        | 5.952   | 5.874           |
| Al ....                                     | 2.525                | 2.519       | 2.756          | 2.751      | 2.490        | 1.683        | 0.048   | 0.126           |
| Al ....                                     | 0.498                | 0.428       | 0.143          | 0.029      | 0.319        | 0.601        | 4.029   | 4.008           |
| Ti ....                                     | 0.235                | 0.266       | 0.291          | 0.406      | 0.220        | 0.099        | 0.036   | 4.144           |
| Fe <sup>3+</sup> ...                        | 0.397                | 0.616       | 0.761          | 1.056      | 0.702        | 0.497        | 0.079   | -               |
| Mg ....                                     | 2.658                | 2.476       | 2.522          | 5.082      | 2.367        | 2.046        | 4.931   | 1.388           |
| Fe <sup>2+</sup> ...                        | 1.777                | 1.745       | 1.341          | 1.848      | 1.883        | 1.561        | 3.611   | 3.635           |
| Mn ....                                     | 0.023                | 0.026       | 0.024          | 0.008      | 0.006        | 0.039        | 0.090   | 0.199           |
| Na ....                                     | 0.051                | 0.051       | 0.038          | 0.052      | 0.038        | 0.313        | -       | 5.978           |
| Ca ....                                     | 0.132                | 0.124       | 0.212          | 1.611      | 0.150        | 1.841        | 1.764   | 2.213           |
| K ....                                      | 1.792                | 1.649       | 1.361          | 1.639      | 1.288        | 0.136        | 0.694   | 0.837           |
| OH ....                                     | 3.896                | 3.773       | 5.226          | 3.761      | 5.095        | 2.467        | -       | -               |
| $\alpha$ ....                               | pale greenish yellow | pale yellow | pale yellow    | yellow     | pale yellow  | yellow       | 62.4    | 60.8 almandine  |
| $\alpha$ ....                               | dark green           | dark green  | greenish brown | dark brown | brownish red | green        | 0.3     | 1.0 andradite   |
| $\gamma$ ....                               | dark green           | dark green  | greenish brown | dark brown | brownish red | bluish green | 24.0    | 21.9 pyrope     |
| 57738 biotite contains chlorite impurities. |                      |             |                |            |              | 2V -62°      | 11.7    | 13.0 grossular  |
|   |                      |             |                |            |              | c:γ 16°      | 1.6     | 3.3 spessartite |

The two garnets have a similar composition and are largely almandine in composition.

Further considerations on the metamorphic significance of some of the analysed minerals are discussed when dealing with the metamorphic facies (p. 68).

## The Pegmatites

### Occurrence

The area is intersected by numerous pegmatites which cut all the rock types, sometimes forming up to 70 % of the outcrop. They reach a maximum thickness of 50 m and may extend for considerable distances. In the gneisses they are represented by: a) concordant or subconcordant dykes or veins often showing a pinch-and-swell structure; b) discordant bodies sometimes ptygmatically folded; c) irregular mobilisates occurring in the tensile stress area between the boudins of amphibolites or constituting the neosome of agmatites. The largest and most numerous pegmatites are found in the veined and banded gneisses. The homogeneous gneisses are also affected by pegmatitic mobilisation but their pegmatites are usually much smaller and less numerous than those of the other gneisses.

In the amphibolites there is not the abundance of pegmatites that is present in the gneisses. Nevertheless amphibolites too are cut or veined by pegmatites and some of the largest have been found to cut this rock type unconformably. In the amphibolite four types can be distinguished: a) a concordant type composed of feldspar and quartz veins, usually a few centimetres thick; b) a subconcordant, slightly discordant type; c) a completely discordant type reaching a maximum thickness of 30–50 metres; d) an irregular type wrapped around ultramafic lenses and very occasionally cutting across them. This last type is genetically identical to that occurring in the shadow pressure areas between the boudins in the gneisses.

### Texture, mineralogy and relation to the host rock

All the pegmatites have a granoblastic texture, becoming locally blastomylonitic as a consequence of younger deformation and crystallisation. Locally a graphic texture may appear ( $P_3$  pegmatites only, see p. 27). The grain size is always rather large, but very fine-grained aplitic pegmatites are present either as autonomous bodies or as a local change in the grain size near the contacts with the host rock. On the basis of the salic minerals, two main groups can be distinguished: a) quartz pegmatites; b) quartz-plagioclase ( $\pm$  microcline) pegmatites.

*Quartz pegmatites*

They are common in all sizes. The thickest one to be observed was an irregular body, 14 m thick, consisting of giant porphyroblasts of quartz and minor albite. Normally no cafemic mineral is associated with quartz. Very occasionally some accessory biotite may appear. In some quartz pegmatites large crystals (10 cm long) of idiomorphic epidote have been found. It is worthy of note that a few quartz pegmatites have been found to be a transformation along the strike of a zoned pegmatite. In this case the quartz pegmatite derives from a thinning of the feldspathic rims (see below), as described elsewhere, for instance by JAHNS (1955) and GRESENS (1967 a & b). As GRESENS points out, this can be due to the rate of opening of the fissures, being lower in the tails than in the centre.

*Quartz-plagioclase ( $\pm$  microcline) pegmatites*

These pegmatites occur either as long concordant to discordant dykes in the gneisses and amphibolites or, less commonly, as irregular bodies between boudins or around mafic lenses. They may exhibit an inward zonation which consists of: 1) a seldom seen, mafic rim of enrichment of femics in the host rock at the contact with the pegmatite; 2) a fine-grained, sometimes aplitic, band in the pegmatite at the contact with the host rock; 3) a coarse-grained zone consisting of quartz, feldspar and mafics; 4) a central zone of pure quartz. This zonation is not always complete and depends to some extent on the total thickness of the pegmatite.

Mineralogically, the abundance of microcline with respect to plagioclase is critical for the following reasons:

- i) the largest pegmatites have been found to be microcline-rich pegmatites;
- ii) if microcline is absent or accessory, the mafic minerals of the pegmatite are the same as those of the host rock. It has been seen, for instance, that the same pegmatite has biotite when crossing a biotite gneiss, and hornblende when the pegmatite passes into hornblende gneisses or amphibolites;
- iii) when microcline becomes the predominant feldspar the mafic mineral of the pegmatite is always biotite regardless of the mineralogy of the host rock. In any case there is a direct correlation between the amount of biotite in the pegmatite and the amount of mafic minerals in the country rock: biotite for instance is more abundant when the pegmatite passes from a gneiss into an amphibolite.

The main characteristics of the minerals are listed as follows.

*Quartz* occurs either as autonomous porphyroblasts or, more rarely, in graphic intergrowth with microcline and in myrmekitic intergrowth with plagioclase.

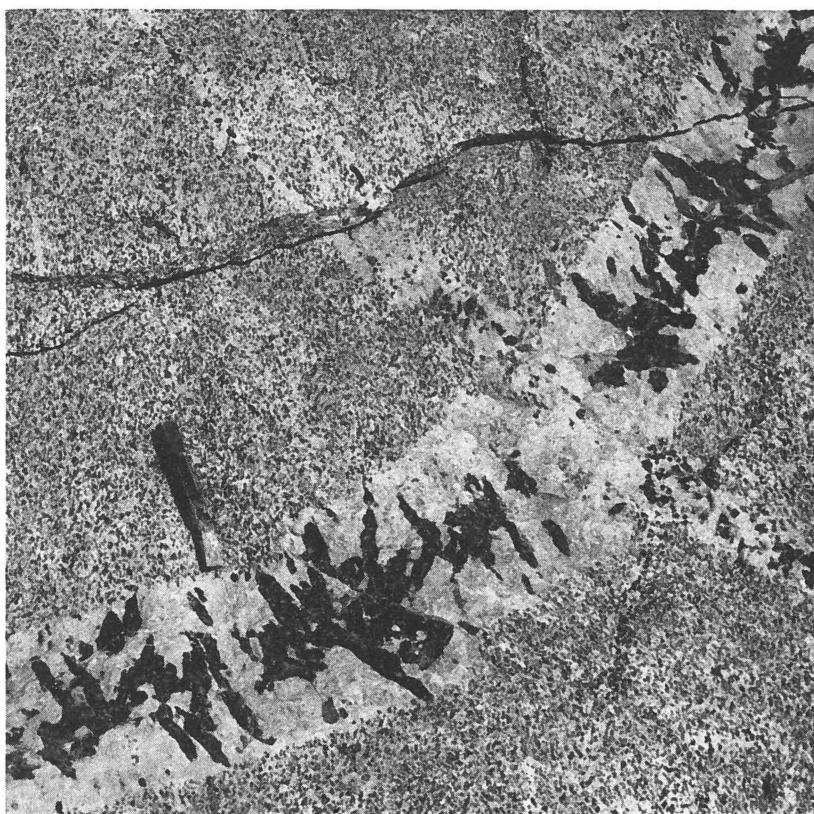


Fig. 8. Hornblende-bearing pegmatite. Hornblende has a zonal distribution and is crystallised with its axis normal to the contacts. 2000 m north-east of mountain 780 m, near the inland ice.

*Plagioclase* is usually oligoclase (up to  $An_{39}$ ). It can either be the sole feldspar or be subordinate in amount to microcline.

*Microcline*, when present in appreciable amounts, can form porphyroblasts of several centimetres. Both perthites (usually microperthites) and antiperthites are common. The best examples of antiperthites occur in the pegmatites west of mountain 780 m, where they are also present in the homogeneous gneiss (see sample 57756).

*Biotite* is normally strongly pleochroic varying from colourless to reddish brown, and may form giant crystals up to one metre broad. It is found both inside the pegmatites (in which case it may have a random distribution or be arranged in strings parallel to the foliation of the host rock) and as a reaction product at the contact of microcline-bearing pegmatites. Small amounts of biotite are found (together with sphene and epidote) as transformation of hornblende in hornblende-bearing pegmatites.

*Hornblende* often forms very large crystals (up to 20 or 30 cm) which may show a random, or a zonal (fig. 8) distribution inside the pegmatite. In the latter case, hornblende crystals are mainly concentrated at the centre of the pegmatite and

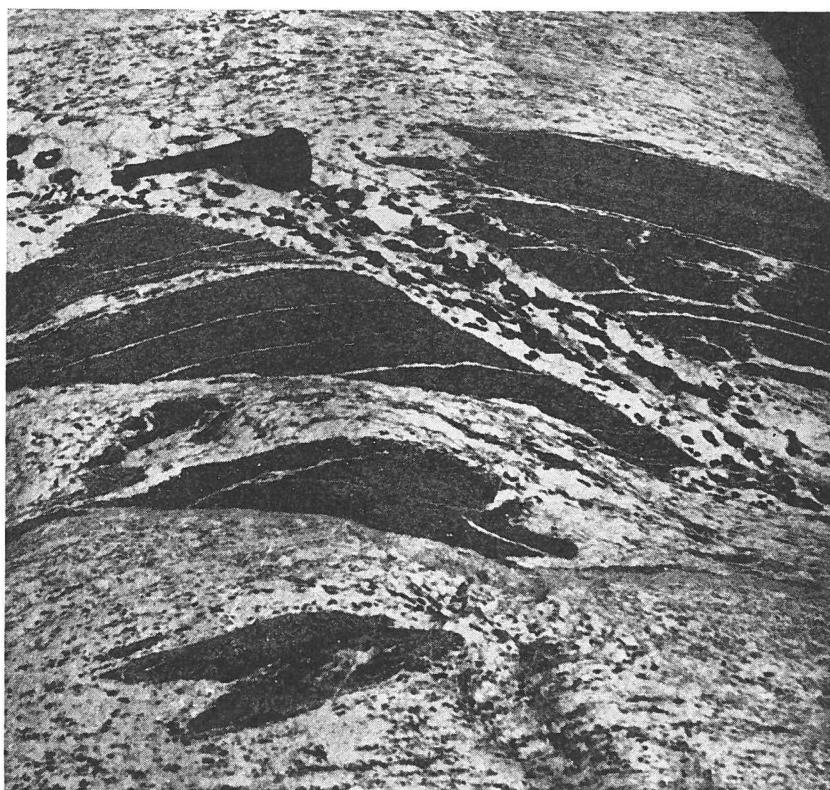


Fig. 9. Pegmatite cross-cutting hornblende gneiss and a hornblende inclusion. The pegmatite contains large hornblende crystals which are more abundant when the pegmatite passes from the gneiss into the inclusion; the same locality as fig. 8.

crystallised with their major axis normal to the contacts. Hornblende is enriched in the pegmatites which cross-cuts rocks rich in this mineral (fig. 9). The relative percentages of hornblende and salic minerals in the pegmatite, as opposed to the host rock, are not very different; for instance, two estimates of the modal composition of two pegmatites, a zonal and a random (carried out by geometric measurement on projected colour slides), have given: ho 33.8 %, pl + qu 66.2 %, and ho 31.6 % pl + qu 68.4 %, respectively. The composition of the country rock is of the same order. A single hornblende crystal has been chemically analysed (Table 3), and it can be seen that it is very similar to those from the gneisses and amphibolites.

*Garnet* (normally a pinkish type), *epidote*, *muscovite* and *allanite* (in crystals up to 8 cm long) are accessories.

*Tourmaline* appears as a reaction product between microcline-rich pegmatites and amphibolites or ultramafics. Tourmaline is usually zoned, with a brownish core and green periphery and forms crystals up to 3 cm. Sometimes it forms massive aggregates with little or no feldspar or quartz.

*Beryl* has been found at the contact between an amoeboidal pegmatite and an ultramafic body. It consists of bluish green transparent crystals up to 4 cm long. A

Table 3. *Chemical analysis of a hornblende from a pegmatite.*

| GGU No.                              | 57766                                 |                  |             |
|--------------------------------------|---------------------------------------|------------------|-------------|
| Oxides wt. %                         | Number of ions on the basis of 24 (0) |                  |             |
| SiO <sub>2</sub> .....               | 42.72                                 | Si               | 6.347       |
| TiO <sub>2</sub> .....               | 0.80                                  | Al               | 1.653       |
| Al <sub>2</sub> O <sub>3</sub> ..... | 13.06                                 | Al               | 0.634       |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 4.55                                  | Ti               | 0.089       |
| FeO.....                             | 12.39                                 | Fe <sup>3+</sup> | 0.508       |
| MnO.....                             | 0.33                                  | Mg               | 2.434       |
| CaO.....                             | 10.44                                 | Fe <sup>2+</sup> | 1.539       |
| MgO.....                             | 10.99                                 | Mn               | 0.040       |
| Na <sub>2</sub> O.....               | 1.17                                  | Na               | 0.338       |
| K <sub>2</sub> O.....                | 0.54                                  | Ca               | 1.662       |
| H <sub>2</sub> O <sup>+</sup> .....  | 2.08                                  | K                | 0.103       |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.13                                  | OH               | 2.037       |
|                                      | 99.20                                 |                  |             |
| c:γ 14°                              |                                       | α                | pale yellow |
| 2V -65°                              |                                       | β                | light green |
|                                      |                                       | γ                | light blue  |

few crystals are present in the pegmatite reaching a maximum size and concentration in the biotite reaction zone and rapidly decrease in the hornblendite zone of the ultramafic lens (p. 42).

### Composition related to timing

There is good field evidence that pegmatites were formed in several episodes during the evolution of the area. A key sub-area, shown in fig. 10, enabled us to establish the following age relationships: the first pegmatites (P<sub>1</sub>) are a consequence of metamorphic differentiation and are conformably folded together with the biotite-rich layers by F<sub>3</sub> folds (p. 80); the second pegmatites (P<sub>2</sub>) are represented by subconcordant to completely discordant bodies whose emplacement took place during the F<sub>3</sub> deformation, as is shown by the fact that the pegmatites have suffered only partially from the folding (see also RAMSAY, 1967, p. 344 for a similar example). P<sub>3</sub> pegmatites are represented by sills and dykes that, in the examples shown by fig. 10, are undeformed; the youngest pegmatite swarm, P<sub>4</sub>, consists of quartz-veins which occur in the tension areas of F<sub>5</sub> folds (p. 82).

The field data and samples are not sufficient to give unambiguous relationships between the composition and the relative ages of the pegmatites, but the following observations can be made.

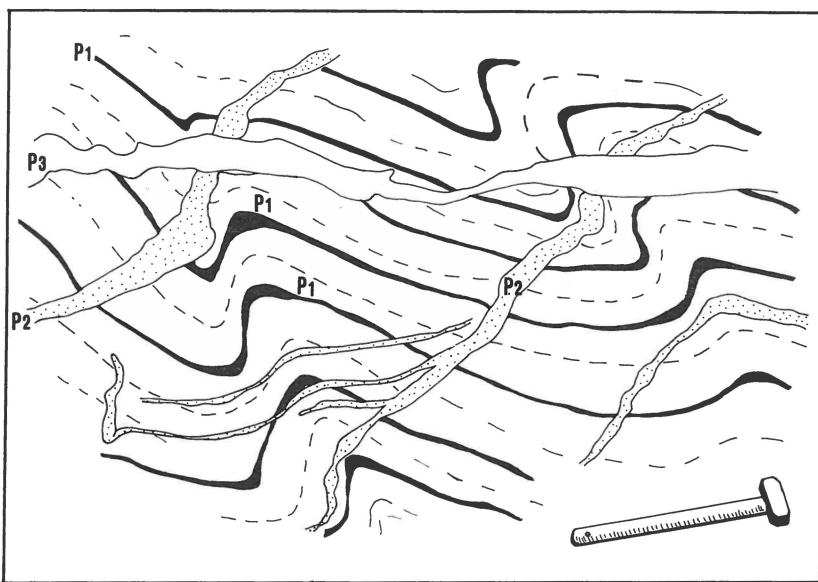


Fig. 10. Pegmatite network in the gneisses. Three generations of pegmatites can be distinguished ( $P_1$ ,  $P_2$ ,  $P_3$ ).  $P_1$  (black) are concordant pegmatites conformably folded in a  $F_3$  fold.  $P_2$  pegmatites (dots) were emplaced during the  $F_3$  deformation as they have suffered only partially from the folding.  $P_3$  (white) are undeformed and cross all the other pegmatites. 2500 m north-east of mountain 933 m. (Drawn from a series of photographs).

The  $P_1$  swarm, concordant with the foliation, contains a little microcline which post-dates plagioclase (see p. 17). The pegmatites of the  $P_2$  swarm have either plagioclase as the only feldspar or microcline subordinate to plagioclase, and this only when the pegmatite cuts biotite gneisses. The  $P_3$  pegmatites, parallel to  $F_4$  axial surfaces (p. 82), contain more microcline than plagioclase, and biotite as the only mafic mineral. The mobilisates connected to  $F_5$  deformation (p. 82) are quartz-rich.

## LAYERS AND INCLUSIONS IN THE GNEISSES

### **Introduction**

Layers of unmigmatised rocks are interbanded with the gneisses and form long, thin horizons which can be traced for many kilometres. These are of great assistance in explaining some of the intricate structures of the area. They consist of four intimately associated lithotypes: 1) amphibolites, which largely predominate; 2) ultramafic lenses; 3) rusty schists; 4) micro-granodioritic rock. The layers formed by these lithotypes are concordant with the gneiss foliation for many kilometres and suggest that the foliation in these places is concordant with a primary bedding. It is possible however that in parts of the area where the amphibolites are lens-shaped, younger foliations may have obliterated the primary orientation through transposition. Discordance between foliation and lithology is seen in the closures of both  $F_1$  and  $F_2$  folds (see p. 78). All these layers tend to fade out along the strike. At first processes of boudinage, conforming with the pattern described by RAMBERG (1956), and of agmatitisation transform the amphibolites into streaky gneisses and agmatites. The final transformation product can be a hornblende gneiss or even a biotite gneiss with few or no relics of amphibolite.

Small lens-shaped or angular inclusions of unmigmatised rocks are also found in the gneisses. They consist of amphibolites, ultramafic rocks, and calc-silicate rocks. Amphibolites and ultramafic rocks occur as both lenses and layers whilst calc-silicate rocks occur only as lenses.

### **Layers in the Gneisses**

#### **Amphibolites**

##### **Description**

It has been mentioned that the lateral passage from an amphibolite horizon to the country gneisses often takes place through an alternation of felsic gneisses and amphibolite bands. Sometimes, however, the contact is sharp. When unaffected by macroscopic signs of migmatisation, the amphibolites are concordant layers which in some places, such as in

the amphibolites of mountain 933 m, possess a strong  $F_2$  cleavage. In many cases the amphibolites possess either continuous or discontinuous compositional layering which has been divided into 3 types:-

- i) thin (up to 10 cm thickness) discontinuous but concordant quartz-feldspar or quartz layers with hornblendite rims;
- ii) layering expressed by an alternation of either grain size or the abundance of the main minerals;
- iii) alternating layers having different mineral assemblages which give the various layers different colours (colour-layering).

In the colour-layering the typical alternation consists of the following types:

- a) *dark-green layers* of almost pure hornblendite with a maximum thickness of 20 cm grade both laterally and along the strike into normal plagioclase amphibolites. They may contain sporadic crystals of clinopyroxene and traces of epidote and feldspar.
- b) *green layers* are the dominant type of amphibolite. They are characterised by the predominance of hornblende over feldspar and by the lack of significant amounts of other minerals.
- c) *light-green layers* are characterised by the presence of epidote and/or clinopyroxene. They have a maximum thickness of 50 cm and are fairly continuous but thin out along the strike.
- d) *garnet-bearing layers*. Garnet is associated with both the dark-green layers and the green layers. When it is present in large amounts, quartz is also abundant. Garnet is either concentrated in irregular pods uniformly distributed within the banding, or is limited to acid veins which cut the amphibolite.

### Petrography

The following petrographic description mainly refers to the colour-layering.

The texture of the normal amphibolites is crystalloblastic and is characterised by a planar arrangement of elongated hornblende prisms with intergranular feldspar. The epidote, pyroxene and garnet-bearing layers often possess an imperfectly developed granoblastic texture. The texture of the hornblendite layers is characterised by the parallel arrangement of the hornblende crystals, which impart either a planar or linear fabric to the rock.

The modal analyses of samples belonging to the different bands are shown in Table 4. No point counting has been carried out on hornblendites, but their hornblende content has been estimated at over 90 %.

## LAYERS AND INCLUSIONS IN THE GNEISSES

### Introduction

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### Layers in the Gneisses

#### Amphibolites

##### Description

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Table 4 (continued).

| GGU No.                              | garnet-bearing green layers |        |       |       |       |       |       |       |
|--------------------------------------|-----------------------------|--------|-------|-------|-------|-------|-------|-------|
|                                      | 73517                       | 73519  | 73520 | 73846 | 73863 | 75815 | 75833 | 75838 |
| <b>Oxides wt. %</b>                  |                             |        |       |       |       |       |       |       |
| SiO <sub>2</sub> .....               | 51.42                       | 47.50  | 48.29 | 54.28 | 53.72 | 51.72 | 52.98 | 59.77 |
| TiO <sub>2</sub> .....               | 2.40                        | 2.59   | 1.00  | 0.92  | 1.86  | 0.45  | 0.81  | 0.86  |
| Al <sub>2</sub> O <sub>3</sub> ..... | 12.29                       | 12.54  | 14.33 | 14.27 | 12.33 | 16.63 | 15.67 | 15.03 |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 3.17                        | 4.44   | 4.13  | 2.80  | 2.96  | 1.48  | 0.87  | 2.17  |
| FeO.....                             | 14.04                       | 14.24  | 12.72 | 10.54 | 16.25 | 8.87  | 11.38 | 6.38  |
| MnO.....                             | 0.28                        | 0.28   | 0.23  | 0.24  | 0.31  | 0.18  | 0.23  | 0.16  |
| CaO.....                             | 9.46                        | 9.92   | 9.39  | 7.64  | 6.80  | 10.41 | 8.13  | 6.52  |
| MgO.....                             | 3.93                        | 5.47   | 5.59  | 5.29  | 3.53  | 6.27  | 5.54  | 3.58  |
| Na <sub>2</sub> O.....               | 1.46                        | 1.56   | 2.14  | 2.39  | 0.41  | 1.60  | 1.98  | 2.62  |
| K <sub>2</sub> O.....                | 0.20                        | 0.34   | 0.26  | 0.31  | 0.19  | 0.52  | 0.35  | 0.32  |
| P <sub>2</sub> O <sub>5</sub> .....  | 0.05                        | 0.04   | 0.02  | 0.01  | 0.13  | 0.02  | 0.06  | 0.03  |
| CO <sub>2</sub> .....                | —                           | —      | —     | —     | —     | —     | —     | —     |
| H <sub>2</sub> O <sup>+</sup> .....  | 1.24                        | 1.36   | 1.70  | 0.86  | 1.20  | 1.77  | 1.36  | 1.72  |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.10                        | 0.12   | 0.19  | 0.07  | 0.03  | 0.06  | 0.03  | 0.09  |
|                                      | 100.04                      | 100.40 | 99.99 | 99.62 | 99.72 | 99.98 | 99.39 | 99.25 |
| mg.....                              | 0.29                        | 0.34   | 0.37  | 0.41  | 0.25  | 0.52  | 0.44  | 0.43  |
| ti.....                              | 4.61                        | 4.49   | 1.75  | 1.82  | 3.84  | 0.83  | 1.57  | 2.08  |
| <b>Mode vol. %</b>                   |                             |        |       |       |       |       |       |       |
| quartz.....                          | 15.5                        | 7.3    | 5.4   | 5.0   | 36.7  | 11.7  | 4.5   | 13.7  |
| plagioclase.....                     | 33.6                        | 14.2   | 27.7  | 49.6  | 5.5   | 10.3  | 44.7  | 54.3  |
| biotite.....                         | 0.6                         | —      | —     | 0.6   | 0.2   | 0.7   | tr    | 4.3   |
| chlorite.....                        | —                           | —      | —     | —     | 0.9   | 2.2   | 0.4   | 2.0   |
| garnet.....                          | 13.9                        | 4.7    | 8.1   | 1.1   | 26.9  | 10.6  | 5.1   | 0.2   |
| hornblende.....                      | 31.8                        | 69.9   | 57.0  | 41.7  | 25.4  | 62.5  | 43.0  | 23.7  |
| clinopyroxene.....                   | —                           | —      | —     | —     | —     | —     | —     | —     |
| epidote.....                         | —                           | —      | tr    | —     | —     | 0.7   | 0.3   | —     |
| sphene.....                          | —                           | 2.0    | 0.1   | —     | —     | tr    | tr    | tr    |
| apatite.....                         | 0.5                         | 0.2    | tr    | 0.1   | 0.9   | 0.3   | tr    | 0.2   |
| zircon.....                          | tr                          | —      | 0.1   | tr    | 0.1   | 0.1   | tr    | 0.1   |
| calcite.....                         | —                           | —      | tr    | —     | —     | —     | —     | —     |
| tourmaline.....                      | —                           | —      | —     | —     | —     | —     | 0.1   | 0.1   |
| zeolites.....                        | —                           | —      | —     | —     | —     | —     | —     | —     |
| opaque.....                          | 4.1                         | 1.7    | 1.5   | 1.8   | 3.4   | 0.9   | 1.8   | 1.4   |
| An % in plagioclase.....             | 36                          | 39     | 41    | 40    | 20    | 69    | 45    | 42    |
| hornblende { 2V.....                 | -56°                        | -61°   | -63°  | -73°  | -67°  | -79°  | -79°  | -78°  |
| c:γ.....                             | 14°                         | 11°    | 14°   | 14°   | 14°   | 15°   | 14°   | 12°   |
| clinopyroxene { 2V.....              | —                           | —      | —     | —     | —     | —     | —     | —     |
| c:γ.....                             | —                           | —      | —     | —     | —     | —     | —     | —     |

75838: modal garnet probably underestimated because of its irregular distribution.

(continued)

Table 4 (continued).

| GGU No.                              | light-green layers |       |       | garnet-bearing light-green layers |        |        |
|--------------------------------------|--------------------|-------|-------|-----------------------------------|--------|--------|
|                                      | 57872              | 73858 | 75904 | 57748                             | 57794  | 75910  |
| <b>Oxides wt. %</b>                  |                    |       |       |                                   |        |        |
| SiO <sub>2</sub> .....               | 48.28              | 50.40 | 56.98 | 46.28                             | 48.39  | 48.18  |
| TiO <sub>2</sub> .....               | 0.70               | 0.28  | 0.60  | 1.35                              | 3.60   | 0.50   |
| Al <sub>2</sub> O <sub>3</sub> ..... | 14.84              | 15.03 | 15.67 | 14.90                             | 12.60  | 13.38  |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 4.15               | 1.64  | 1.65  | 2.63                              | 6.84   | 3.27   |
| FeO.....                             | 8.48               | 8.25  | 6.87  | 9.34                              | 14.65  | 8.95   |
| MnO.....                             | 0.22               | 0.19  | 0.24  | 0.22                              | 0.23   | 0.21   |
| CaO.....                             | 12.69              | 12.90 | 8.34  | 12.48                             | 8.88   | 12.90  |
| MgO.....                             | 7.21               | 7.91  | 4.23  | 10.58                             | 2.89   | 8.01   |
| Na <sub>2</sub> O.....               | 1.76               | 1.29  | 2.07  | 1.18                              | 1.02   | 2.05   |
| K <sub>2</sub> O.....                | 0.14               | 0.22  | 1.38  | 0.19                              | 0.07   | 0.20   |
| P <sub>2</sub> O <sub>5</sub> .....  | 0.04               | 0.01  | 0.03  | 0.05                              | 0.09   | 0.03   |
| CO <sub>2</sub> .....                | 0.53               | 0.26  | —     | —                                 | 0.10   | 0.92   |
| H <sub>2</sub> O <sup>+</sup> .....  | 0.99               | 0.97  | 1.41  | 1.12                              | 0.97   | 1.73   |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.05               | 0.04  | 0.09  | 0.04                              | 0.05   | 0.10   |
|                                      | 100.08             | 99.39 | 99.56 | 100.36                            | 100.38 | 100.43 |
| <i>mg</i> .....                      | 0.51               | 0.59  | 0.47  | 0.61                              | 0.20   | 0.54   |
| <i>ti</i> .....                      | 1.16               | 0.48  | 1.31  | 2.07                              | 6.79   | 0.82   |
| <b>Mode vol. %</b>                   |                    |       |       |                                   |        |        |
| quartz.....                          | 3.1                | 2.4   | 7.7   | 0.1                               | 12.2   | 1.2    |
| plagioclase .....                    | 19.7               | 32.0  | 32.6  | 29.7                              | 29.0   | 11.1   |
| biotite.....                         | —                  | —     | 8.2   | 0.6                               | 0.1    | —      |
| chlorite.....                        | —                  | —     | 2.7   | —                                 | —      | 0.7    |
| garnet.....                          | —                  | —     | 0.4   | 8.6                               | 8.6    | 4.0    |
| hornblende.....                      | 61.2               | 53.7  | 35.9  | 40.7                              | 20.8   | 63.9   |
| clinopyroxene .....                  | —                  | 5.5   | —     | 15.7                              | 13.5   | 0.5    |
| epidote .....                        | 13.4               | 4.3   | 10.0  | 3.3                               | 0.3    | 15.5   |
| sphene .....                         | 1.6                | 0.5   | 1.2   | 0.1                               | tr     | 1.4    |
| apatite .....                        | 0.1                | 0.2   | 0.5   | 0.1                               | 0.7    | 0.3    |
| zircon .....                         | tr                 | tr    | 0.1   | —                                 | —      | tr     |
| calcite .....                        | 0.6                | 1.1   | —     | —                                 | 0.1    | 1.2    |
| tourmaline .....                     | —                  | —     | 0.7   | —                                 | —      | —      |
| zeolites .....                       | —                  | —     | —     | —                                 | —      | —      |
| opaque .....                         | 0.3                | 0.3   | tr    | 1.0                               | 14.7   | 0.2    |
| An % in plagioclase                  | 68-36              | 78    | 45    | 53                                | 40     | 71     |
| hornblende { 2V ..                   | -73°               | -77°  | -76°  | -77°                              | -49°   | -76°   |
| c:γ ..                               | 15°                | 15°   | 15°   | 13°                               | 13°    | 14°    |
| clinopyroxene { 2V ..                | —                  | +59°  | —     | +61°                              | +62°   | —      |
| c:γ ..                               | —                  | 41°   | —     | 39°                               | 46°    | —      |

75910: 0.34 % of altered plagioclase.

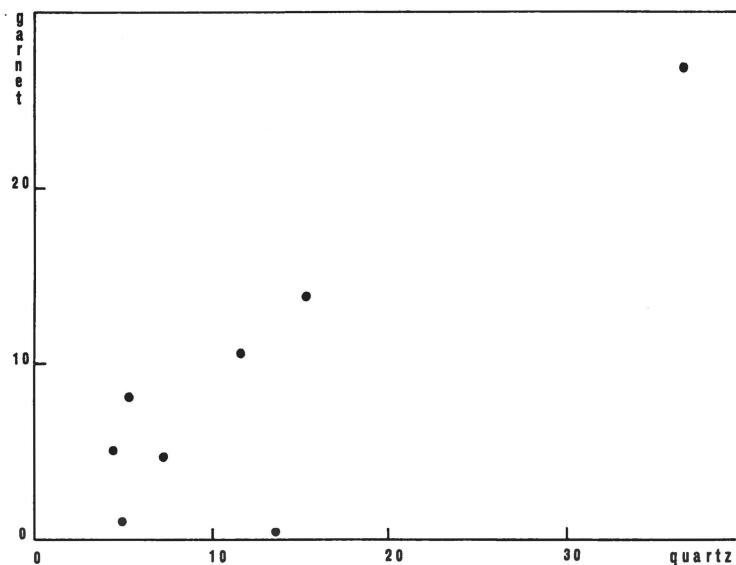


Fig. 11. Plot of modal quartz against modal garnet in the amphibolites of the garnet-bearing light-green layers (Table 4). A good positive correlation can be seen.

The following observations have been deduced from the modal analyses:

1) quartz is present in the majority of the samples. It reaches a maximum level when garnet is present in the hornblende-plagioclase-bearing layers. In these layers its increase is proportional to that of the garnet (fig. 11).

2) the normal amphibolites contain between 51–77 % hornblende; the light-green layers contain between 36–61 % hornblende; the green layers which possess garnet contain between 23–70 % hornblende, while the light-green layers which possess garnet contain 21–64 % hornblende.

3) the anorthite content in plagioclase is higher in the pyroxene–epidote (–garnet)-bearing layers than in the other types.

4) biotite occurs only in very small amounts in the various layers, regardless of their mineralogical composition.

As many of the mineralogical features are similar in the different layers, their mineralogy will be discussed together to avoid repetition; differences will be pointed out as they occur.

*Hornblende* is normally green with  $\alpha$  = yellow,  $\beta$  = green (greyish, brownish green),  $\gamma$  = brownish green (green, bluish green). The  $2V\alpha$  in the examined hornblendes is generally  $70^\circ$ , with a maximum of  $84^\circ$ . A good correlation exists between Niggli  $mg$  of the rocks and  $2V$  of hornblende (fig. 12). Therefore, as the  $mg$  of the rock depends mainly on hornblende, it can be inferred that the  $2V$  in hornblende largely depends

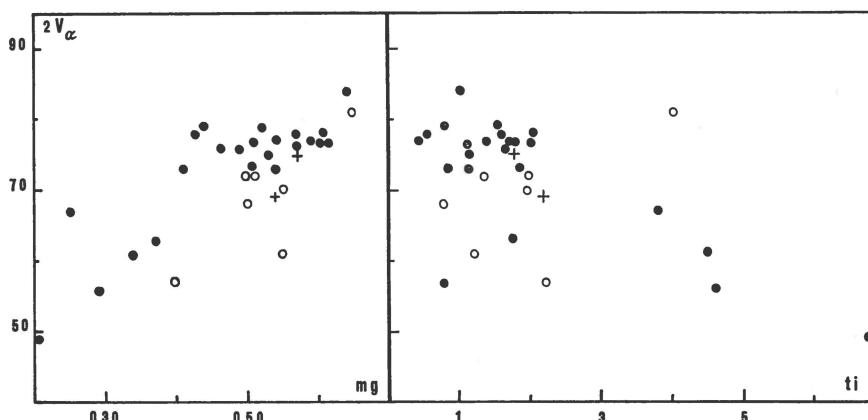


Fig. 12. Plot of Niggli  $mg$  and  $ti$  of the rocks against  $2V$  of hornblende. (Circles = amphibolite boudins in the gneisses; dots = major amphibolite layers, crosses = discordant amphibolite dykes).

on its magnesium content, in agreement with what was observed by WINCHELL & WINCHELL (1951, p. 434) and TRÖGER (1952, p. 77). A poorer correlation also exists between the  $2V$  of hornblende and Niggli  $ti$  (fig. 12), but this is to be expected because the titanium in the rock does not depend only on hornblende, but also on the presence of titanium minerals such as ilmenite, sphene and rutile. Nevertheless, it is likely that titanium also influences the  $2V$  in the sense that as titanium increases, the  $2V$  decreases.

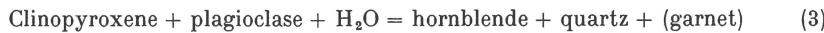
Secondary transformations to actinolite are often present. The chemical composition of a hornblende can be seen in Table 5.

*Plagioclase* has a wide range of composition with an average of  $An_{40}$  and more than half the samples have an anorthite content between 35 % and 45 %.

Plagioclase is usually fresh but may possess some alteration to sericite and epidote. In the vicinity of a fault, plagioclase is completely recrystallised as albite. In one case the plagioclase was microantiperthitic (plate 2b), which is unusual because the antiperthites co-exist with an An content of 78 %. A similar phenomenon, antiperthites on calcic plagioclase, was found by CAPEDRI (1968) on high-grade metamorphic basic rocks of the Ivrea-Verbano zone in Italy. Sometimes an irregular zoning is present, with a core of An content between 60 % and 70 % rimmed by An between 35 % to 45 %.

*Clinopyroxene* occurs in the light-green layers as pale green (rarely colourless) pleochroic crystals, together with epidote. It is normally a diopsidic salite or augite. The optical data on pyroxene occurring in the analysed samples are reported in Table 4.

Pyroxene seems to be an unstable phase; it shows rounded resorption contacts towards the other minerals and is often substituted by alteration products such as saussuritic aggregates, iron oxides and epidote (which may replace it completely). Its relationship to hornblende is not clear; sometimes they seem to co-exist in equilibrium, while at other times it is partially replaced by hornblende either marginally or along cleavage planes. Actinolite is also seen replacing pyroxene. The relationship between pyroxene and hornblende can be represented by the following equation:



3\*

Table 5. *Chemical analyses of hornblende and garnet from amphibolites.*

| GGU No.                            | Oxides wt. % |        | Number of ions on the basis of 24 (0) |        |                            |        |       |
|------------------------------------|--------------|--------|---------------------------------------|--------|----------------------------|--------|-------|
|                                    | 57731        | 57794  | 57731                                 | 57794  | hornblende                 | garnet |       |
|                                    | hornblende   | garnet | hornblende                            | garnet |                            |        |       |
| SiO <sub>2</sub> ...               | 44.09        | 37.72  | Si ..... 6.496                        | 8.000  | Si ..... 6.013             | 6.013  |       |
| TiO <sub>2</sub> ...               | 0.98         | 0.47   | Al ..... 1.504                        |        | Al ..... -                 | -      |       |
| Al <sub>2</sub> O <sub>3</sub> ... | 12.23        | 19.94  | Al ..... 0.619                        |        | Al ..... 3.746             | 3.746  |       |
| Fe <sub>2</sub> O <sub>3</sub> ... | 2.81         | 2.14   | Ti ..... 0.104                        |        | Ti ..... 0.056             | 4.059  |       |
| FeO ...                            | 11.88        | 26.32  | Fe <sup>3+</sup> ... 0.311            |        | Fe <sup>3+</sup> ... 0.257 | 0.257  |       |
| MnO ...                            | 0.23         | 0.68   | Mg ..... 2.541                        | 5.068  | Mg ..... 0.839             | 0.839  |       |
| CaO ...                            | 11.64        | 8.17   | Fe <sup>2+</sup> ... 1.464            |        | Fe <sup>2+</sup> ... 3.509 | 3.509  |       |
| MgO ...                            | 11.57        | 3.53   | Mn ..... 0.029                        |        | Mn ..... 0.092             | 0.092  |       |
| Na <sub>2</sub> O ...              | 1.11         | 0.06   | Na ..... 0.316                        |        | Na ..... 0.016             | 0.016  | 5.855 |
| K <sub>2</sub> O ...               | 0.94         | 0.02   | Ca ..... 1.838                        | 2.330  | Ca ..... 1.396             | 1.396  |       |
| P <sub>2</sub> O <sub>5</sub> ...  | 0.10         | -      | K ..... 0.176                         |        | K ..... 0.003              | 0.003  |       |
| H <sub>2</sub> O <sup>+</sup> ...  | 2.02         | -      | OH..... 2.020                         |        |                            |        |       |
| H <sub>2</sub> O <sup>-</sup> ...  | 0.15         | 0.17   |                                       |        |                            |        |       |
|                                    | 99.70        | 99.22  | 2V ..... -78°                         |        | almandine                  | 60.1   |       |
|                                    |              |        | c: γ ..... 14°                        |        | andradite                  | 8.0    |       |
|                                    |              |        | α ..... yellow                        |        | pyrope                     | 14.4   |       |
|                                    |              |        | β ..... brownish green                |        | grossular                  | 15.9   |       |
|                                    |              |        | γ ..... dark green                    |        | spessartite                | 1.6    |       |

Clearly, the right-hand side of the equation represents lower metamorphic conditions or at least conditions in which the increase in the water pressure has led to dis-equilibrium between pyroxene and plagioclase.

*Garnet* occurs as pink poikiloblastic crystals, rich in inclusions of all the other minerals. In some thin sections it can be observed that the development of garnet either followed equation 3 above or equation 4 described later, by replacement of pyroxene (plate 3a).

The chemical composition of one garnet is reported in Table 5. It is rich in almandine, but an appreciable amount of grossular and pyrope are present.

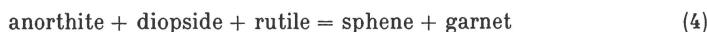
*Quartz* is present in small amounts in the majority of the samples. It occurs as both primary, irregular-shaped crystals and as small porphyroblasts derived from the reaction pyroxene-hornblende or hornblende-epidote.

*Biotite* occurs as brownish red pleochroic flakes which are frequently altered to chlorite (notably in shear zones). Sometimes biotite occurs in the cracks of garnets. Biotite was the last mafic, except for chlorite and epidote, to crystallise.

*Epidote* usually occurs as porphyroblasts of pistacite (2V = - 78°) but optically positive clinzoisite and allanite also occur. The formation of epidote is related to the calcium content of the rock.

*Titanium minerals* are represented either by ilmenite or by sphene. Rutile is absent because of the high iron content of the amphibolites. The ilmenite content may in some cases reach 15 % (see for instance sample 57794, Table 4). It is unstable

and is partially replaced by sphene with the release of iron oxides. Sometimes sphene is the only titanium mineral in the rock. In some cases it is associated with garnet which suggests that the reaction



(RAMBERG, 1952, p. 73) has occurred. According to RAMBERG, the lower grade of metamorphism is represented by the right-hand side. This reaction is very clear in sample 73819 (plate 3a).

Accessory minerals are apatite, zircon, tourmaline, calcite, zeolites, pyrrhotite, calcopyrite and pyrite.

### Chemistry

Chemical and modal analyses of amphibolites are reported in Table 4. These analyses are representative of the principal amphibolites of the area and are arranged in accordance with the different paragenetic bands to which they belong. RIVALENTI (1971) has undertaken a special study of the amphibolites and it is only noted here that the amphibolites generally have a non-alkaline composition which may be seen by plotting  $\text{SiO}_2$  against  $\text{Na}_2\text{O} + \text{K}_2\text{O}$ .

## Ultramafic rocks

### Description

Ultramafic rocks occur as discontinuous lens-shaped bodies in the concordant amphibolites. They are easily recognisable in the field, even from a long distance, because of their rounded erosion surface which is red or grey in colour. A single lens may reach a maximum length of several hundred metres and a maximum thickness of 200 metres. The elongation of the lenses in a single amphibolite horizon is roughly parallel to the foliation of the amphibolite, which indicates that formerly the lenses were connected in a single horizon and their actual shape may be a tectonic phenomenon, such as boudinage. The untransformed ultramafics do not generally show any foliation except an occasional closely-spaced set of parallel fractures, and are fine- to medium-grained, equigranular rocks. Foliation is, however, present when serpentinite or talc rock replaces a peridotite. Contacts with amphibolites are always marked by a rim of pure hornblendite which rapidly grades into the amphibolite by enrichment with feldspar. The ultramafic lenses possess concentric zonation. Four zones can be distinguished which are from core to rim: a) nucleus; b) light-green zone; c) anthophyllite zone; d) hornblendite zone. It is worth noting that a similar zoning was found in ultramafic rocks by DAWES (1970), while working on J. A. D. Jensens Nunatakker and Dalagers Nunatakker north of the present area and by MÍSAŘ (in press) in the area north of Sermilik, see also WALTON (1966).

The transitions among the different zones, although appearing abrupt in the field, are gradational when observed under the microscope.

### Petrography

#### Nucleus

The nucleus generally consists of peridotite, partially transformed outwards into serpentinite. When the transformation is very advanced serpentinite alone may form the nucleus. When untransformed, the peridotite core is massive, equigranular (seldom heterograngular) fine- to medium-grained with a composition that varies from pure dunite (90–100% olivine, STRECKEISEN, 1967) to amphibole–(mica-) peridotite which is normally an amphibole saxonite.

The *olivine* composition (Table 6) varies from  $Fo_{70}$  (sample 57874, amphibole-mica saxonite) to a maximum  $Fo_{98}$  (sample 73840, amphibole saxonite). Olivine occurs both as fresh subidiomorphic grains and partially altered to chrysotile and talc. In

Table 6. *Composition of olivines and orthopyroxene co-existing in ultramafic rocks.*

| GGU Sample<br>Nos. | olivine | 2V   | orthopyroxene | 100 Mg<br>Mg + Fe <sup>2+</sup> + Fe <sup>3+</sup> + Mn |
|--------------------|---------|------|---------------|---|
|                    | Fo %    |      |               |   |
| 57753.....         | 95      | −72° | 77            |   |
| 57786.....         | 93      | +83° | 93            |   |
| 57806.....         | 81      | —    | —             |   |
| 57814.....         | 88      | —    | —             |   |
| 57858.....         | 93      | +85° | 90            |   |
| 57860.....         | 90      | —    | —             |   |
| 57866.....         | 83      | −83° | 83            |   |
| 57870.....         | 87      | −64° | 71            |   |
| 57874.....         | 70      | —    | —             |   |
| 57888.....         | 91      | —    | —             |   |
| 58856.....         | 90      | —    | —             |   |
| 73566.....         | 89      | —    | —             |   |
| 73591.....         | 94      | —    | —             |   |
| 73809.....         | 87      | —    | —             |   |
| 73826.....         | 84      | −86° | 86            |   |
| 73839.....         | 94      | +55° | 100           |   |
| 73840.....         | 98      | +85° | 90            |   |
| 75813.....         | 83      | −85° | 85            |   |
| 75853.....         | 81      | −80° | 82            |   |

The  $Fo\%$  in olivine has been determined by X-ray diffractometry (YODER & SAHAMA, 1957).

The composition of orthopyroxene has been optically determined (DEER, HOWIE & ZUSSMAN, 1963, vol. 2, p. 28).

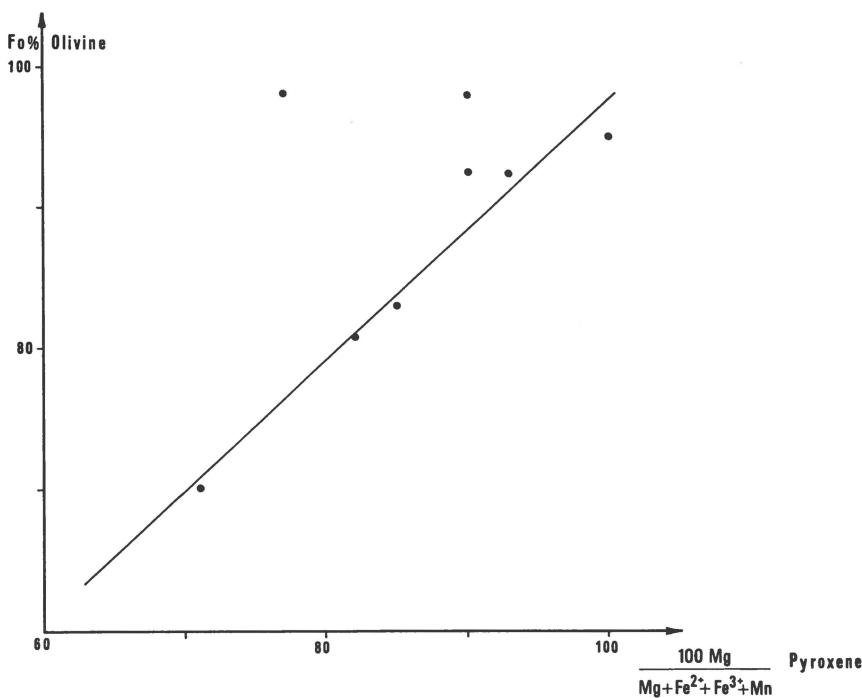
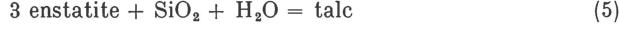


Fig. 13. Plot of Fo % in olivine against 100 Mg/ Mg + Fe<sup>2+</sup> + Fe<sup>3+</sup> + Mn in orthopyroxene from peridotites. The samples plot on a line at approximately 45°. Samples 57753 and 73840, which plot far from the line, are of orthopyroxene from monomineralic veins or masses.

some cases olivine (and the co-existing orthopyroxene and amphibole) appear to be derived from the recrystallisation of a former serpentinite. The serpentine consists of large lamellae of chrysotile completely different in form and shape from those derived from the transformation of the peridotite nucleus (described in more detail below).

*Orthopyroxene* occurs in variable amounts but occurs in two forms: i) normally as crystals of the same size as olivine, and uniformly distributed in the rock; ii) rarely as monomineralic pegmatitic veins or masses in the ultramafic rock. The orthopyroxene composition varies from pure enstatite to hypersthene. The variations of Mg and Fe content in the pyroxene is directly correlated with the variation in olivine (see fig. 13). All the analysed samples plot on a line at approximately 45°, except two (57753 and 73840) in which pyroxene is present as monomineralic veins or masses cutting a dunite. The slope of the correlation line between pyroxene and olivine indicates that the two minerals have crystallised in equilibrium (RAMBERG & DEVORE, 1951; MEDARIS, 1969). Pyroxene has often been almost completely transformed into talc and in smaller amounts into amphibole and serpentine. Talc probably developed by the reaction



(suggested by DEER, HOWIE & ZUSSMAN, 1963, vol. 2, p. 25).

*Amphibole* normally consists of a colourless or faintly pleochroic (pale green) magnesian hornblende. It has a 2V around 90°, and is sometimes positive and some-

times negative. Amphibole is present in the majority of the samples. Its relationship to olivine and orthopyroxene is not clear; in some cases amphibole seems to be a later transformation product, indicating disequilibrium, but in most cases no transformation contacts exist and amphibole exhibits a polygonal form (plate 3b) with 120° junctions indicating equilibrium (RAST, 1965).

*Chlorite*, identified by X-ray diffractometry, is constantly present as large colourless lamellae.

*Carbonate* is a minor, but common, constituent. Large xenoblasts of dolomite or magnesian calcite have crystallised among the other minerals. Carbonate may be related to the transformation of amphibole shown by plate 4a.

Accessory minerals are *phlogopite*, *clinopyroxene*, *spinel* and *opaque minerals*.

The serpentinite which forms the outer transformation zone of the peridotite consists of *chrysotile* and subordinate *antigorite*, with few relics of olivine, orthopyroxene or amphibole. The serpentinite is usually richer in carbonates than the peridotite and may contain some talc.

### Light-green zone

This zone, not always present, occurs adjacent to either the serpentinite or the peridotite. It consists of a fine-grained rock with two main mineral assemblages. They are: 1) *amphibole-chlorite* (-*phlogopite*) (-*anthophyllite*) and 2) *talc-chlorite* (-*anthophyllite*). The former usually occurs as an inner shell, while the latter is more common as an outer zone.

#### *Amphibole-chlorite assemblage.*

The *amphibole* is very similar to that found in the nucleus. It is a faintly pleochroic type, sometimes replaced by talc, carbonates and chlorite. Its chemical composition can be seen in Table 7. An amphibole having a similar composition and classified as 'hornblende' has been reported by DEER, HOWIE & ZUSSMAN (1963, vol. 2, p. 276) as occurring in a hornblende-mica peridotite. Other occurrences of magnesian hornblendes have been described by several authors (see ESKOLA, 1952; LOSCHI GHITTONI, 1968). By comparing the present hornblende with that analysed by LOSCHI GHITTONI from a peridotite in granulite facies, it may be seen that the hornblende of sample 57867 has lower hexacoordinate Al and higher Mn and Fe.

*Phlogopite* occurs as small flakes. Its analysis is presented in Table 7. It has been classified as a phlogopite because the Mg/Fe ratio is higher than 2:1 (DEER, HOWIE & ZUSSMAN, 1963, vol. 3, p. 42).

*Anthophyllite* occurs rarely as prismatic crystals which cut across all other minerals.

#### *Talc-chlorite assemblage.*

*Talc*, with subordinate *chlorite* is predominant. *Anthophyllite* is more abundant than in the amphibole-chlorite assemblage but it is still scarce and may replace both talc and chlorite.

### Anthophyllite zone

This is a peculiar shell which occurs towards the rim of the ultramafic body. The same association of minerals has also been found to cut

Table 7. *Chemical analyses of minerals of ultramafic lenses.*

| GGU No.                               | Light-green zone   |                               | Anthophyllite zone     | Hornblendite zone       |
|---------------------------------------|--------------------|-------------------------------|------------------------|-------------------------|
|                                       | 57867<br>amphibole | 57867<br>phlogopite           | 57896<br>anthophyllite | 57898<br>hornblende     |
| Oxides wt. %/o                        |                    |                               |                        |                         |
| SiO <sub>2</sub> .....                | 47.64              | 39.78                         | 56.68                  | 46.07                   |
| TiO <sub>2</sub> .....                | 0.70               | 0.90                          | tr                     | 0.55                    |
| Al <sub>2</sub> O <sub>3</sub> .....  | 8.54               | 14.94                         | 1.33                   | 10.51                   |
| Fe <sub>2</sub> O <sub>3</sub> .....  | 1.35               | 1.50                          | 1.08                   | 5.10                    |
| FeO .....                             | 6.53               | 6.40                          | 10.20                  | 7.48                    |
| MnO .....                             | 0.12               | 0.04                          | 0.26                   | 0.22                    |
| CaO .....                             | 9.73               | 0.60                          | 0.42                   | 12.09                   |
| MgO .....                             | 19.55              | 21.92                         | 28.15                  | 15.42                   |
| Na <sub>2</sub> O .....               | 1.14               | 0.18                          | tr                     | 0.80                    |
| K <sub>2</sub> O .....                | 0.67               | 9.00                          | tr                     | 0.34                    |
| P <sub>2</sub> O <sub>5</sub> .....   | -                  | -                             | tr                     | 0.05                    |
| H <sub>2</sub> O <sup>+</sup> .....   | 2.79               | 4.53                          | 2.26                   | 1.85                    |
| H <sub>2</sub> O <sup>-</sup> .....   | 0.45               | 0.75                          | 0.21                   | 0.13                    |
|                                       | 99.21              | 100.54                        | 100.59                 | 100.61                  |
| Number of ions on the basis of 24 (0) |                    |                               |                        |                         |
| Si .....                              | 6.770 } 8.000      | 5.660 } 8.000                 | 7.746 } 7.959          | 6.623 } 8.000           |
| Al .....                              | 1.230 } 1.230      | 2.340 } 0.213                 | 1.377 } 7.959          | 0.403 } 8.000           |
| Al .....                              | 0.200 } 0.200      | 0.166 } -                     | - } 0.060              | - } 0.060               |
| Ti .....                              | 0.075 } 0.075      | 0.097 } -                     | - } 0.522              | - } 0.522               |
| Fe <sup>3+</sup> .....                | 0.144 } 5.351      | 0.161 } 5.835                 | 0.110 } 5.732          | 5.213 } 3.303           |
| Mg .....                              | 4.140 } 5.351      | 4.646 } 5.835                 | 1.166 } 5.732          | 0.898 } 3.303           |
| Fe <sup>2+</sup> .....                | 0.777 } 0.015      | 0.761 } 0.004                 | 7.099 } 0.029          | 0.027 } 0.027           |
| Mn .....                              | 0.015 } 0.316      | 0.004 } 0.048                 | - } 0.029              | 0.221 } 0.221           |
| Na .....                              | 0.316 } 1.481      | 0.048 } 1.919                 | - } 1.771              | 1.862 } 2.144           |
| Ca .....                              | 1.481 } 0.122      | 0.091 } 1.632                 | 0.062 } -              | 0.061 } 0.061           |
| OH .....                              | 2.626 } 2.626      | 4.324 } 2.036                 | - } 2.036              | 1.799 } 1.799           |
| 2V .....                              | +88°               | nd                            | +89°                   | -81°                    |
| c:γ .....                             | 18°                |                               |                        | 18°                     |
| α .....                               | nd<br>(colourless) | nd<br>(light yellow)          | 1.622<br>(colourless)  | 1.646<br>(yellow)       |
| β .....                               | nd<br>(colourless) | nd<br>(light brownish<br>red) | 1.644<br>(colourless)  | 1.659<br>(green)        |
| γ .....                               | nd<br>(colourless) | nd<br>(light brownish<br>red) | 1.649<br>(colourless)  | 1.672<br>(bluish green) |

nd = refractive index not determined.

the ultramafic mass as an irregular network of veins. It consists of a zone, from 1 to 20 cm wide, of asbestos crystals of anthophyllite in a radial arrangement. According to WALTON (1966), this feature indicates a post-tectonic crystallisation. Its chemical composition can be seen in Table 7. Anthophyllite replaces talc with which it is often associated, but subsequently it is itself partially transformed into talc.

### Hornblendite zone

This is the outermost zone of the ultramafics and is directly enclosed by the amphibolite. The passage from hornblendite to amphibolite takes place within a few centimetres by progressive enrichment in salic minerals. The inner contact of the hornblendite zone is separated from the anthophyllite zone by a thin shell of a talc-amphibole assemblage in which the green hornblende appears. The hornblendite shell has a fairly constant thickness of about 20 or 30 cm regardless of the total size of the ultrabasic body of which it forms part. Normally it consists of more than 90 % hornblende. Small amounts of ilmenite, sphene, quartz, biotite and apatite may also be present. More rarely the hornblende is accompanied by large amounts of clinopyroxene.

The *hornblende* is greenish to greenish brown in colour. Its chemical composition is represented in Table 7. Pyroxene is present in only two of the examined samples. It is an augite or a faintly pleochroic diopside varying from colourless to pale green. Hornblende seems to have crystallised later than, and as a transformation product from, clinopyroxene, because it occurs as a rim around the pyroxene or along its cleavages.

When in contact with pegmatites, the hornblendite rim is partially transformed into biotite. Sometimes tourmaline, and in one case even beryl, occur in conjunction with biotite.

### Chemistry

Seven samples of ultramafics have been chemically analysed and the results are reported in Table 8. Some of them represent zones of the same lens. In the diagram of fig. 14 the various oxides are plotted against  $\text{SiO}_2$  to show the variations between the different zones of the same lens. Passing from the nucleus to the green zone, there is an increase in  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{FeO}$ ,  $\text{CaO}$ ,  $\text{Na}_2\text{O}$  and a decrease in  $\text{Fe}_2\text{O}_3$ ,  $\text{MgO}$ ,  $\text{K}_2\text{O}$ . In the hornblendite zone,  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{CaO}$  and  $\text{Na}_2\text{O}$  increase further whilst  $\text{Fe}_2\text{O}_3$  also increases, and  $\text{FeO}$  and  $\text{MgO}$  decrease in respect to the green zone. The variations between the nucleus and the green zone, although

Table 8. *Chemical composition and mineral assemblages of ultramafic rocks.*

| GGU No.                              | 57858 | 57862  | 57866  | 57867 | 57888 | 57889 | 57890 |
|--------------------------------------|-------|--------|--------|-------|-------|-------|-------|
| Oxides wt. %                         |       |        |        |       |       |       |       |
| SiO <sub>2</sub> .....               | 41.75 | 39.36  | 45.07  | 49.13 | 41.90 | 46.71 | 47.65 |
| TiO <sub>2</sub> .....               | 0.49  | 0.49   | 0.58   | 0.70  | 0.34  | 0.42  | 0.48  |
| Al <sub>2</sub> O <sub>3</sub> ..... | 2.93  | 3.69   | 6.25   | 6.85  | 5.22  | 9.30  | 11.08 |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 2.55  | 7.94   | 2.16   | 1.23  | 2.75  | 0.35  | 1.10  |
| FeO .....                            | 9.88  | 7.17   | 6.39   | 8.22  | 6.67  | 9.28  | 8.83  |
| MnO .....                            | 0.21  | 0.17   | 0.21   | 0.16  | 0.16  | 0.18  | 0.20  |
| CaO .....                            | 2.66  | 3.61   | 5.96   | 7.67  | 5.02  | 7.43  | 10.97 |
| MgO .....                            | 38.15 | 30.59  | 28.02  | 22.26 | 30.95 | 21.67 | 16.27 |
| Na <sub>2</sub> O .....              | 0.01  | 0.17   | 0.48   | 0.78  | 0.28  | 0.81  | 1.28  |
| K <sub>2</sub> O .....               | 0.01  | 0.01   | 1.08   | 0.75  | 0.56  | 0.05  | 0.13  |
| P <sub>2</sub> O <sub>5</sub> .....  | 0.05  | 0.06   | 0.03   | 0.02  | 0.03  | 0.03  | 0.03  |
| CO <sub>2</sub> .....                | —     | 0.34   | —      | —     | 0.45  | 0.20  | —     |
| H <sub>2</sub> O <sup>+</sup> .....  | 1.19  | 6.89   | 3.84   | 2.09  | 5.02  | 3.02  | 1.36  |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.07  | 0.14   | 0.11   | 0.05  | 0.09  | 0.08  | 0.08  |
|                                      | 99.95 | 100.63 | 100.18 | 99.91 | 99.44 | 99.53 | 99.46 |

## Mineral assemblages

57858: Amphibole saxonite. Olivine (70–80 %), orthopyroxene (5–10 %), amphibole (5 %), chlorite (5 %), opaque.

57862: Serpentinite. Chrysotile (70 %), amphibole (10–20 %), chlorite (5–10 %), carbonates, opaque.

57866: Amphibole saxonite. Olivine (30–40 %), orthopyroxene (10–20 %), amphibole (30–40 %), talc, chrysotile, chlorite.

57867: "Green zone". Amphibole (80–90 %), phlogopite (5 %), chlorite, opaque.

57888: Amphibole peridotite. Olivine (40–50 %), amphibole (30–40 %), chlorite, chrysotile (10–20 %), opaque, carbonate.

57889: "Green zone". Colourless Mg-amphibole (40–50 %), green hornblende (40–45 %), chlorite (10–20 %), opaque, carbonates.

57890: Hornblendite. Pale green hornblende.

57866, 57867: nucleus and "green zone" respectively of the same ultramafic.

57888, 57889, 57890: nucleus, "green zone", hornblendite zone respectively of the same ultramafic.

having different absolute values, are similar in both the occurrences examined.

It is reasonable to believe that the chemical variations shown above are a consequence of metasomatic exchanges with the country rock.

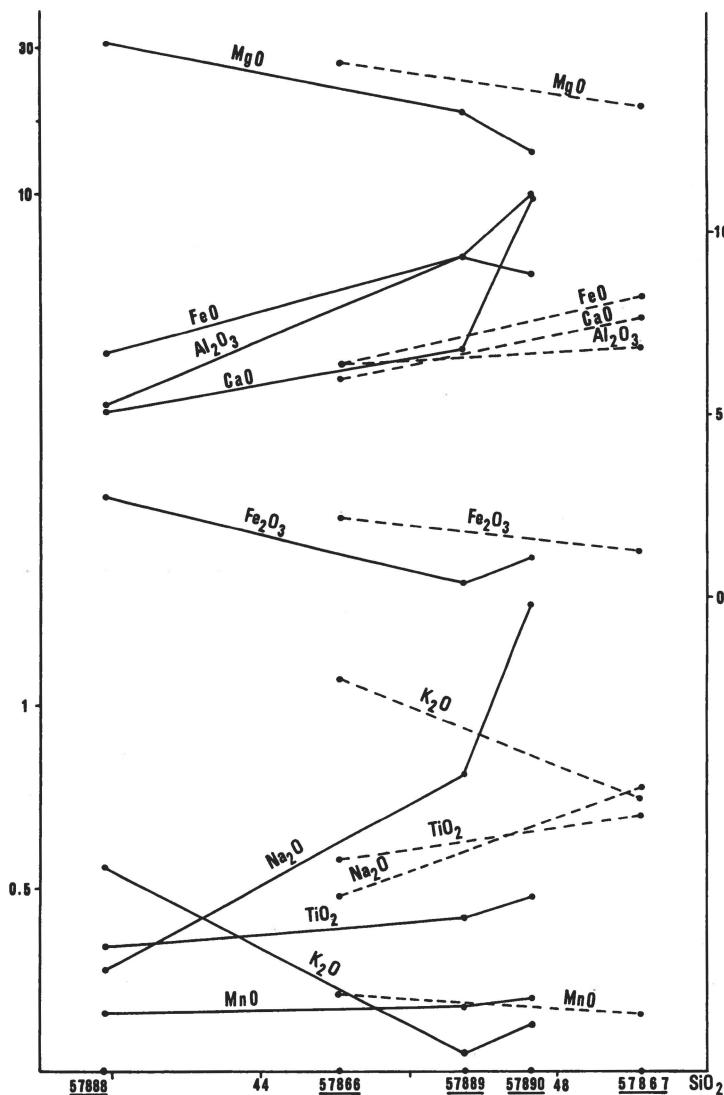


Fig. 14. Chemical variations of the major oxides against  $\text{SiO}_2$  in ultramafic lenses. Samples 57888, 57889 and 57890 are the nucleus, green zone and hornblendite zone, respectively, of the same lens. Samples 57866 and 57867 are the nucleus and green zone of the same lens.

### Rusty schist

#### Description

Rusty coloured rocks are frequently seen in the field. This coloration is most often due to a superficial alteration of mafic rocks: amphibolites, ultramafics and sometimes dykes. But in one case the rusty-brown colour helps one to recognise a lithological type: the rusty schist. The colour is in this case mainly due to the weathering of biotite. However, small amounts of sulphides such as pyrite, chalcopyrite and pyrrhotite, have been found, and their alteration contributes to the colour of the rock surface. The rusty schist forms thin, normally single, layers of rock which lie conformably within the concordant amphibolites. These rusty horizons are, in spite of being thin, very continuous; in the amphibolite complex which passes north of mountain 933 m, rusty schist is found inside the amphibolite for a distance of several tens of kilometres with a maximum thickness of only 20 m. Thus it provides a good marker horizon. In other cases the thin rusty schist band is limited to an intercalation a few centimetres or tens of centimetres wide.

The importance of the rusty schist arises also from the fact that it shows the best preserved primary lithological contacts in the area: those between rusty schist and amphibolite, and between quartzite layers and more pelitic layers within the rusty horizons. These lithological distinctions, whose primary origin can hardly be doubted, are preserved because the enclosing, comparatively resistant, amphibolite has protected the schist from the affects of migmatisation and consequently metamorphism of the schist has taken place in a closed system. Where the protective amphibolite layer is lacking, the rusty schist passes gradually into a normal biotite and garnet-bearing gneiss.

### Petrography

There are several pointers to the metasedimentary origin of the rusty schist. The rock consists of a regular alternation of beds whose composition varies from pure quartzite to argillaceous sandstone. Graphite is common, mainly in the beds which closely approach pelite in composition. The mineral assemblages are varied, reflecting both original differences and metamorphic differentiation. The most important assemblages, excluding the almost monomineralic quartzites, are: 1) a *biotite-garnet-sillimanite assemblage* consisting of quartz, plagioclase, biotite, garnet,  $\pm$  sillimanite, (K-feldspar, muscovite, epidote, rutile); and 2) an *anthophyllite assemblage* with quartz, plagioclase, anthophyllite (or hornblende), (muscovite, epidote, rutile). The first assemblage, which

best reflects the original composition, is found in layers regularly alternating with quartzites and is by far the most common. The second, which is rare, is derived from the first through transformations described later.

Table 9. *Chemical and modal composition of rusty schists.*

| GGU No.                              | biotite-garnet-sillimanite assemblage |       |       |       |       |       |       |
|--------------------------------------|---------------------------------------|-------|-------|-------|-------|-------|-------|
|                                      | 57820                                 | 57856 | 58884 | 73590 | 73859 | 75805 | 75842 |
| <b>Oxides wt. %</b>                  |                                       |       |       |       |       |       |       |
| SiO <sub>2</sub> .....               | 60.24                                 | 61.77 | 64.70 | 56.85 | 65.82 | 66.11 | 67.39 |
| TiO <sub>2</sub> .....               | 1.15                                  | 2.20  | 0.51  | 1.11  | 0.49  | 0.69  | 0.40  |
| Al <sub>2</sub> O <sub>3</sub> ..... | 17.01                                 | 18.03 | 16.56 | 19.75 | 16.94 | 17.20 | 16.43 |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 0.02                                  | 1.01  | 2.50  | 1.27  | 0.89  | 1.35  | 0.81  |
| FeO .....                            | 8.03                                  | 7.11  | 1.62  | 8.32  | 2.40  | 3.87  | 1.79  |
| MnO .....                            | 0.18                                  | 0.23  | 0.07  | 0.24  | 0.04  | 0.03  | 0.03  |
| CaO .....                            | 4.20                                  | 3.71  | 2.59  | 4.10  | 4.98  | 3.36  | 4.84  |
| MgO .....                            | 3.53                                  | 2.17  | 1.96  | 2.85  | 1.97  | 1.81  | 1.91  |
| Na <sub>2</sub> O .....              | 2.36                                  | 2.11  | 4.12  | 2.16  | 3.69  | 2.51  | 2.54  |
| K <sub>2</sub> O .....               | 1.23                                  | 0.24  | 1.36  | 1.12  | 1.23  | 0.83  | 1.20  |
| P <sub>2</sub> O <sub>5</sub> .....  | 0.02                                  | 0.09  | 0.01  | 0.02  | 0.03  | 0.03  | 0.03  |
| H <sub>2</sub> O <sup>+</sup> .....  | 1.69                                  | 0.67  | 3.04  | 1.50  | 0.72  | 1.43  | 1.82  |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.12                                  | 0.13  | 0.20  | 0.13  | 0.12  | 0.13  | 0.13  |
|                                      | 99.78                                 | 99.47 | 99.24 | 99.42 | 99.32 | 99.35 | 99.32 |
| <b>Mode vol. %</b>                   |                                       |       |       |       |       |       |       |
| quartz .....                         | 12.5                                  | 21.8  | 16.3  | 11.1  | 23.7  | 20.8  | 29.0  |
| plagioclase .....                    | 56.4                                  | 57.2  | 42.6  | 50.1  | 61.6  | 50.2  | 54.4  |
| K-feldspar .....                     | 0.4                                   | 0.1   | 0.1   | —     | —     | —     | 0.2   |
| biotite .....                        | 15.6                                  | 6.0   | 17.6  | 20.1  | 13.1  | 13.8  | 5.7   |
| garnet .....                         | 12.0                                  | 10.7  | 12.2  | 11.7  | 0.6   | 4.9   | —     |
| sillimanite .....                    | —                                     | 3.4   | 7.6   | 4.0   | —     | 5.5   | —     |
| staurolite .....                     | —                                     | 0.1   | 0.2   | tr    | —     | —     | —     |
| amphibole .....                      | —                                     | —     | —     | —     | —     | —     | —     |
| cordierite .....                     | —                                     | —     | tr    | —     | —     | —     | —     |
| kyanite .....                        | —                                     | —     | —     | —     | —     | —     | —     |
| muscovite .....                      | —                                     | —     | 0.3   | 1.0   | 0.1   | 3.6   | 1.0   |
| chlorite .....                       | tr                                    | —     | 1.2   | tr    | 0.2   | —     | 5.7   |
| epidote .....                        | 0.3                                   | —     | 0.4   | —     | 0.2   | —     | 2.9   |
| calcite .....                        | —                                     | —     | —     | —     | —     | —     | —     |
| rutile .....                         | 0.1                                   | 0.1   | 0.1   | 0.2   | —     | —     | 0.1   |
| tourmaline .....                     | tr                                    | tr    | tr    | —     | —     | —     | —     |
| zircon .....                         | tr                                    | tr    | tr    | 0.1   | tr    | tr    | tr    |
| apatite .....                        | 0.2                                   | —     | —     | —     | 0.1   | —     | 0.1   |
| opaque .....                         | 2.4                                   | 0.5   | 1.3   | 1.6   | 0.4   | 1.2   | 0.8   |
| <b>An % in</b>                       |                                       |       |       |       |       |       |       |
| plagioclase....                      | 39                                    | 38    | 31    | 36    | 31    | 31    | 50    |

(continued)

Table 9 (continued).

| GGU No.                              | biotite-garnet-sillimanite assemblage |       |        |       | anthophyllite assemblage |       |
|--------------------------------------|---------------------------------------|-------|--------|-------|--------------------------|-------|
|                                      | 75880                                 | 75884 | 75892  | 75898 | 75899                    | 75900 |
| <b>Oxides wt. %</b>                  |                                       |       |        |       |                          |       |
| SiO <sub>2</sub> .....               | 64.71                                 | 58.93 | 57.22  | 67.00 | 59.05                    | 59.53 |
| TiO <sub>2</sub> .....               | 0.50                                  | 1.26  | 0.70   | 0.43  | 1.45                     | 1.59  |
| Al <sub>2</sub> O <sub>3</sub> ..... | 15.06                                 | 18.47 | 18.00  | 14.84 | 18.25                    | 18.41 |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 0.55                                  | 0.29  | 1.72   | 1.22  | 0.28                     | 0.49  |
| FeO.....                             | 5.03                                  | 5.46  | 10.21  | 2.73  | 3.58                     | 2.67  |
| MnO .....                            | 0.27                                  | 0.18  | 0.53   | 0.12  | 0.09                     | 0.06  |
| CaO.....                             | 5.68                                  | 6.03  | 4.07   | 5.75  | 9.74                     | 9.95  |
| MgO .....                            | 3.12                                  | 3.43  | 5.14   | 2.37  | 5.24                     | 5.04  |
| Na <sub>2</sub> O.....               | 0.86                                  | 3.00  | 0.31   | 1.04  | 0.58                     | 0.34  |
| K <sub>2</sub> O.....                | 1.62                                  | 1.11  | 0.25   | 1.96  | 0.07                     | 0.06  |
| P <sub>2</sub> O <sub>5</sub> .....  | 0.04                                  | 0.06  | 0.01   | 0.04  | 0.02                     | 0.04  |
| H <sub>2</sub> O <sup>+</sup> .....  | 1.80                                  | 1.06  | 1.65   | 1.55  | 1.12                     | 1.08  |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.14                                  | 0.08  | 0.23   | 0.13  | 0.08                     | 0.08  |
|                                      | 99.38                                 | 99.36 | 100.04 | 99.18 | 99.55                    | 99.34 |
| <b>Mode vol. %</b>                   |                                       |       |        |       |                          |       |
| quartz.....                          | 39.3                                  | 14.4  | a      | 37.8  | 30.8                     | 32.2  |
| plagioclase.....                     | 34.5                                  | 59.3  | va     | 36.7  | 47.7                     | 39.2  |
| K-feldspar .....                     | 0.1                                   | —     | —      | —     | 1.1                      | —     |
| biotite.....                         | 21.0                                  | 16.0  | a      | 17.5  | 8.9                      | —     |
| garnet.....                          | 2.3                                   | 5.0   | a      | 1.4   | —                        | —     |
| sillimanite .....                    | —                                     | —     | a      | —     | —                        | —     |
| staurolite .....                     | —                                     | 0.2   | —      | —     | —                        | —     |
| amphibole .....                      | —                                     | —     | —      | —     | 2.8                      | 23.4  |
| cordierite .....                     | —                                     | —     | tr     | —     | 0.3                      | —     |
| kyanite .....                        | —                                     | —     | tr     | —     | —                        | —     |
| muscovite.....                       | 0.2                                   | 0.1   | —      | 2.1   | —                        | —     |
| chlorite .....                       | —                                     | 1.6   | —      | —     | 5.9                      | 1.1   |
| epidote .....                        | —                                     | 1.5   | —      | 2.9   | 0.9                      | 3.1   |
| calcite .....                        | —                                     | —     | —      | 0.4   | —                        | —     |
| rutile.....                          | —                                     | 0.2   | tr     | tr    | 1.1                      | 0.7   |
| tourmaline .....                     | 0.1                                   | —     | a      | —     | —                        | —     |
| zircon .....                         | tr                                    | tr    | —      | tr    | tr                       | —     |
| apatite .....                        | 0.6                                   | 0.4   | —      | tr    | 0.4                      | —     |
| opaque .....                         | 1.9                                   | 1.2   | —      | 1.1   | —                        | 0.2   |
| An % in plagioclase ...              | 80-60                                 | 48    | 89     | 80-62 | 89                       | 89    |

75892: modal analysis not made owing to the irregular grain size and distribution of the minerals, (va = very abundant; a = abundant; tr = traces).

75899: anthophyllite. 2V = + 79°.

75900: hornblende. 2V = 90°; c: γ = 16°.

75880 and 75898: zoned plagioclase. The zones with An 80 % are micro-antiperthitic.

Table 10. Chemical analyses of co-existing garnets, biotites and of an amphibole from the rusty schists.

| GGU No.                              | Garnets |       |        |        | Biotites |       |       |        | Amphibole<br>75900 |
|--------------------------------------|---------|-------|--------|--------|----------|-------|-------|--------|--------------------|
|                                      | 57820   | 57856 | 75880  | 75884  | 57820    | 57856 | 75880 | 75884  |                    |
| <b>Oxides wt. %</b>                  |         |       |        |        |          |       |       |        |                    |
| SiO <sub>2</sub> .....               | 37.32   | 39.16 | 38.01  | 39.78  | 35.40    | 35.73 | 38.05 | 37.90  | 47.96              |
| TiO <sub>2</sub> .....               | 0.67    | 0.25  | 0.15   | 0.50   | 2.61     | 1.70  | 1.65  | 2.00   | 1.36               |
| Al <sub>2</sub> O <sub>3</sub> ..... | 22.74   | 20.57 | 25.99  | 21.72  | 17.84    | 18.41 | 18.38 | 18.22  | 14.09              |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 1.59    | 0.26  | 1.61   | 1.74   | 2.37     | 2.77  | 2.31  | 1.92   | 1.79               |
| FeO .....                            | 26.50   | 28.96 | 19.27  | 23.48  | 11.45    | 13.32 | 11.89 | 13.09  | 6.45               |
| MnO .....                            | 1.25    | 1.04  | 10.40  | 0.98   | tr       | tr    | 0.36  | 0.06   | 0.15               |
| CaO .....                            | 2.73    | 3.15  | 3.64   | 2.84   | 0.77     | 0.67  | 0.95  | 1.26   | 10.61              |
| MgO .....                            | 7.21    | 5.64  | 1.41   | 8.84   | 13.81    | 11.62 | 12.10 | 12.50  | 12.23              |
| Na <sub>2</sub> O .....              | 0.07    | 0.05  | 0.09   | 0.03   | 0.18     | 0.31  | 0.12  | 0.28   | 0.69               |
| K <sub>2</sub> O .....               | 0.07    | 0.04  | 0.13   | 0.05   | 8.00     | 7.90  | 8.64  | 7.94   | 0.17               |
| H <sub>2</sub> O <sup>+</sup> .....  | —       | —     | —      | —      | 5.26     | 4.58  | 4.59  | 4.56   | 2.68               |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.33    | 0.12  | 0.25   | 0.04   | 1.58     | 1.00  | 0.40  | 0.70   | 0.69               |
|                                      | 100.48  | 99.24 | 100.95 | 100.00 | 99.27    | 98.01 | 99.44 | 100.43 | 98.87              |

Number of ions on the basis of 24(0)

|                  |       |       |       |       |       |       |       |       |       |       |       |       |       |       |       |       |       |       |       |
|------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| Si               | 5.794 | 6.000 | 6.172 | 5.885 | 6.000 | 6.067 | 5.220 | 8.000 | 5.401 | 8.000 | 5.573 | 8.000 | 5.523 | 8.000 | 6.780 | 8.000 |       |       |       |
| Al               | 0.206 | —     | —     | 0.115 | 6.000 | —     | 2.780 | —     | 2.599 | —     | 2.427 | —     | 2.477 | —     | 1.220 | —     |       |       |       |
| Al               | 3.957 | —     | 3.822 | —     | 4.629 | —     | 3.905 | —     | 0.320 | —     | 0.680 | —     | 0.746 | —     | 0.651 | —     |       |       |       |
| Ti               | 0.077 | 4.220 | 0.045 | 3.897 | 0.018 | 4.835 | 0.058 | 4.163 | —     | 0.289 | —     | 0.194 | —     | 0.182 | —     | 0.220 | —     |       |       |
| Fe <sup>3+</sup> | 0.186 | —     | 0.030 | —     | 0.188 | —     | 0.200 | —     | 0.263 | —     | 0.315 | —     | 0.255 | —     | 0.210 | —     |       |       |       |
| Mg               | 1.669 | —     | 1.324 | —     | 0.326 | —     | 2.008 | —     | 3.035 | 5.319 | 2.529 | 5.401 | 2.642 | 5.325 | 2.714 | 5.397 | —     |       |       |
| Fe <sup>2+</sup> | 3.440 | —     | 3.816 | —     | 2.495 | —     | 2.994 | —     | 1.412 | —     | 1.683 | —     | 1.455 | —     | 1.594 | —     | 0.762 | —     |       |
| Mn               | 0.165 | —     | 0.139 | —     | 1.365 | —     | 0.126 | —     | —     | —     | —     | —     | 0.045 | —     | 0.008 | —     | 0.017 | —     |       |
| Na               | 0.020 | 5.783 | 0.016 | 5.834 | 0.027 | 4.843 | 0.007 | 5.608 | —     | 0.050 | —     | 0.091 | —     | 0.034 | —     | 0.080 | —     | 0.189 | —     |
| Ca               | 0.453 | —     | 0.531 | —     | 0.604 | —     | 0.464 | —     | 0.121 | 1.676 | 0.079 | 1.694 | 0.150 | 1.798 | 0.196 | 1.751 | —     | 1.607 | 1.827 |
| K                | 0.036 | —     | 0.008 | —     | 0.026 | —     | 0.009 | —     | 1.505 | —     | 1.524 | —     | 1.614 | —     | 1.475 | —     | 0.031 | —     | —     |
| OH               | —     | —     | —     | —     | —     | —     | —     | —     | 5.186 | —     | 4.596 | —     | 4.452 | —     | 4.430 | —     | 2.528 | —     | —     |

|                     |      |      |      |      |                  |             |             |             |             |              |
|---------------------|------|------|------|------|------------------|-------------|-------------|-------------|-------------|--------------|
| almandine . . . .   | 60.1 | 65.7 | 52.1 | 53.5 | $\alpha$         | pale yellow | pale yellow | pale yellow | pale yellow | colourless   |
| andradite . . . .   | 6.9  | 1.9  | 6.5  | 6.9  | $\beta = \gamma$ | reddish     | reddish     | brown       | reddish     | pale greyish |
| pyrope . . . .      | 29.1 | 22.8 | 6.8  | 35.9 |                  | brown       | brown       | brown       | brown       | brown        |
| grossular . . . .   | 1.0  | 7.2  | 7.1  | 1.4  |                  |             |             |             |             | 2V 90°       |
| spessartite . . . . | 2.9  | 2.4  | 28.5 | 2.3  |                  |             |             |             |             | c:γ 16°      |

*Biotite-garnet-sillimanite assemblage.*

Modal analyses of samples belonging to this group are reported in Table 9. The minerals have the following characteristics:

*Quartz* is always abundant and consists of crystalloblasts which show undulose extinction.

*Plagioclase* occurs in variable amounts usually as fresh, twinned crystals. Its composition varies between  $An_{31}$  and  $An_{89}$ . The CaO content of the rock does not vary with increasing An content but there is a significant decrease in  $Na_2O$  (Table 9). The more anorthitic feldspars present an irregular recrystallisation with zones of  $An_{80}$  and  $An_{60}$  (see samples 75880 and 75898, Table 9, plate 4b). The parts with 80 % An are micro- (almost crypto-) antiperthitic (plate 4b) which is unusual in such calcic plagioclases. A similar phenomenon (anthiperthites in calcic plagioclases) has been described in granulite facies rocks of the Alps (Capedri, 1968). Rusty schists with plagioclases containing 80 % An are limited to the amphibolite of mountain 933 m, at the zone between mountain 880 m and mountain 760 m in the eastern part of the area.

*K-feldspar* is normally rare and occurs as microcline. But small, roundish patches of orthoclase occur, however, inside the microperthitic plagioclase described above.

*Biotite* is the commonest mafic mineral. It is pleochroic from  $\alpha$  = yellow or colourless, to  $\beta$  =  $\gamma$  = reddish brown. Analyses of four biotites co-existing with garnet are reported in Table 10. Biotite has complex relationships with garnet and sillimanite (described below under the sections on garnet and sillimanite). It is sometimes altered to chlorite in association with the alteration of garnet and plagioclase as a consequence of local retrogression due to faulting. It may also be associated with large flakes of muscovite.

*Garnet* presents some interesting features. It occurs both as deformed porphyroblasts (elongated in the plane of the foliation) that are strongly poikilitic and corroded by quartz and feldspar, and as crystals that are much more compact and poor in inclusions and look undeformed, yet which still have corroded rims. The two types may occur together in the same thin section. The compact, more idioblastic, garnet may occur as an outer zone around the strongly poikilitic type. Moreover atoll structures, such as those described by Rast (1965), are sometimes present (plate 5a). Biotite presents complex relationships to garnet. The following observations have been made:

a) garnet has grown without pushing aside the biotite matrix, whereas the biotite laminae are interrupted without any deformation by garnet: b) biotite occurs as inclusions inside garnet, which should not be expected since, according to Ramberg (1952, p. 134), the interfacial energy between garnet and (001) in biotite is high; c) biotite occurs as veins cutting across garnet; d) biotite is pushed apart and wraps around the growing garnet.

Of particular interest are the inclusions found in garnet. They consist of quartz, plagioclase, biotite, staurolite, sillimanite, rutile and opaque minerals. The quartz grains forming the inclusions are smaller than both the quartz grains of the matrix and of the quartz which corrodes the garnet. The same holds good for feldspar. Sometimes quartz and feldspar, both singly and as aggregates, may present a pseudo-idioblastic hexagonal or polygonal shape inside garnet (plate 5b). Sillimanite is a rare inclusion, but in one case, biotite, sillimanite and fibrolite occur in the inner part of an atoll garnet (plate 7a). In this case fibrolite is clearly formed on biotite and after

sillimanite. Opaque minerals are represented by granules of ilmenite either orientated in planes which are parallel to the foliation of the rock, indicating that garnet crystallisation post-dates the development of the S-surface (TURNER & WEISS, 1963, p. 208), or having a concentrical arrangement (plate 6a & b).

Analyses of garnets are reported in Table 10.

*Sillimanite* occurs in variable amounts. It may be associated with biotite, muscovite, or with sericite. The association with sericite is clearly due to low metamorphic grade, possibly deuteritic transformation, but the association with muscovite has important metamorphic implications which will be discussed later on (page 64). Fibrolite is uncommon, but in a few cases has been derived from large crystals of sillimanite (plate 7a).

*Staurolite* has been observed in small amounts in only three of the samples examined (Table 9). It always occurs as irregular, corroded crystals inside garnet and it therefore appears to be an unstable relic phase.

*Kyanite* has been observed in only one sample as a small crystal. It has not been possible to establish its relation to sillimanite which occurs in the same sample and its significance is therefore doubtful.

*Muscovite* may be present in large amounts. It is always secondary after biotite or sillimanite. Its significance will be discussed later on.

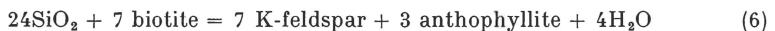
*Rutile* is common as large crystals in most of the samples studied and, apart from ilmenite, represents the sole titanium mineral. It plays an important role in determining the pressure to which these rocks have been subjected.

Accessory minerals include tourmaline (sometimes very abundant), epidote, zircon, apatite and in one case a very small amount of calcite. Opaque minerals include small grains of graphite, pyrrhotite, pyrite, ilmenite and in one case marcasite (57820).

#### *Anthophyllite assemblage.*

A rusty schist with this assemblage has only been found between the mountains 880 m and 760 m in the eastern part of the area (plate 11). Two samples have been taken, one (75899) from the limb and the other (75900) from the hinge of a  $F_4$  fold. Their modal analyses are reported in Table 9. They deserve two separate descriptions because of the different significance of their mineral assemblage.

*Sample 75899* is a greyish rock consisting of the following minerals: *Plagioclase* is very fresh, twinned and with  $An_{90}$ . *Quartz* shows undulose extinction. *Anthophyllite* occurs as altered porphyroblasts and is clearly secondary after biotite. K-feldspar is formed during this reaction and remains either as small inclusions or as crystals wrapped around the anthophyllite. A reaction giving these products is reported by RAMBERG (1952, p. 152):



The  $2V$  is approximately  $+80^\circ$  indicating that iron is present in this amphibole.

*Biotite* is a relic phase. It is represented by the reddish brown type typical of the rusty schist. *Cordierite* is very difficult to identify because it occurs sporadically and is almost completely transformed into a pinitite aggregate. *Rutile* forms large crystals. Some apatite, epidote, and chlorite are present as accessory minerals.

*Sample 75900* differs from sample 75899 owing to the presence of an *amphibole* which in a hand sample is yellowish and in thin section is pleochroic, ranging from

colourless ( $\alpha$ ) to pale greyish brown ( $\beta = \gamma$ ). It forms lineations with the same trend as the  $F_4$  fold axis. Its analysis is reported in Table 10. This hornblende is formed by the reactions

$$\text{anorthite} + \text{anthophyllite} = \text{hornblende} + \text{SiO}_2 \quad (7)$$

*Quartz* is released during this reaction and remains poikilitically included in the hornblende or forms porphyroblasts at its border.

*Biotite* is completely absent. The other characteristics are the same as in sample 75899.

### Chemistry

Thirteen chemical analyses of rusty schists, representing all lithological varieties except quartzite, are presented in Table 9. Eleven analyses are of rocks with a biotite-garnet-sillimanite assemblage and two have an anthophyllite assemblage. Analyses of co-existing biotite and garnet in four samples of rusty schist are presented in Table 10 and discussed together with the metamorphism (p. 62).

The original rocks were sandstone, represented now by quartzites, and more pelitic rocks, represented now by rocks with high  $\text{Al}_2\text{O}_3$  content and with sillimanite. The low alkalis, especially  $\text{K}_2\text{O}$ , indicate, however, that little clay was present in the original rock, and therefore the analyses suggest an original greywacke composition; in particular the analyses are comparable with those of certain rare aluminous greywackes (PETTIGELOW, 1957, p. 306).

There is no correlation between the anorthite content of the plagioclase and the  $\text{CaO}$  content of the rock. There is, however, a negative correlation between the anorthite content of the plagioclase and the  $\text{Na}_2\text{O}$  content of the rock.

The anthophyllite(-cordierite)-bearing rocks have the lowest alkali content and their composition does not correspond with that of any sedimentary or igneous rock.

### Microgranodiorite

#### Description

This is a lithotype seldom met with in the area. It is usually granodioritic or quartz dioritic in composition. It outcrops in the form of small, thin, lenticular bodies, which are normally too small to be mapped, concordantly inside the amphibolites. Only in two cases, in the hinge zones of  $F_4$  folds near mountain 933 m and between mountains 880 and 760 m, were they sufficiently large to be indicated on the map. The microgranodiorite is a fine-grained generally homogeneous rock which in some places possesses nebulitic structures.

Table 11. *Chemical and modal composition of microgranodiorite.*

| GGU No.                              | 57813          | 58887  | 75835 |                   | 57813 | 58887 | 75835 |
|--------------------------------------|----------------|--------|-------|-------------------|-------|-------|-------|
| Oxides wt. %                         |                |        |       |                   |       |       |       |
| SiO <sub>2</sub> .....               | 58.36          | 67.41  | 70.19 | quartz .....      | 6.1   | 30.8  | 33.2  |
| TiO <sub>2</sub> .....               | 0.74           | 0.49   | 0.40  | plagioclase ..... | 68.5  | 57.1  | 49.0  |
| Al <sub>2</sub> O <sub>3</sub> ..... | 19.11          | 15.99  | 15.49 | microcline.....   | —     | 0.4   | 3.2   |
| Fe <sub>2</sub> O <sub>3</sub> ....  | 1.33           | 0.86   | 0.62  | biotite .....     | 14.2  | 7.0   | 10.4  |
| FeO.....                             | 5.88           | 2.76   | 1.60  | garnet .....      | 0.8   | —     | —     |
| MnO.....                             | 0.22           | 0.10   | 0.03  | hornblende .....  | 9.1   | 0.4   | —     |
| CaO.....                             | 5.54           | 5.22   | 2.80  | muscovite .....   | —     | —     | 3.8   |
| MgO.....                             | 2.02           | 2.02   | 1.31  | chlorite .....    | tr    | —     | tr    |
| Na <sub>2</sub> O.....               | 3.70           | 3.64   | 4.52  | epidote .....     | —     | tr    | —     |
| K <sub>2</sub> O.....                | 1.06           | 0.83   | 2.16  | sphene .....      | tr    | 4.0   | —     |
| P <sub>2</sub> O <sub>5</sub> .....  | 0.11           | 0.02   | 0.01  | apatite .....     | 0.3   | 0.1   | 0.2   |
| H <sub>2</sub> O <sup>+</sup> ....   | 1.03           | 0.66   | 0.67  | zircon .....      | 0.1   | tr    | 0.1   |
| H <sub>2</sub> O <sup>-</sup> ....   | 0.13           | 0.09   | 0.06  | rutile .....      | —     | tr    | —     |
|                                      | 99.23          | 100.09 | 99.86 | opaque .....      | 0.9   | 0.2   | 0.1   |
|                                      | An % in plag.. |        | 39    | 39                | 39    | 39    | 29    |

### Petrography

The three modal analyses (Table 11) of this rock indicate that the association quartz, plagioclase, (microcline), biotite, garnet, (hornblende) is the commonest; some of the more interesting mineralogical characteristics are as follows:

*Biotite* is always a reddish brown colour, identical to that of the rusty schist. *Microcline* occurs in variable amounts. *Plagioclase* varies in composition from An<sub>29</sub> to An<sub>39</sub>. *Garnet* only occurs in one sample where it forms pink, idioblastic crystals poor in inclusions. *Hornblende* is the usual bluish green type similar to that encountered in the gneisses. *Rutile* occurs as large crystals as a primary mineral associated with ilmenite. *Epidote* and *muscovite* are common as secondary minerals.

A general retrogression is present in the area of the large N-S fault near the large lake between the mountains 880 and 760 m. Signs of recrystallisation are evident, such as indented rims to some of the minerals, substitutions by microcline, formation of albite, chlorite and epidote.

### Chemistry

Three samples have been chemically analysed and the results are reported in Table 11. Their composition varies from granodiorite to quartz diorite. The chemistry of these rocks is similar to that of migmatitic gneisses bearing the same minerals.

In Table 12 the chemical analyses of co-existing garnet and biotite of one sample are reported. Its significance will be considered later (p. 66).

Table 12. *Chemical composition of co-existing garnet and biotite in a microgranodiorite.*

| GGU No.                              |              |         | 57813                                 |         |                |
|--------------------------------------|--------------|---------|---------------------------------------|---------|----------------|
|                                      | Oxides wt. % |         | Number of ions on the basis of 24 (0) |         |                |
|                                      | garnet       | biotite | garnet                                | biotite |                |
| SiO <sub>2</sub> .....               | 37.16        | 36.18   | Si .....                              | 5.869   | 5.443          |
| TiO <sub>2</sub> .....               | 0.57         | 1.60    | Al .....                              | 6.000   | 8.000          |
| Al <sub>2</sub> O <sub>3</sub> ..... | 20.70        | 18.54   | Al .....                              | 0.131   | 2.557          |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 3.03         | 4.03    | Ti .....                              | 3.724   | 0.730          |
| FeO.....                             | 25.92        | 16.19   | Fe <sup>+3</sup> .....                | 0.067   | 4.152          |
| MnO .....                            | 2.35         | 0.12    | Mg .....                              | 0.361   | 0.181          |
| CaO.....                             | 6.31         | 1.47    | Fe <sup>+2</sup> .....                | 0.983   | 0.457          |
| MgO .....                            | 4.17         | 9.78    | Mn .....                              | 3.424   | 5.611          |
| Na <sub>2</sub> O.....               | 0.10         | 0.22    | Na .....                              | 0.314   | 2.193          |
| K <sub>2</sub> O.....                | 0.02         | 6.40    | Ca .....                              | 5.820   | 2.036          |
| P <sub>2</sub> O <sub>5</sub> .....  | 0.03         | 0.06    | K .....                               | 0.028   | 0.014          |
| H <sub>2</sub> O <sup>+</sup> .....  | —            | 3.94    | OH .....                              | 1.068   | 0.063          |
| H <sub>2</sub> O <sup>—</sup> .....  | 0.17         | 0.72    |                                       | 0.003   | 0.237          |
|                                      | 100.53       | 99.25   |                                       | —       | 1.528          |
|                                      |              |         | almandine .....                       | 59.2    | α pale yellow  |
|                                      |              |         | andradite .....                       | 11.1    | β brownish red |
|                                      |              |         | pyrope .....                          | 17.0    | γ brownish red |
|                                      |              |         | grossular .....                       | 7.3     |                |
|                                      |              |         | spessartite .....                     | 5.4     |                |

## Inclusions in the Gneisses

### Amphibolites

#### Description

The boudins and schlierens of amphibolite which occur in the gneisses can be divided into two categories: 1) those which have been migmatised and are fringed by a marginal zone of hornblende gneiss and 2) boudins which have arisen by the deformation of an amphibole, without evident introduction of new material. The layering which was present in the continuous amphibolite horizons can often be recognised in the boudins.

### Petrography

In the first type, quartz, biotite and feldspar increase progressively from the centre to the rim of the amphibole inclusions. Biotite forms from hornblende, epidote and sphene are other products of this reaction (see p. 18) and the anorthite content of the plagioclase decreases from the

Table 13. *Chemical and modal composition of inclusions in the gneisses.*

| GGU No.                              | Amphibolites |        |       |        |       |        | Calc-silicate inclusion<br>57759 |       |
|--------------------------------------|--------------|--------|-------|--------|-------|--------|----------------------------------|-------|
|                                      | Type 1       |        |       | Type 2 |       |        |                                  |       |
|                                      | 57787A       | 57787B | 57891 | 57721  | 57759 | 58805A | 58805B                           |       |
| Oxides wt. %                         |              |        |       |        |       |        |                                  |       |
| SiO <sub>2</sub> .....               | 45.90        | 52.87  | 57.16 | 47.87  | 48.35 | 44.55  | 50.23                            | 51.29 |
| TiO <sub>2</sub> .....               | 1.23         | 0.38   | 1.95  | 0.81   | 0.72  | 1.30   | 1.21                             | 0.75  |
| Al <sub>2</sub> O <sub>3</sub> ..... | 15.71        | 21.47  | 13.47 | 14.27  | 15.68 | 14.08  | 17.68                            | 15.19 |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 3.99         | 1.52   | 2.11  | 4.12   | 4.23  | 4.43   | 5.51                             | 3.20  |
| FeO.....                             | 8.62         | 3.77   | 5.76  | 8.63   | 7.63  | 10.20  | 3.98                             | 7.39  |
| MnO.....                             | 0.20         | 0.09   | 0.16  | 0.21   | 0.20  | 0.30   | 0.22                             | 0.33  |
| CaO.....                             | 10.97        | 8.94   | 7.71  | 12.62  | 10.67 | 12.35  | 14.62                            | 14.93 |
| MgO.....                             | 8.50         | 2.94   | 8.27  | 7.46   | 7.94  | 8.04   | 3.48                             | 3.48  |
| Na <sub>2</sub> O.....               | 2.40         | 5.40   | 1.00  | 1.35   | 2.46  | 2.04   | 2.10                             | 0.44  |
| K <sub>2</sub> O.....                | 0.28         | 0.21   | 0.53  | 0.33   | 0.58  | 1.00   | 0.47                             | 0.23  |
| P <sub>2</sub> O <sub>5</sub> .....  | 0.04         | 0.02   | 0.05  | 0.06   | 0.04  | 0.02   | 0.02                             | 0.07  |
| CO <sub>2</sub> .....                | 0.57         | 1.10   | nd    | nd     | nd    | nd     | nd                               | 1.35  |
| H <sub>2</sub> O <sup>+</sup> .....  | 0.78         | 0.65   | 1.17  | 1.70   | 0.99  | 1.27   | 1.05                             | 0.80  |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.04         | 0.05   | 0.06  | 0.09   | 0.11  | 0.02   | 0.02                             | 0.03  |
|                                      | 99.23        | 99.41  | 99.40 | 99.52  | 99.60 | 99.60  | 100.59                           | 99.43 |
| mg.....                              | 0.55         | 0.50   | 0.65  | 0.51   | 0.55  | 0.50   | 0.40                             |       |
| ti.....                              | 1.99         | 0.79   | 4.03  | 1.35   | 1.20  | 2.03   | 2.21                             |       |
| Mode vol. %                          |              |        |       |        |       |        |                                  |       |
| quartz.....                          | 0.9          | 0.7    | 20.1  | 1.7    | 1.9   | 0.2    | 6.4                              |       |
| plagioclase.....                     | 23.0         | 70.1   | 50.9  | 15.3   | 33.5  | 12.8   | 12.3                             |       |
| biotite.....                         | 0.3          | 0.7    | 4.9   | —      | —     | 0.1    | —                                |       |
| garnet.....                          | —            | —      | —     | tr     | —     | —      | —                                |       |
| hornblende.....                      | 72.0         | 26.0   | 23.9  | 67.6   | 60.5  | 84.8   | 34.8                             |       |
| clinopyroxene.....                   | —            | —      | —     | 4.6    | 1.4   | —      | 0.5                              |       |
| epidote.....                         | 0.4          | —      | —     | 9.7    | 2.1   | 0.1    | 42.1                             |       |
| sphene.....                          | 0.4          | tr     | tr    | 0.6    | —     | 1.9    | 3.1                              |       |
| apatite.....                         | 0.2          | 0.2    | tr    | 0.2    | 0.1   | tr     | tr                               |       |
| zircon.....                          | 0.1          | 0.1    | tr    | 0.1    | 0.5   | 0.1    | 0.1                              |       |
| calcite.....                         | 1.4          | 1.7    | —     | —      | —     | —      | —                                |       |
| opaque.....                          | 1.2          | 0.5    | 0.1   | 0.2    | —     | tr     | 0.7                              |       |
| An % in                              |              |        |       |        |       |        |                                  |       |
| plagioclase.....                     | 45           | 38     | 36    | 43     | 38    | 39     | 35                               |       |
| hornblende { 2 V -70° -68° -81°      |              |        |       | -72°   | -61°  | -72°   | -57°                             |       |
| c:γ 15° 15° 15°                      |              |        |       | 16°    | 16°   | 14°    | 15°                              |       |
| clinopyroxene { 2 V -                |              |        |       | +67    | +63°  | —      | +64°                             |       |
| c:γ - - -                            |              |        |       | 38     | 47°   | —      | 48°                              |       |

57787 A and B: the inner and outer zone respectively of the same inclusion.

57891: the outer zone of an inclusion.

58805 A and B: different bands of the same inclusion.

57759: analysis carried out on a strip across all the zones of the calc-silicate inclusion.

nucleus to the transformed rim. Modal analyses of samples belonging to this type are reported in Table 13.

The second type of amphibolite inclusion is characterised by the presence of quartz-epidote and quartz-actinolite symplectites (plate 7b). They always occur at the contact between hornblende and plagioclase and are possibly formed by a reaction of the type:



Actinolite is formed by the excess of Mg and Fe and sphene by the Ti present in the reacting hornblende. These symplectites never occur in the major amphibolite horizons, only in the inclusions. This is most probably a consequence of the different environmental conditions ( $P_{\text{H}_2\text{O}}$  and  $P_{\text{CO}_2}$  higher in the migmatitic environment) which caused an instability between calcic plagioclase and hornblende. Modal analyses of samples belonging to this group are reported in Table 13.

### Chemistry

Chemical analyses of the two types of amphibolite inclusions are shown in Table 13.

*First type.* If two samples (57787 A & B) of the same boudin are compared, it can be seen that the more highly transformed one is characterised by an increase of Si, Al and Na and a decrease of Ti,  $\text{Fe}^{3+}$ ,  $\text{Fe}^{2+}$ , Ca and Mg while K does not vary appreciably. An introduction of salic elements is evident.

*Second type.* The chemical composition of the boudins is quite similar to that of the layers of major amphibolites. The boudins therefore acted as a closed system during migmatisation, at least for the major elements.

## Ultramafic inclusions

### Description and petrography

The ultramafic inclusions are serpentinites and hornblendites. The former are very rare, but the latter are quite common. The serpentinite inclusions are small ball-shaped bodies whose occurrence shows that they were closely associated with amphibolites. The serpentinites acted as resistors and have survived even when the amphibolite has been completely transformed by migmatisation. Towards the gneisses they show reaction zones similar to those described for the zoned ultramafic lenses (p. 37) except that the innermost part is here always constituted of serpentinite.

Hornblendite inclusions are very common both in agmatite zones and as isolated boudins in the gneisses. The size of single inclusions varies from a few centimetres to a maximum of 150 centimetres. They have no counterpart in the lithologies of the major layers. They cannot correspond to the uniformly narrow hornblendite layer (30 centimetres thick) found around the ultramafic lenses because the hornblendite inclusions are up to 150 centimetres thick and never enclose a core of different composition. They cannot correspond to the thin hornblendite layers intercalated with amphibolites because there is normally no intercalation in the inclusions and they are much thicker.

The hornblendite inclusions are equigranular, medium- to coarse-grained, and are composed of hornblende (more than 90 %). This hornblende is dark green in a hand sample and is pleochroic in thin section from colourless or yellowish ( $\alpha$ ) to bluish green ( $\gamma$ ). It is sometimes accompanied by significant amounts of diopsidic pyroxene and minor amounts of biotite. Quartz, plagioclase, sphene and opaque minerals are

Table 14. *Chemical composition of a hornblende and a clinopyroxene from a hornblendite inclusion.*

| GGU No.                              | 57720        |                    | Number of ions on the basis of |               |          |
|--------------------------------------|--------------|--------------------|--------------------------------|---------------|----------|
|                                      | Oxides wt. % |                    | 24(0)                          | 6(0)          |          |
|                                      | hornblende   | clinopyro-<br>xene | hornblende                     | clinopyroxene |          |
| SiO <sub>2</sub> .....               | 50.69        | 52.70              | Si .....                       | 7.284         | 1.933    |
| TiO <sub>2</sub> .....               | 0.65         | 0.24               | Al .....                       | 8.000         | 0.055    |
| Al <sub>2</sub> O <sub>3</sub> ..... | 4.71         | 1.27               | Al .....                       | 0.081         | —        |
| Fe <sub>2</sub> O <sub>3</sub> ..... | 0.78         | 2.72               | Ti .....                       | 0.070         | 0.006    |
| FeO .....                            | 9.61         | 4.31               | Fe <sup>3+</sup> .....         | 0.085         | 0.075    |
| MnO .....                            | 0.23         | 0.29               | Mg .....                       | 3.713         | 0.835    |
| CaO .....                            | 12.34        | 23.41              | Fe <sup>2+</sup> .....         | 5.131         | 1.057    |
| MgO .....                            | 17.34        | 15.27              | Mn .....                       | 1.155         | 0.132    |
| Na <sub>2</sub> O .....              | 0.48         | 0.28               | Na .....                       | 0.027         | 0.009    |
| K <sub>2</sub> O .....               | 0.35         | 0.02               | Ca .....                       | 0.136         | 0.020    |
| H <sub>2</sub> O <sup>+</sup> .....  | 2.25         | 0.15               | K .....                        | 1.900         | 0.920    |
| H <sub>2</sub> O <sup>-</sup> .....  | 0.12         | 0.07               | OH .....                       | 2.100         | 0.941    |
|                                      | 99.55        | 100.73             | K .....                        | 0.064         | 0.001    |
| 2V .....                             | —73°         | +57°               | OH .....                       | 2.141         | 0.043    |
| c:γ .....                            | 18°          | 40°                |                                |               |          |
| α .....                              | pale yellow  | colourless         |                                |               | Mg 58.93 |
| β .....                              | green        | pale green         |                                |               | Fe 39.22 |
| γ .....                              | bluish green | pale green         |                                |               | Ca 1.84  |

accessories. Analyses of co-existing hornblende and pyroxene in one sample (Table 14) show that the hornblende is actinolitic, with low  $\text{Al}_2\text{O}_3$ , and that the pyroxene is salite.

## Calc-silicate inclusions

### Description and petrography

Small inclusions of calc-silicate skarns are sometimes present within the gneisses. Inclusions of this kind have already been described in the Frederikshåb district (WALTON, 1966 and BRØGGER SØRENSEN, 1968) both in the gneisses and in the hornblende schists. In the area described here these lenses occur up to tens of centimetres wide. They are characterised by a zonation similar to that described by BRØGGER SØRENSEN. When complete, the zonation comprises (with reference to sample 57769) from the centre of the lens outwards:

- i) *garnet zone* characterised by the presence of a poikilitic pink garnet with inclusions of pink calcite, salitic green diopside, epidote, sphene and quartz,
- ii) the garnet zone gradually passes outwards to a shell in which plagioclase (An 50 %) and scapolite predominate. Scapolite replaces plagioclase. The mafics are green diopside, bluish green hornblende, epidote, and minor amounts of garnet,
- iii) an *outer zone* is composed of plagioclase-diopside-hornblende-epidote-bearing amphibolite.

This complete threefold zonation is rarely seen. In many cases the calc-silicate inclusions consist of only an epidote-rich nucleus sheathed by amphibolite.

The chemical analysis of a calc-silicate skarn is reported in Table 13.

## DISCORDANT AMPHIBOLITE DYKES

### Description

Only two examples of discordant amphibolites have been found and these occur in the eastern part of the area. They have the form of discontinuous dykes, which reach a maximum thickness of one metre and are dark green in colour with a fine-grained, massive texture. However, sometimes inhomogeneities are present as a weak foliation or garnet clusters.

### Petrography

Two thin sections have been examined (73843 and 73844) of samples taken from a discordant amphibolite on the peninsula north of Sermilik avangnardleq. Sample 73843 contains a relic igneous texture, while sample 73844 has a granoblastic texture. In both samples there is a small amount of quartz with undulose extinction. The plagioclase of sample 73843 shows normal zoning (nucleus  $An_{11}$ ; periphery  $An_{33}$ ) while in sample 73844 plagioclase is not zoned and has an anorthite content of 45 %. Hornblende in both samples has pleochroism  $\alpha$  = yellow;  $\beta$  = deep green;  $\gamma$  = greenish blue, but the former has a  $2V$  of  $-75^\circ$  and  $c:\gamma$  angle of  $16^\circ$  whilst the latter has a  $2V$  of  $-69^\circ$  and  $c:\gamma$  angle of  $17^\circ$ . A secondary transformation into actinolite is present. Biotite is present in both samples as an accessory mineral. Porphyroblasts of garnet occur in sample 73844. Apatite, zircon and ilmenite are accessories.

It can be seen that there is no substantial difference in mineral assemblage between these rocks and the normal amphibolites (see p. 29 et seq.).

### Chemistry

The chemical analyses of samples 73843 and 73844 are reported in Table 15. These discordant amphibolites have the same chemical composition as the concordant amphibolites.

Table 15. *Chemical composition of discordant amphibolites.*

| GGU No.                                  | 73843 | 73844 |
|--|-------|-------|
| SiO <sub>2</sub> . . . . .               | 50.55 | 50.11 |
| TiO <sub>2</sub> . . . . .               | 1.00  | 1.25  |
| Al <sub>2</sub> O <sub>3</sub> . . . . . | 14.65 | 14.52 |
| Fe <sub>2</sub> O <sub>3</sub> . . . . . | 2.40  | 3.20  |
| FeO . . . . .                            | 9.01  | 8.69  |
| MnO . . . . .                            | 0.17  | 0.17  |
| CaO . . . . .                            | 9.96  | 9.92  |
| MgO . . . . .                            | 8.42  | 7.71  |
| Na <sub>2</sub> O . . . . .              | 1.80  | 2.34  |
| K <sub>2</sub> O . . . . .               | 0.15  | 0.21  |
| P <sub>2</sub> O <sub>5</sub> . . . . .  | 0.01  | 0.01  |
| H <sub>2</sub> O <sup>+</sup> . . . . .  | 1.18  | 1.21  |
| H <sub>2</sub> O <sup>-</sup> . . . . .  | 0.06  | 0.06  |
|  | 99.36 | 99.40 |
| <i>mg</i> . . . . .                      | 0.57  | 0.54  |
| <i>ti</i> . . . . .                      | 1.74  | 2.19  |

## PETROGENESIS

The petrography shows that the concordant layers in the migmatitic gneisses are the most useful indicators of the metamorphic history. These layers will therefore be discussed first and followed by a discussion of the migmatitic gneisses and the inclusions found in them.

### The Concordant Layers

#### Origin of the rocks

It has already been shown that the rusty schist is a metasedimentary rock, whose composition approaches that of sandstones and greywackes.

Field and petrographic observations do not help us to understand the origin of amphibolites, as has been demonstrated by several authors (e.g. LEAKE, 1964; RIVALENTI, 1966; RIVALENTI & SIGHINOLFI, 1969). The origin of their banding is also uncertain: if the banding of type i (p. 30) was formed by early metamorphic differentiation, then the banding of types ii and iii could either represent a pre-metamorphic feature, or be of metamorphic origin (RIVALENTI, 1971).

Many theories have been proposed to explain the origin of ultramafic rocks (e.g. SØRENSEN, 1953; VUAGNAT, 1954). The ultramafics of the present area are in most cases continuous single layers intercalated with the other lithotypes. There is a striking similarity between this pattern and that described by TURNER & VERHOOGEN (1960, p. 307) as "Peridotites and serpentinites of the 'alpine type' occurring in folded geosynclinal sediments of orogenic belts". This similarity seems to be more than accidental and in this case the layers as a whole could represent remnants of an old geosynclinal series. The amphibolites could therefore represent the volcanics (basalts, spilites, etc.) which are usually present in the geosynclinal sediments.

The microgranodiorite appears to have been derived by migmatisation from either the rusty schist or from amphibolite because its contacts are concordant with the foliation of the other rocks and its composition is similar to that of the gneisses. When large crystals of rutile are present they suggest a derivation from rusty schist which is the only other rock

in which rutile occurs; but when hornblende is present, it suggests a derivation from amphibolite.

### Metamorphic facies and petrogenesis

The best indicators of metamorphic petrogenesis are the mineral assemblages which occur in the rusty schist. These indicate that the rusty schist is a pelitic rock metamorphosed in the almandine-amphibolite facies of regional metamorphism. Several phenomena throw light on the metamorphic history of this rock.

i) Garnet sometimes contains inclusions of staurolite, which has never been found elsewhere. A similar phenomenon has been described by LAL & MOORHOUSE (1969) who consider it indicative of a progressive metamorphism starting from the staurolite-almandine subfacies. Staurolite may have disappeared through a reaction of the type:



(WINKLER, 1965, p. 90).

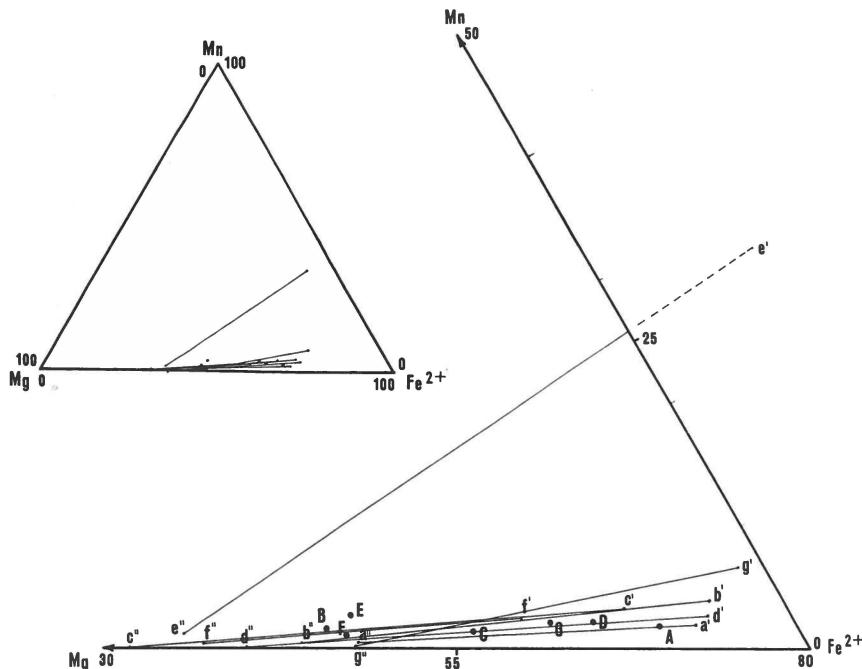


Fig. 15. Triangular Mn-Fe<sup>2+</sup>-Mg plot of co-existing garnets and biotites and their host rocks. A represents the whole rock sample A; a' the garnet in sample A and a'' the biotite in sample A. The samples A-G correspond with GGU sample A = 57738, B = 73543, C = 57820, D = 57856, E = 75880, F = 75884 and G = 57813. A and B are gneisses, C-F are rusty schists and G is microgranodiorite. See text p. 62 for discussion of the results.

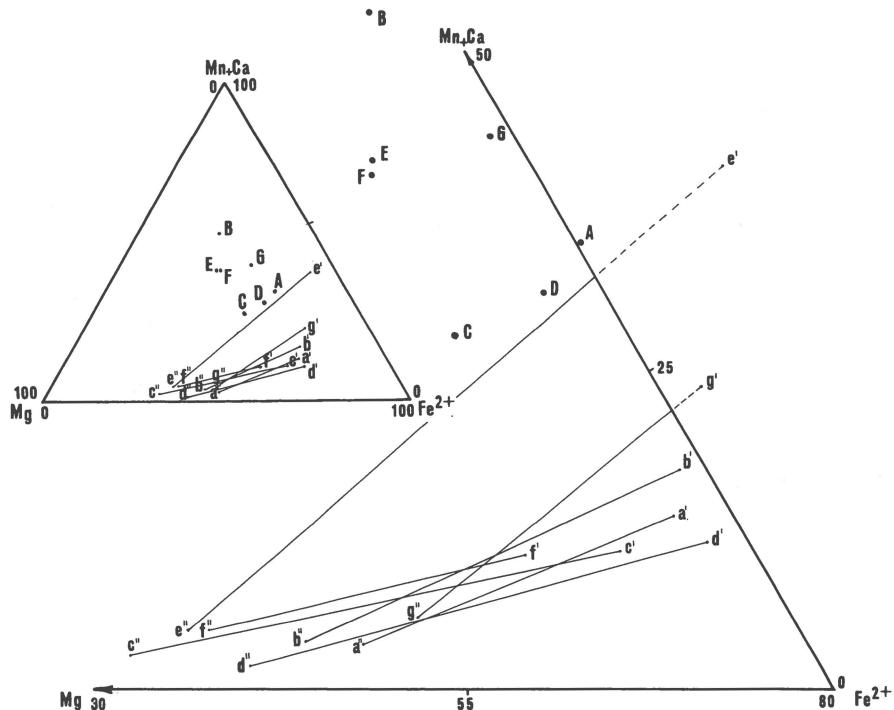
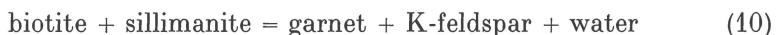


Fig. 16. Triangular  $(\text{Mn} + \text{Ca}) - \text{Fe}^{2+} - \text{Mg}$  plot of the same co-existing garnets and biotites and their host rocks as fig. 15. Explanation of the letters the same as that in fig. 15.

ii) The atoll structure of garnet indicates that it crystallised in two periods of metamorphism (RAST, 1965); this is also suggested by the relationship between garnet and other minerals, mainly to biotite. The older generation is pre- or syntectonic, while the later one exhibits post-tectonic features. In order to ascertain the relationship between garnet crystallisation and metamorphic grade, co-existing garnets and biotites (Table 10) have been examined following the method described by ATHERTON (1965). The tie-line pattern is shown in figs 15 and 16. Samples 57820, 57856 and 75884 (mainly syntectonic garnet) have parallel tie-lines, while that of sample 75880 (mainly post-tectonic garnet) has a completely different slope, showing that this garnet grew under different metamorphic conditions. Since the Mn content increases sharply in the garnet of sample 75880, it can be assumed that it crystallised under a lower grade of metamorphism than the garnets in other samples (MIYASHIRO, 1953; SIGHINOLFI, 1967).

iii) The occurrence of sillimanite inside garnet possibly indicates a reaction of the type:



occurring at the amphibolite-granulite facies border.

iv) According to RAMBERG, the transformation of biotite into anthophyllite and K-feldspar (reaction (6), p. 50) indicates conditions approaching granulite facies.

v) The association anthophyllite-cordierite. GRANT (1968) has suggested that anthophyllite-cordierite-bearing rocks formed from assemblages of the type feldspar-plagioclase-biotite-sillimanite-quartz by partial melting, filter pressing and recrystallisation, phenomena which suggest a high-grade environment. LAL & MOORHOUSE (1969) suggest that gedrite-cordierite assemblages may have originated by partial anatexis and removal of the melted granitic fraction. An anatectic process would clearly fit the rusty schist too, because a derivation from a normal biotite-sillimanite-garnetiferous schist is evident. Our assemblages differ from other known assemblages, however, in that they contain a fairly high amount of CaO, which entered the plagioclase and did not primarily form calcic amphibole. DE WAARD (1966) puts the anthophyllite-cordierite assemblage in the biotite-cordierite-almandine subfacies of the low hornblende-granulite facies.

vi) Rusty schists with a biotite-garnet-sillimanite assemblage have a chemical composition that corresponds only to that of rare aluminous greywackes (see p. 51). An anatectic process, with partial removal of a granitic melt (BROWN & FYFE, 1970) affords the best explanation of their composition.

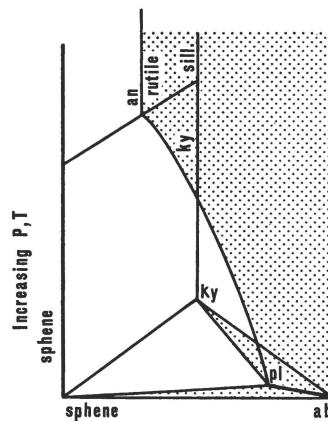


Fig. 17. Equilibrium diagram plagioclase-sphene-kyanite (from RAMBERG, 1952, p. 74).

vii) The widespread occurrence of large crystals of rutile, which together with ilmenite are the only titanium minerals, is also significant. Sphene is completely absent. In the several reactions leading to the formation of rutile reported by RAMBERG (1952, p. 73), the side of the equations where rutile is formed is considered to represent higher  $P-T$  conditions. It is

worth noting that rutile remains unchanged even when plagioclase is very calcic (Table 9). According to RAMBERG's diagram of equilibrium between sphene, plagioclase and kyanite (fig. 17) this again represents high  $P-T$  conditions. Similar significance is attached to rutile by DEER, HOWIE & ZUSSMAN (1962, vol. 5, p. 34). Several occurrences of rutile have been found in granulite facies rocks (BERTOLANI, 1968). The latter author has also discovered that rutile, once formed, is not easily re-transformed with decreasing metamorphism (personal communication).

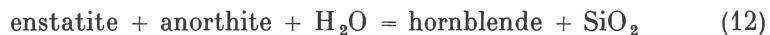
viii) The more calcic plagioclase is micro- (crypto-)antiperthitic. A similar phenomenon has been described in granulite facies rocks (CAPEPRI, 1968). Its replacement by more sodic plagioclase indicates a decrease in metamorphism.

ix) Sillimanite is retrogressively transformed into muscovite. Possibly this transformation takes place through a reaction of the type:



(RAMBERG, 1952, p. 151) which occurs at the boundary between granulite and amphibolite facies.

x) The transformation of anthophyllite and plagioclase into hornblende (reaction (7) p. 51) indicates instability due to the decreasing metamorphism. There is a similarity between reaction (7) and that reported by RAMBERG (1952, p. 54):



RAMBERG points out that the right-hand side of the equation is stable at lower temperatures and that enstatite can be substituted by anthophyllite, depending on the water pressure.

xi) The transformation biotite-chlorite and possibly biotite-muscovite, the pinitisation of cordierite and the formation of epidote indicate low-grade transformations.

All these phenomena show that the metamorphic evolution of the rusty schist was complex. A possible interpretation is given below.

i) Very early regional metamorphism took place in the staurolite-almandine subfacies of the almandine-amphibolite facies (recognised by the presence of staurolite enclosed by garnet).

ii) Increasing conditions of  $T$  and  $P$  gave rise to the formation of sillimanite and garnet. This phase must have reached the uppermost part of the almandine-amphibolite facies or the lowest part of the

granulite facies. During the culmination of this phase anatetic conditions were created and a granitic melt was mobilised. The overall effect of anatexis was weak, although at certain points (cordierite-anthophyllite assemblages) it was more pronounced owing to tectonic factors. This high-grade environment was characterised by high water pressure because anthophyllite occurs instead of enstatite and suggests that the rocks were metamorphosed in the lower part of the almandine-biotite-cordierite subfacies of the hornblende-granulite facies (DE WAARD 1966). Relics of high-grade assemblages are scattered throughout the rusty schist but are more commonly seen between the mountains 880 m and 760 m in the eastern part of the area.

iii) The peak of progressive metamorphism was followed by retrogression. The main evolution must have occurred within the almandine-amphibolite facies. On the whole, the mineral assemblages fit well with those found in the sillimanite-almandine-muscovite subfacies of Barrovian metamorphism (TURNER & VERHOOGEN, 1960, p. 548), but the lowest grade attained by these rocks extends to the lowest part of the almandine-amphibolite facies (staurolite-almandine subfacies again?). The widespread occurrence of large crystals of epidote associated with muscovite and chlorite indicates that in some cases even lower metamorphic grades were attained. Garnet probably recrystallised during this phase and so did plagioclase, becoming less calcic and more sodic. The equilibrium anthophyllite-anorthite was upset and consequently hornblende was formed.

Sometimes even sillimanite recrystallised in the form of fibrolitic aggregates.

iv) Some of the lowest grade transformations are connected with faulting and are only locally important.

The amphibolites also show signs of disequilibrium and polymetamorphism. Titanium minerals are as usual one of the most important indicators of metamorphic petrogenesis. Ilmenite, which in these iron-rich rocks has the same significance as rutile in the rusty schist, shows instability and partial transformation into sphene which clearly represents retrogression (see for instance reaction (4)). Reaction (3) has a similar meaning. The presence of microantiperthites probably indicates high-grade metamorphism (p. 35). The recrystallisation of normally zoned plagioclase indicates that the more calcic plagioclase became unstable. The widespread occurrence of large crystals of epidote indicates a retrograde transformation. Therefore, although it is difficult to assign an exact subfacies to these mafic assemblages, the amphibolites would

appear to have undergone a metamorphic evolution, from a high to a lower grade of the almandine–amphibolite facies, similar to that of the rusty schist.

The ultrabasic rocks also conform to this general picture. If the peridotite nucleus recrystallised from a former serpentinite (which is not certain, see p. 39) this should have occurred with decreasing  $P_{H_2O}$  and increasing metamorphism, which could correspond to the first relic phase of increasing metamorphism in the rusty schist.

The peridotite mineral assemblage of the nucleus would have been stable from the granulite facies to the almandine–amphibolite facies. But when chlorite occurs in the nucleus in equilibrium with olivine, orthopyroxene and amphibole, this assemblage is stable in a medium amphibolite facies (TROMMSDORF & EVANS, 1969).

On page 40 it has been shown that the amphibole of this assemblage is characterised by lower hexacoordinate Al and higher Mn and Fe than amphiboles occurring in peridotites in granulite facies rocks.

The transformation of the nucleus into serpentine and talc does not necessarily indicate greenschist facies conditions. BOWEN & TUTTLE (1949) and RAMBERG (1952) have pointed out that the appearance of these minerals is controlled by the partial  $P_{H_2O}$  of the environment: if  $P_{H_2O}$  is high, then talc and serpentine are stable up to 700°C and 500°C respectively. In our occurrences a high water pressure is indicated by the crystallisation of anthophyllite instead of enstatite (p. 64) in the rusty schist. The formation of talc and serpentine may thus have occurred in the almandine–amphibolite facies. As regards anthophyllite, HINRICHSEN (1967) has shown experimentally that Fe-rich anthophyllites are stable above  $520^\circ \pm 10^\circ$  at a constant  $P_{H_2O}$  of 1000 bar. In our case anthophyllite substitutes talc. This may be the result of a decrease in water pressure occurring in the almandine–amphibolite facies. The petrogenesis of the ultramafic rocks is complicated by metasomatic exchanges with the country rock and redistribution of elements within the body (see p. 43). A metasomatic mechanism has probably produced the rim of hornblendite. The ultramafic rocks also underwent low-grade transformations such as a weak chloritisation of hornblende and the partial transformation of anthophyllite into talc.

The microgranodioritic rock does not present any interesting metamorphic features. The presence of large crystals of rutile probably means that this rock has undergone high-pressure metamorphism. Its mineral assemblages, when not downgraded to the epidote–amphibolite or greenschist facies by local faulting, conform with those of the almandine–amphibolite facies. A garnet–biotite tie-line for a microgranodiorite (sample 57813) is reported in figs 35 and 36. It intersects

the high-grade tie-lines of the rusty schist couples and therefore the garnet in the microgranodiorite might have crystallised in a medium to low almandine–amphibolite facies, but the presence of some hornblende makes the interpretation doubtful.

## Inclusions in the Gneisses

### Metamorphic facies and petrogenesis

All the inclusions possess mineral assemblages which are stable in the almandine–amphibolite facies. Assemblages indicating an upper almandine–amphibolite facies are found in the calc-silicate inclusions. Some transformations (for instance, the biotitisation of hornblende or that shown by reaction (8)), have clearly occurred by decreasing the metamorphic grade.

It has been shown that most of the inclusions correspond to lithologies found in the major layers and are formed through boudinage or agmatisation with or without chemical exchanges with the migmatites. A possible explanation of the hornblendite inclusions which cannot correspond to the hornblendites found in the layers (p. 56) is that they could have originated through basification, either by the subtraction of salic elements or by the introduction and fixation of mafic elements (RAMBERG, 1952, p. 262). As RAMBERG points out, basic elements may become fixed in remnant inclusions in zones of general quartz-feldspathisation such as that under consideration.

Many theories have been proposed for the origin of the calc-silicate lenses. In another part of the Frederikshåb district, WALTON (1966) considers that they developed from pillow lavas. This theory has been criticised by BRØGGER SØRENSEN (1968) who proposed a derivation from original Ca-rich rocks. HENTSCHEL (1943), BACKLUND (1953) and MEHNERT (1968, p. 304) consider similar skarns to be a product of metamorphic–metasomatic transformation of limestones and marbles, while SHAW *et al.* (1962) describe calc-silicate skarns which they interpret as altered amphibolites and metagabbros.

Although it is possible that the skarns described in this paper are derived from marbles, it is considered probable that they formed by the metamorphism of amphibolites for the following reasons: 1) they only occur in connection with agmatites; 2) epidote–pyroxene–hornblende-bearing layers (p. 30) with the same mineral composition as the inclusions occur in major amphibolites; 3) the chemical composition of the inclusions is similar to that of amphibolites (compare the analysis of sample 57769 (Table 13) with those of Table 4).

## Discordant Amphibolites

The discordant amphibolites possess almandine–amphibolite facies assemblages which are similar to those of other rocks and therefore they probably represent mafic dykes which were emplaced early in the history of the region.

## Migmatitic Gneisses and Pegmatites

### Metamorphic facies

The gneisses possess almandine–amphibolite facies assemblages but they lack critical minerals, such as Al-silicates, which would enable the subfacies to be established exactly. However, indications are found also in the gneisses of retrogression from the sillimanite–almandine muscovite subfacies to the almandine–amphibolite facies. This retrogression is shown by the following transformations:

- i) Biotite is often recrystallised as a green type, with release of titanium in the form of sogenitic rutile. It is well known that, as metamorphic grade decreases, the titanium content in biotite also decreases (ENGEL & ENGEL, 1960; AOKI, 1961; SIGHINOLFI, 1968);
- ii) The transformation of hornblende into biotite is accompanied by the formation of epidote and sphene. MEHNERT (1968, pp. 52 and 262) observed that the titanium content of biotite is higher than that of hornblende and suggested the reaction:



This equation is correct if the transformation takes place at a high metamorphic grade. But if, as in the present case, potash is added with decreasing metamorphism, it may happen that biotite takes up less titanium than is present in hornblende, because the amount of titanium in biotite decreases with decreasing metamorphism.

- iii) Muscovite and epidote formed by the reaction (1).

Two co-existing garnets and biotites from gneisses with the same composition as that of the rusty schist have been analysed (Table 2) to obtain information on the metamorphic grade under which garnet crystallised. The tie-lines are reported in figs 15 and 16.

Fig. 16 shows that both the gneiss tie-lines intersect those of rusty schists whose garnet was formed during high-grade metamorphic conditions, but they do not cross that of sample 75880, whose garnet was formed under lower metamorphic conditions (p. 62). In fig. 15 one

gneiss conforms to this pattern, but the tie-line of the other is subparallel to those of rusty schists with high-grade garnet. The difference in behaviour may reflect the influence of host rock composition upon the garnet. However, according to ATHERTON (1965), Ca in garnets plays a role similar to that of Mn during retrogressive metamorphism. Although unambiguous conclusions cannot be based on the analyses of only two couples, it seems that garnet in the migmatitic gneisses crystallised in low almandine–amphibolite facies.

All the mineral transformations and assemblages considered so far are consistent and seem to have occurred during retrogression from high to low almandine–amphibolite facies. Further transformations into epidote–amphibolite and greenschist facies assemblages are sometimes found. These are indicated by the chloritisation of mafics and the transformation of plagioclase into albite and epidote.

The facies of the pegmatites generally reflects that of the country gneisses as is shown by the similarity of their mineral composition and reactions (see p. 23). Only the quartz and epidote–bearing mobilisates ( $P_4$ ) can represent a low grade hydrothermal mobilisation.

### Petrogenesis

Interpretation of the origin and evolution of polymigmatitic gneisses and related pegmatites is a difficult subject (MEHNERT, 1968) which requires much more detailed work than it was possible to carry out in the present area. However, a few comments can be made here. There are three main possible explanations of the origin of the gneisses: a) they represent the product of an isochemical metamorphism of pre-existing rocks (an explanation proposed, for instance, by KALSBEK (1970) for gneisses exposed 50 km south of Frederikshåb); b) they were formed by anatexis; or c) they were formed by metasomatism.

The phenomena which we have observed which are relevant to a discussion of these three hypotheses are summarised below.

i) Layers (the concordant layers, p. 29) of a pre-migmatitic series are scattered throughout the migmatitic gneisses. We therefore consider that the gneisses were chiefly derived by migmatisation from the lithotypes of the layers because most of the relic inclusions in the gneisses correspond to lithotypes found in the layers.

ii) Amphibolite boudins in the gneisses are often enclosed within a sheath of hornblende gneiss.

iii) during retrogressive metamorphism microcline crystalloblasts developed by replacement in the gneisses; hornblende was replaced by biotite (p. 18), and epidote and muscovite formed from plagioclase (p. 17).

iv) The composition of the older pegmatites ( $P_1$  and  $P_2$ ) is strictly related to that of the host rock, while the younger  $P_3$  are richer in K-feldspar, irrespective of the country-rock composition.

v) The pegmatites appear to have been formed by several mechanisms. The concordant pegmatites and veins sometimes have dark rims, and are most likely to be the product of metamorphic differentiation. When the composition of discordant pegmatites is the same as that of the host rock, and the host-rock structures continue as shadows in the pegmatite, they are considered to be replacement bodies. Pinch-and-swell structure and zoning were considered by RAMBERG (1956) to be proof that pegmatite had grown by metasomatism in a solid state, while GRESENS (1967a & b) believes that zoned pegmatites formed by the migration of hydrothermal intergranular fluids under tectonic influence. Dilatational intrusive pegmatites have also been found: for instance the  $P_3$  K-feldspar-rich pegmatites which cross-cut the amphibolites.

vi) Feldspar porphyroblasts occur which have grown by forceful shouldering aside of biotite (p. 15).

No single theory can explain all these phenomena. The hypothesis of isochemical crystallisation, however, can be completely rejected because the hornblende gneiss has no counterpart in the pre-migmatitic series and it is statistically unlikely that, if such a lithotype existed, it would have been completely migmatised in all cases.

Both intracrystalline and intercrystalline metasomatism (MEHNERT, 1968, p. 285) has been experimentally demonstrated to be effective only on a small scale (BLACKBURN, 1968). Nevertheless, phenomena generally accepted as metasomatic occur in the present migmatites.

As far as anatexis is concerned the composition of an anatetic mobilisate should be granitic according to the experimental works of several authors (see e.g. BROWN & FYFE, 1970), while our gneisses have a dioritic composition. An evolution towards a dioritic composition has only been experimentally obtained in unnatural conditions (for instance with very high  $T$  and  $P_{H_2O}$ , JAMES & HAMILTON, 1969). However, anatexis cannot be completely excluded for the following reasons: 1) laboratory conditions do not correspond to natural ones; 2) many anatetic migmatites have leucosomes of non-granitic composition (see for example MEHNERT, 1962); 3) if anatetic phenomena are repeated after the formation of a primary granitic melt the following melts must trend towards a more dioritic composition because the primary granitic melt might migrate (see BROWN & FYFE, 1970; SIGHINOLFI, 1971); 4) the melting

of biotite by anatexis depends on the water pressure and thus, if water pressure is sufficiently high and other K-bearing minerals are absent, then the composition of the melt must tend towards diorite.

To conclude, neither anatexis nor metasomatism can be excluded in the present area and a mechanism which might be proposed is a combined activity of both: i.e. an anatetic mobilisate metasomatically reacting with basic rocks.

The sequence of events which led to the formation on the migmatites is considered to be:

1. High-grade (possibly low granulite facies) metamorphism of a series of rocks similar to those which now form the concordant layers.
2. The rocks were deformed and reacted in different ways depending on their composition; the metasediments behaved plastically whilst the amphibolites were boudinaged.
3. Partial anatexis occurred with migration of a granitic melt. This phenomenon is suggested by the anthophyllite-cordierite assemblage in the rusty schist (pp. 63 and 65). Migmatisation may have begun here, with the formation of concordant leucosomes ( $P_1$ ) and the reaction of the mobilisates with the amphibolites to give hornblende gneisses. It continued with increase in  $P_{H_2O}$ , in high almandine-amphibolite facies where it reached its maximum. The mobilisates were always granodioritic to quartz dioritic (quartz dioritisation) and there was a tendency towards general homogenisation, because the amphibolites were transformed by reaction with material coming from felsic gneisses. The process reached completion (probably under the influence of tectonic factors) at certain points where homogeneous gneisses were found (diatexesites of MEHNERT, 1968, p. 253).
4. Retrogressive metamorphism took place in low almandine-amphibolite facies and the last stages of migmatisation occurred forming microcline crystalloblast, microcline-rich pegmatites and the biotitisation of hornblendes. This last phase might have followed immediately after quartz dioritisation, but the possibility of a gap cannot be excluded.

### Conclusions

The gneisses of the area were formed from a geosynclinal series, which first underwent prograde metamorphism from the staurolite-almandine subfacies to the low hornblende granulite facies (almandine-biotite-cordierite subfacies, p. 64). At the highest stage of metamorphism partial anatexis occurred (p. 65) and the period of migmatisation began. Subsequently, there was a period of retrograde metamorphism from the upper (sillimanite-almandine-muscovite subfacies) to the lower al-

mandine–amphibolite facies. The effects of migmatisation were most marked during the period of decreasing metamorphism. The general “quartz (grano-) dioritisation” can be ascribed to the high almandine–amphibolite facies, and microcline blastesis to the medium and low part of the same facies. Epidote–amphibolite facies and greenschist facies metamorphism were late phenomena locally associated with brittle deformation. Mafic dykes (now discordant amphibolites) were intruded at some stage before the almandine–amphibolite facies metamorphism.

## POST-OROGENIC BASIC DYKES

### Introduction

The basic dykes are the subject of a paper by RIVALENTI & SIGHINOLFI, (1971) and therefore they will only be briefly described here. The whole area is intersected by basic dykes whose thicknesses vary from a few centimetres up to 300 metres. All the dykes weather to a reddish brown colour and, with the exception of the TDs which always lie in erosional hollows, they stand out in relief with respect to the gneisses. The most widespread dykes are the first dolerite generations which are typically covered by several metres of an alteration gravel. The following generations of dykes have been established in the southern part of the Frederikshåb district (JENSEN, 1966).

MD<sub>1</sub>—the oldest, with a general strike north-south.

MD<sub>2</sub>—general strike north-east.

MD<sub>3</sub>—general strike south-east or east-south-east.

BD (brown dolerites)—general strike east-south-east or north-east.

TD (trap dolerites)—the youngest, with a strike parallel to the coast (i.e. south-south-east).

Lamprophyres—irregular strike.

The first three generations are thought (JENSEN, 1966) to be pre-Ketilidian (> 2500 m.y.) since, farther south, in the Ivigtut area, they are deformed by Ketilidian structures. BDs are considered to be of Gardar age (c. 1250 m.y.), while TDs are much younger with ages of 138 and 164 m.y. (LARSEN & MØLLER, 1968; WATT, 1969) indicating a Mesozoic age. The age of the lamprophyres in our area is not well defined. This dyke terminology is used in the present description because all these generations of dykes have been found, except MD<sub>1</sub> and BD. No Gardar dyke has been found west of the present area (CHADWICK, 1969).

### Description and petrography

#### MD<sub>2</sub> dykes

The thickest dykes of the area belong to this generation. The maximum thickness observed is 300 m (in the dyke crossing the area from Aorngo to the inland ice and forming the 670 m and 780 m hill-tops).

Locally two sub-generations which cross each other have been distinguished. The older and younger sub-generations are shown on Plate 10 as  $MD_2(1)$  and  $MD_2(2)$  respectively. All the  $MD_2$ s have chilled contacts with the country rock. The general strike of this swarm is north-east, but local swings to north-north-east or east-north-east often occur. 'En échelon' and 'en bajonette' arrangements have been found sporadically. A peculiar feature of this 'en échelon' structure is that dykes suddenly stop, swarming out in a large number of minor dykes, and begin again some hundred metres to the side. It is apparent from the map that this also happens to the thickest dykes. They are sheared by the south-east and north-south set of faults and fractures but not usually by the north-east ones, although some laminations occur along the strike. Veins of epidote and other hydrothermal veins occur along the south-east fractures; these are described later together with fractures. The dykes of this generation, when large enough, exhibit a zonal structure. From the margins, where the rock has an aphanitic glassy texture, the grain size increases toward the centre. Another feature, which is also present in the  $MD_3$  dykes, is that they become porphyritic along the strike in a discontinuous way. For instance, the thin dykes which extend from the south-western part of Igassortoq and pass a few hundred metres north of mountain 880 m have large plagioclase phenocrysts in only a small part of their total length, while in the remainder they have a normal coarse-grained texture without phenocrysts for no apparent cause. For this reason no porphyritic dyke has been mapped as such.

The mineral texture of  $MD_2$  dykes is characterised by the ophitic intergrowth of plagioclase and clinopyroxene. Plagioclase is zoned. Clinopyroxene is either colourless or pale brown. Some minor amounts of olivine have been found. A mineral zoning occurs in the thicker dykes; olivine may be present towards the contacts together with clinopyroxene and plagioclase but is absent in the centre. This can be explained by the different rate of cooling of the contact and the centre, producing a weak differentiation. Minor amounts of brown biotite are also present. The primary minerals are partially uralitised and consist of aggregates of chlorite, talc, sericite with accessory opaque minerals and apatite.

#### *$MD_3$ dykes*

$MD_3$  dykes are generally thinner and much less weathered than  $MD_2$ s. Their thickness is usually limited to 10–20 m and in only one case, the  $MD_3$  crossing the peninsula Igassortoq and forming the top of mountain 560 m, does it attain a thickness of 80 m. They generally trend south-east, but swings to east-west or even north-east are frequent. The contacts are always chilled. As in the  $MD_2$  dykes, the grain size increases from the contacts to centre, a feature limited to the thickest dykes. The

$MD_3$ s also possess the peculiar 'en échelon' structure, described for the  $MD_2$  dykes although less commonly.

$MD_3$  dykes are sheared by north-south faulting. Sometimes, as in the case of the thin dyke which crosses the whole area from Nunaqarfia they are sheared by south-east fractures. The mineral composition and texture of these dykes is similar to that of the  $MD_2$ s.

#### *TD dykes*

Only three dykes of this generation have been found in this area. The first dyke, one metre thick, was found east of Nunaqarfia cutting both an  $MD_2$  and an  $MD_3$ . The second is a long, thin dyke crossing the western part of the area from the complex of lakes north-west of mountain 880 m to Qingua. The third is a dyke 10–20 metres thick crossing the rocks of the eastern part of the present area. They have a general south-south-east strike, but swings towards a north-south direction occur. Their contacts are always chilled. Field examination in other areas suggests that TDs are never affected by shearing but in this area, the TD dyke of the eastern peninsula in the inland ice is strongly sheared along the strike east of mountain 920 m. The sheared dyke looks completely transformed and albitisation phenomena occur. This indicates a relatively recent rejuvenation of the north-south fractures. The mineralogy of the TDs differs a little from that of the preceding generations in that biotite is present in much larger amounts. Clinopyroxene is a pale brownish type which is commonly zoned and twinned.

Primary hornblende may be present, as well as some olivine. Plagioclase is often zoned and presents a subophitic texture with pyroxene. Alteration of these minerals is common. The TD dykes are characterised by the presence of vesicles of up to two centimetres in diameter. They are filled with fibrous zeolites which, according to X-ray determinations, belong to the natrolite group.

#### *Lamprophyres*

Only two lamprophyres have been found: one is a very thin dyke (four metres thick) which is traceable only for a few hundred metres in the western part of the area, near mountain 880 m, and the other, a little thicker (5–10 m), near mountain 560 m in the peninsula north of Sermilik avangnardleq. Their age relationships are not well defined; the first was not found to cut any other dyke, while the second was found to intersect an  $MD_3$  but not a TD. The first is an irregular east-west dyke while the second has a definite north-east strike.

The lamprophyre near mountain 880 m is a dark green spessartite presenting grain-size inhomogeneities: coarse-grained roundish pebbles are embedded in a finer grained matrix which nevertheless has the same composition. The texture is porphyritic. The mineral assemblage consists of a large predominance of barkevikitic

zoned amphibole and fairly acid plagioclase. The centre of the amphibole phenocrysts is pleochroic with  $\alpha$  = pinkish yellow,  $\beta$  = reddish brown,  $\gamma$  = dark brown, while at the periphery, the colours become darker and greener. Transformations into bluish amphibole and chlorite are common. Small amounts of augite-like, pink clinopyroxene are present. Accessory minerals are represented by sphene, apatite and sogenitic rutile which forms a network in the bluish amphibole and chlorite.

The lamprophyre near mountain 560 m is mineralogically different from that described above. It is thin but it exhibits clear zoning in both grain size and mineralogical composition. Near the chilled contacts it consists of an ultramafic assemblage dominated by secondary serpentine after olivine, and pinkish clinopyroxene with minor amounts of biotite and plagioclase. Half-way between the contacts and the centre olivine decreases in abundance though some crystals are still present. The clinopyroxene content remains more or less unchanged, while plagioclase increases. At the centre a fairly high amount of hornblende is associated with clinopyroxene, and biotite, as well as plagioclase, increases slightly. Opaque minerals are common throughout the dyke. Uralitisation of pyroxene occurs. The mineral assemblage just described is found a little north of mountain 560 m, near the intersection with the MD<sub>3</sub>. Southwards this dyke disappears under drift in a deep valley and a little farther on it is replaced by a north-south striking green dyke, in which chlorite and feldspar predominate.

# STRUCTURE

## Introduction

In spite of the structural complexity of the area, the presence of lichen-free surfaces, polished by glacier activity, and the continuity of the outcrops have enabled us to distinguish several structural events on a mesoscopic scale. The interpretation of the mesoscopic structures and the presence of more or less continuous marker horizons, given by the lithotypes of the concordant layers, provide a key to the study of the large scale geometry. The events summarised below have been distinguished by means of their mutual relationships.

Ductile structures:

$S_0$ —original lithological layering

$F_1$ —rootless intrafolial folds

$F_2$ —isoclinal or tight meso- and macroscopic folds

$S_1$ —original foliation

$F_3$ —plastic syn-migmatitic folding

$S_2$ —foliation associated with  $F_3$

$F_4$ —late migmatitic folding

$F_5$ —post-migmatitic folding

Brittle structures:

NE fractures

SE fractures

N-S to SSE fractures.

## Ductile Structures

### Mesoscopic scale

$S_0$  denotes the lithological layering which pre-dates any other structure. Normally it coincides with  $S_1$ , because of the isoclinal character of the folds affecting  $S_0$ . The discordance is evident in the closures of  $F_1$  and  $F_2$  and sometimes on the limbs of  $F_1$  folds (figs 18 and 19). This old layering is assumed to be of pre-metamorphic origin even though in some cases it may be confused with S-surfaces created by an early metamorphic differentiation. The lithological layering seems in some cases to be accentuated by an old metamorphic differentiation. In the concordant

layers and in the inclusions in the gneiss (fig. 18) thin felsic layers occur as  $S_0$ . They are often surrounded by thin hornblendite rims (fig. 18), which are cut by the  $S_1$  foliation. These felsic layers and mafic rims are considered to be the result of an early metamorphic differentiation.

$F_1$ . The earliest recognisable deformation is represented by scattered relics of intrafolial folds which occur in amphibolite boudins within the gneisses. Their style varies from isoclinal to tight. The limbs may be parallel to or truncated by the gneiss foliation (fig. 18). Their axial surface is parallel to the gneiss foliation and is often marked by axial plane structures. Their fold axis has a variable orientation depending on the orientation of the gneisses containing the folded schlieren. It was not possible to determine whether all these folds belonged to a single pre-migmatitic deformation or whether they represent several deformations. The variations in their style might be thought to favour the second hypothesis.

$F_2$ . Most of the folds of this phase have been observed in the amphibolite layers on both a mesoscopic and macroscopic scale. Their style is normally isoclinal; very occasionally it is tight. The axial plane of  $F_2$  folds is sometimes marked by foliation but never by migmatisation and is always parallel to the gneiss foliation (see for instance fig. 19). Normally the fold closures are approximately semicircular, occasionally they assume an angular shape. When the closures are semicircular the gneiss foliation changes its direction locally but is not conformable with the amphibolite and thus there is locally a discordance between  $S_1$  of the gneiss and  $S_0$  of the amphibolite. Not uncommonly an amoeboidal segregation pegmatite occurs in the gneisses at the point of closure, as this point was a low pressure area. When the closure is angular, the amphibolite dies out by thinning in the gneisses and no evident discordance can be seen. The trend of the fold axis is quite variable because of the later refolding. Parasitic isoclinal folds have been found on the limbs of the major  $F_2$  folds.

$S_1$  consists of a planar orientation of minerals which gives a foliation to the rocks. It can be seen within a single bed; in banded or veined rocks it is parallel with the banding or veining (and to the  $P_1$  pegmatites, p. 27). In homogeneous gneisses this foliation is formed by the planar orientation of micas and hornblende crystals. The  $S_1$  foliation is always parallel to the axial plane of  $F_2$  (and  $F_1$ ) folds. When a compositional banding is seen in the rocks of the concordant layers, the gneiss foliation is parallel with this banding, except for closures of  $F_2$  (and for closures and sometimes limbs of  $F_1$ ). As  $F_2$  folds are usually isoclinal,  $S_1$  and  $S_0$  merge on the limbs of these folds and only in the hinges does a discordance

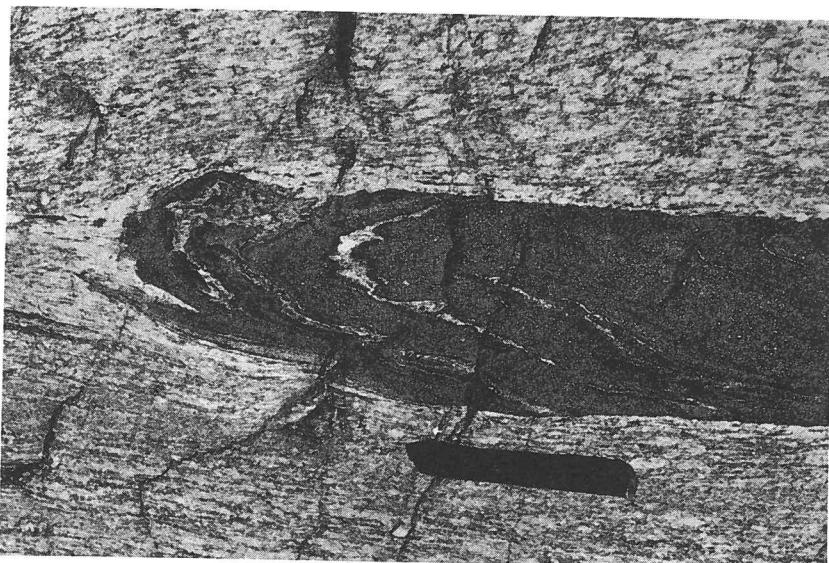


Fig. 18. Rootless intrafolial fold. Note that the felsic layers in the inclusion are surrounded by rims of hornblendite which define an  $S_0$  surface cut by  $S_1$  of the gneisses.  
2000 m north-west of mountain 880 m.



Fig. 19. Isoclinal  $F_2$  fold of amphibolite. Axial plane parallel to  $S_1$ . 1500 m east-south-east of mountain 760 m.

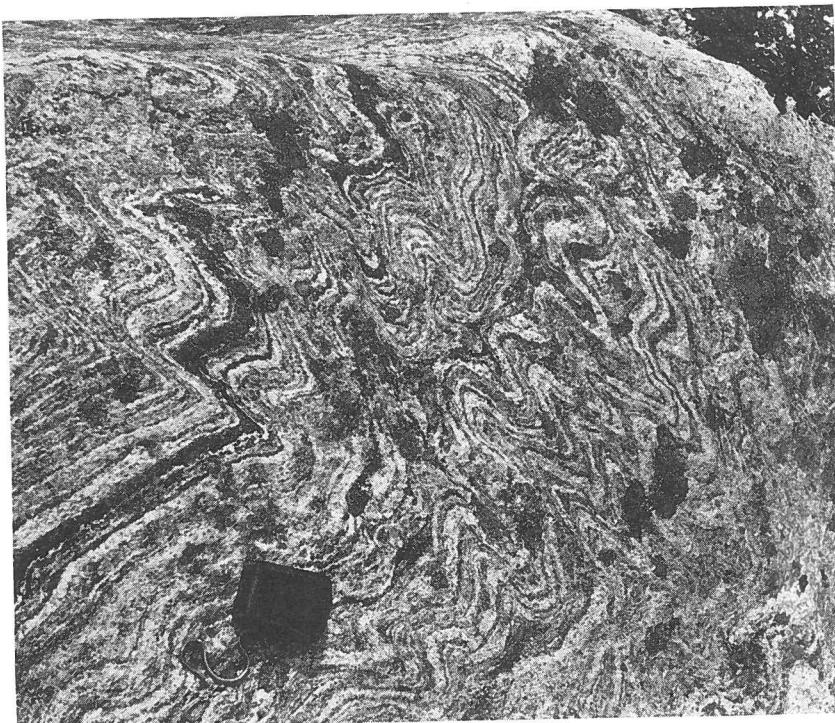


Fig. 20. Plastic  $F_3$  folding in the gneisses. The axial plane is sometimes curviplanar. At some points limbs are sheared, defining an  $S_2$ . 800 m north-north-east of mountain 880 m.

make the distinction clear. All these features indicate that the  $S_1$  foliation was probably formed by transposition during the  $F_2$  folding. Therefore the  $P_1$  pegmatites, which are always parallel to  $S_1$  and the boudins elongated in the sense of  $S_1$ , were also formed during the  $F_2$  deformation.

**$F_3$  and  $S_2$ .** The majority of  $F_3$  folds occur in the small-folded, veined and banded gneisses. They are the commonest mesoscopic folds in the area, and large scale structures are less frequent. Their style is highly plastic (fig. 20). They range from tight, almost isoclinal, to open folds and usually have a similar shape with strongly thickened hinges and slender to sheared limbs. Owing to their peculiar characteristics they fit the description of 'flow' folds (TURNER & WEISS, 1963, p. 481). When layers of different competency were folded together, disharmonic folds were produced. The axial surface is often curviplanar even when not refolded by younger folds, and gives rise locally to conjugate folds. Adjacent curviplanar axial surfaces are not necessarily parallel. The fold axis is marked by lineations which consist of: 1) the orientation of hornblende

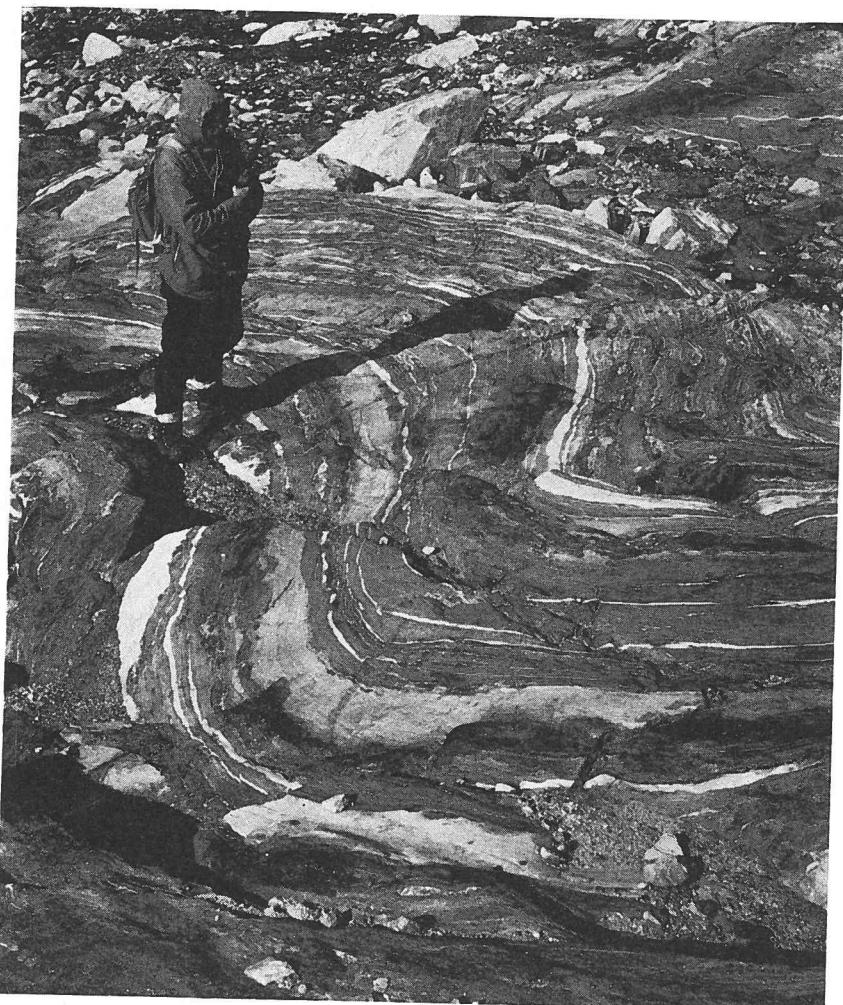


Fig. 21.  $F_4$  open fold at the contact between amphibolites and gneisses near lake 650 m.

crystals; 2) the growth of quartz-feldspar porphyroblasts; 3) micro-crenulations of the hinges of parasitic folds. An axial plane foliation, formed by strain-slip cleavage, sometimes occurs. The axial plane structure and the shearing of the limbs may locally create an  $S_2$  foliation which does not seem to be present on a large scale; it occurs on a large scale only in the area 3200 m west-south-west of mountain 880 m in the western part of the area. The orientation of the  $F_3$  folds is variable and is related to the interference with other folds, but a trend of the fold axis toward NE is statistically the most frequent. The axial planes of  $F_3$  folds are associated with a migmatitic neosome which also occurs as thin intrusions, conformable or subconformable with the foliation of the folded

gneisses. The  $P_2$  pegmatites described on p. 27 have partially suffered  $F_3$  deformation (fig. 10) indicating that the formation of these pegmatites is syntectonic with respect to  $F_3$ . These various considerations, i.e. the highly plastic style of  $F_3$  folds and their relationships to the migmatitic neosome, suggest that the  $F_3$  deformation was broadly syn-migmatitic.

$F_4$ . When unaffected by successive foldings, the folds of this phase have a general orientation NW to NNW or SE to SSE.  $F_4$  folds are found both on a mesoscopic and macroscopic scale. They are normally open to tight folds with a similar style (fig. 21).  $P_3$  pegmatites often occur parallel to axial planes of these folds. Lineations consisting of elongated porphyroblasts of feldspar and quartz emphasise the fold axis direction. A strong lineation of the amphibole described on p. 51 occurs parallel to the fold axis in an  $F_4$  mesoscopic fold, occurring in the rusty schist between lakes 650 m and 745 m in the southern part of the area. This indicates that  $F_4$  too took place during a main phase of metamorphism. The more regular shape of  $F_4$  folds relative to  $F_3$  indicates that the environmental conditions had a lower degree of plasticity.

$F_5$  folds are open and their profiles approach a concentric form. The fold axis is variable both in trend and in plunge. The trend changes from E-W in the western part of the area to NE or NNE in the eastern and northern parts. The axial plane, marked by a fan-shaped fracturing, is sub-vertical and has an average NE-SW strike. This folding post-dates migmatisation. Its general style and its axial-plane fracturing, indicate that it occurred in relatively brittle conditions with respect to the preceding deformations.

### Mesoscopic interference pattern

Some of the commonest interference patterns which have enabled us to recognise the various phases of folding are reported below.

When the axial surfaces of two successive generations are at a high angle to each other and so are the fold axes and the axial planes are roughly perpendicular to the plane containing the fold axes, then a succession of domes and basins is formed (RAMSAY's type 1 interference pattern, 1967, p. 521). This type has been seen in interferences between  $F_5$  and  $F_4$ .

Other simple interference patterns result from the refolding of  $F_2$  by subsequent phases and sometimes from the interference of  $F_4$  with  $F_3$  (Figs 22, 23). This type of structure occurs when the fold axes of the interfering sets are roughly parallel and the axial planes are not (RAMSAY's type 3 interference pattern, 1967, p. 530). When the axial planes and fold plunge directions are at a moderate angle to each other (RAMSAY's type 2 interference pattern, 1967, p. 525) the resulting inter-

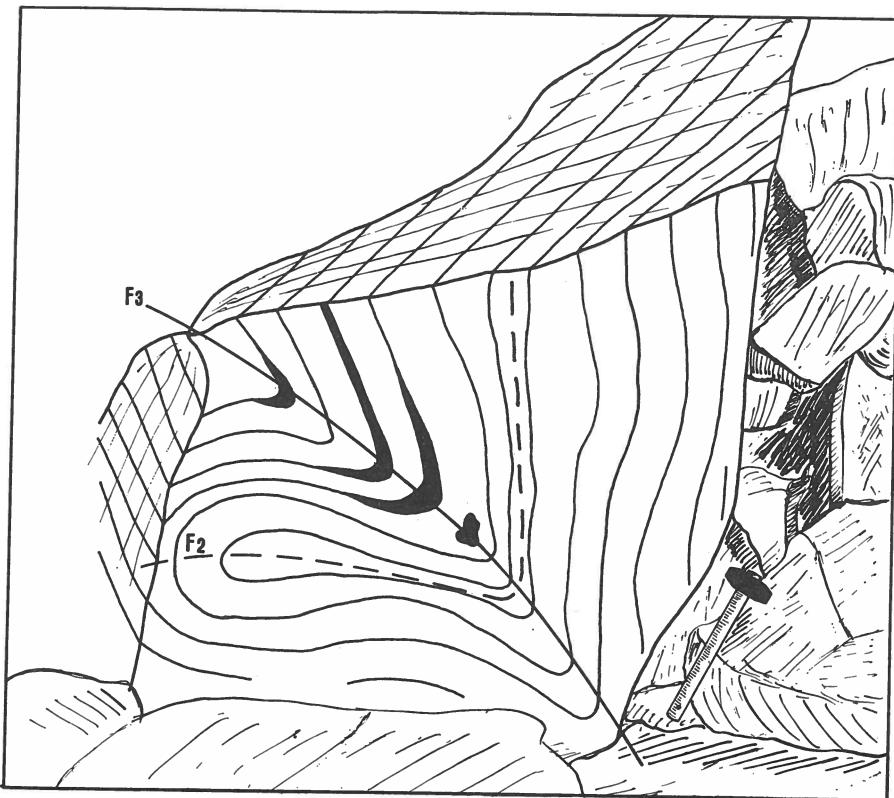


Fig. 22.  $F_2$  refolded by  $F_3$  near the contact between amphibolite and gneisses west of lake 530 m. (Drawn from a photograph).

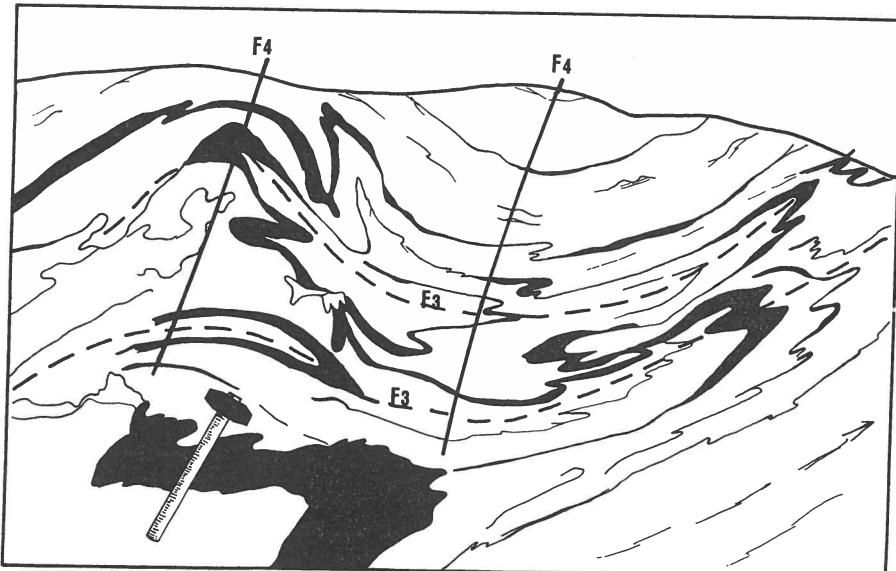


Fig. 23.  $F_3$  refolded by  $F_4$ . 100 m north-east of mountain 933 m. (Drawn from a photograph).

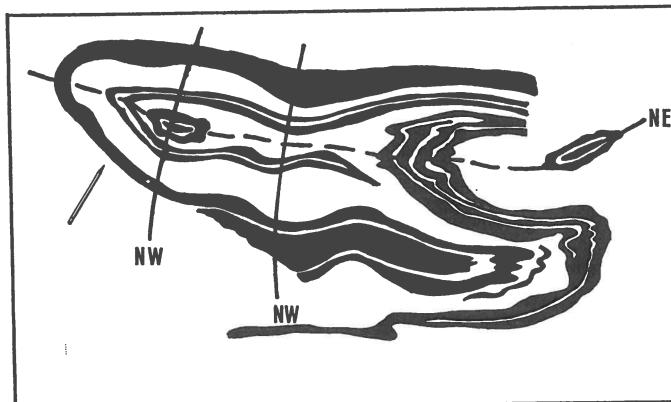


Fig. 24. Interference pattern among three folds. The oldest is an isoclinal folding ( $F_2$ ), refolded by  $F_3$ , refolded in turn by  $F_4$ . At the contact between amphibolite and gneisses near lake 650 m. (Drawn from a photograph).

ference patterns are much more complicated and the two-dimensional erosion surfaces display mushroom, crescent- or prong-like figures, which have been found in interferences between  $F_5$  and  $F_4$  or  $F_3$ . The trend of the older hinge can be approximately determined by joining the ends of the prongs (HOLMES & REYNOLDS, 1954). Fig. 24 shows a complicated interference between three folds. The oldest is isoclinal and is clearly refolded in a highly plastic way about a roughly NE axis. It was subsequently refolded by an open fold with a NW axial plane trace ( $F_2-F_3-F_4$ ).

#### Macroscopic scale

The macroscopic structures are as complicated as the mesoscopic ones. The presence of marker horizons enables one to gain an insight into the general structure. When marker horizons are absent the study is much more problematical although measurements of the systematic changes of the trends of the gneiss foliation help one to understand something of the geometry of the outcrops. The whole area has been divided into 32 sub-areas (plate 8). Normals to foliation planes, minor fold axes and constructed  $\beta$  axes are plotted on an equal area projection, lower hemisphere, from each sub-area (plate 9).

In the following description the area will be divided into an eastern and a western field; this is for simplicity only, because this division does not in fact correspond to any structural boundary, some of the structures of the eastern part continuing in the western part.

#### Western field

The western field consists of the area limited to the west by the fjord Qagssip kangerdluarssua and to the east by the valley north of

Qíngua. Sub-areas and plots from 1 to 15 (plates 8 & 9) concern this field. In some sub-areas the foliation planes do not form a single great circle, but are spread out. In these cases two or even three great circles have been tentatively traced, but they are of doubtful significance. The constructed  $\beta$ -axis is roughly parallel to the minor fold axes and lineations which have been measured in the field. When they are not parallel, it means either that the  $\beta$  axis is not a real fold axis or, as is commonly the case, that not all the measured axes belong to the same fold generation.

The main structures of the area consist of a succession of antiforms and synforms with almost horizontal fold axis (sometimes plunging towards NE in the north-eastern part of the area. Normally this folding is approximately upright but it is sometimes steeply inclined to the NE. Only the synform passing near mountain 880 m seems to be overturned near lake 620 m (and its axial plane dips to SE), but this cannot be definitely proved. The axial plane is marked by fractures. This fracturing and their general style suggests a probable  $F_5$  age for these folds. The lack of an axial plane migmatisation, as can be seen for instance in the amphibolite synformally folded near lake 670 m, also conforms to the assumption that they are  $F_5$  and not  $F_3$  folds. Apart from the presence of large scale  $F_4$  folds, which occur in the western part of the area and in the garnetiferous gneiss of Nigerdlip qôrорssua (sub-areas and plots from 1 to 7, plates 8 & 9), an important structure can be seen in the group of amphibolites north-east of lake 530 m. This structure is the result of the interference of three phases of folding which are: 1) an isoclinal  $F_2$  folding, having a roughly horizontal axis and axial plane running along the middle of one of the amphibolites, 2) a tight  $F_3$  folding with NNE plunging axis (the axial surface of this folding is marked by migmatisation), and 3) an  $F_4$  NW open folding, characterised by a swarm of pegmatites along its axial surface, which affects the amphibolites in their south-western part and refolds the  $F_3$  and  $F_2$  axial surfaces.

The amphibolites lying on the southern limb of the synform by mountain 880 m are affected by an old isoclinal folding seen in the amphibolite of the southern of the lakes 615 m. The hinge of the isoclinal folding is strongly thinned. In the field its axial plane is steep. The axial plane trace of the mountain 880 m synform is roughly parallel to that of the  $F_2$  fold.

#### *Eastern field*

This field stretches from the valley north of Qíngua as far as the inland ice. In the western part of this field the folding seen in the preceding area and suggested as being of  $F_5$  age, continues with two main structures, the Qíngua antiform (having its NW limb in the preceding field) and the

Aorngo synform, and two less important structures in the northern part: the overturned antiform with SE dipping axial plane and the synform near the glacier (sub-area 17). Sub-areas from 14 to 18, concerning this western part, show NE to NNE trending  $\beta$  axes. Here too, as happened in many sub-areas of the preceding field, some minor fold axes, plot on the projections with a large angle to the  $\beta$  axes, while others lie parallel to it. On the other hand, minor structures of different ages from the main folding are common, and these cause a large spread of the foliation planes when the sub-area considered is large (see, for instance, sub-areas 14 and 16).

The most striking structure characterising the eastern field occurs in its central part and is marked by the long amphibolite horizon which passes north of mountain 993 m. This structure looks like a mushroom closed at the eastern end. The oldest recognisable folding entering this structure is a large scale  $F_2$  isoclinal folding which is particularly evident in the mountain 933 m amphibolite complex, immediately north and west of mountain 933 m. This folding causes clear repetitions in the stratigraphy. As is evident from the map (plate 11), its axial plane is clearly folded by a succession of major  $F_4$  folds which, together with the  $F_2$  folding, are the easier to see at first sight. The major  $F_4$  folds from west to east are:

1) An antiform with an almost rectilinear axial plane trace, passing from the south near lake 420 m to the NNW a little west of mountain 933 m. The fold axis generally plunges NNW at angles which vary from sub-area to sub-area (sub-areas 20, 23 and 26). The axis of this antiform probably extends towards lake 460 m, since small, carefully selected sub-areas give NW trending  $\beta$  axes, and minor NW folds are common near lake 460 m. In this case the axial plane appears to be folded and the large antiform immediately west of mountain 933 m dies out southwards.

2) A synform having an axial plane trace roughly parallel to the preceding one swinging slightly from NNW to NW. Its fold axis plunges towards the NW near mountain 933 m (sub-areas 24 and 25) and to the SE near lake 650 m. Sub-area 30 refers to this last part and shows that the normal to the foliation planes and minor fold axes plot on a girdle which is probably due to interfering folds. Several minor  $F_3$  folds have been seen in this area.

3) An antiform with axial plane trace diverging from the preceding ones. Its fold axis plunges steeply to the NNW near lake 690 m (sub-area 21) and, more gently, to the SSE in the southern part, with a culmination in the area of lake 750 m. Sub-area 31, which refers to the southern part, shows a large spread of data due to the superimposition of folds.

A common characteristic of these NW folds is that they are non-cylindrical. This is evident in the two westernmost  $F_4$  structures, which have clear variations in the plunge of the fold axes, while in the eastern one the variation is smaller.

Other folds belonging to this structure are present in the area of lake 300 m, south of Aorngo. They are a small synform and an antiform with a NE axial surface, probably affecting or affected by other folds, as is indicated by the spread of the foliation planes in their sub-areas (19 and 22). Their axial surface is not traceable with confidence towards the NE although in some sub-areas (27 and 28) a NE plunging  $\beta$  axis is present. These folds exhibit axial plane migmatisation, and are therefore  $F_3$  in age.

Minor plastic syn-migmatitic  $F_3$  folds occur all over the area such as the small-folded gneiss areas in the north-eastern and eastern part of this structure.

Towards the glacier the foliation shows some large-scale bucklings defining a NE plunging  $\beta$  axis (sub-areas 29) and believed to be of  $F_5$  age. The description of this structure suggests that its actual shape is the consequence of the interference of several phases of folding.

The following interpretation is proposed. Early  $F_2$  folds were refolded during the  $F_3$  deformation, now recognisable mainly on a mesoscopic scale but also by a few minor large-scale structures. Later on an intense  $F_4$  folding affected the  $F_3$  structures. In the area between mountain 933 m and lake 650 m the  $F_3$  folding was probably represented by an anticline having its axis roughly perpendicular to that of  $F_4$ . This is indicated by the opposing plunges to the north and south of the  $F_4$  axes and by the similarity of these folds with brachiformal structures. The whole picture was later complicated by the  $F_5$  deformation, which caused the bucklings near the glacier. Its effects westwards are doubtful, although  $F_5$  too probably enters the interference pattern, since they occur on a mesoscopic scale.

The structures on the peninsula partly surrounded by the inland ice, east of the area described above, look rather simple. Here the foliation planes have an almost constant NE strike and NW dip. They are thought to form the sout-eastern limb of a big synform whose axial plane runs approximately in the area of the rivers bordering the peninsula to the south-west. Unfortunately, the axial zone of this fold is completely hidden, thus it is difficult to establish its relative age. The plot of the normal to foliation planes gives a fairly good great circle girdle and indicates a NE  $\beta$  axis. Possibly, by analogy with what has been seen in the western part of the present area, it represents the last phase of deformation, although the possibility of an  $F_3$  age should not be excluded.

Two other structures are present in the peninsula: an inclined synform, closing near mountain 920 m, and another inclined structured

closing east of lake 680 m. They are tight to isoclinal folds with associated axial plane migmatisation. The closure of the fold east of lake 680 m is almost completely hidden by migmatisation. These folds do not appear to interfere with other structures, but their tight form, their axial plane migmatisation and their general style show them to be of  $F_3$  age.

### Brittle Deformations

Not much time has been spent in the field studying the brittle phenomena, but some observations, set out below, may help in further studies. There are three main sets of fractures:

- a) those with a general NE strike, but swings to NNE or ENE are common;
- b) those with a general SE strike, with variations towards a more easterly or southerly strike;
- c) those with a general SSE strike and variations trending N-S or SSW.

All these fractures generally form vertical planes and for the most part do not seem to displace the rocks they cross. A few, minor displacements were found along the dyke crossing the amphibolite synformally folded near lake 670 m, western area, and in the thick  $MD_2$  dyke and amphibolites of the central part of the eastern area.

The first NE fractures are associated with zones of mylonites characterised by large amounts of epidote, albite and chlorite. This fracture probably determined the formation of Qagssip kangerdluarssua and Qingua fjords. Later shearing of the mylonites, shearing along the  $MD_2$  dykes and the two sets of  $MD_2$  dykes indicate that these fractures were rejuvenated.

The second set of fractures was accompanied by the emplacement of the  $MD_3$  dykes. Its major effects include the formation of the Qagssit fjord, aligned along a fracture, continuing through a zone of rivers and lakes up to the inland ice, and the Nigerdlip qôrørssua valley, whose fracture is traceable up to the easternmost border. These fractures were accompanied by shearing of the rocks, but no large scale displacement is evident. It is possible that vertical movement occurred. The fractures are sometimes accompanied by hydrothermal veins, which mostly consist of almost pure epidote but which sometimes have a more complex composition. For instance, the veins crossing the thick  $MD_2$  dyke near lake 690 m and the other thick  $MD_2$  dyke slightly displaced by a SW fault 1800 m east-south-east of mountain 933 m, are formed by an assemblage of quartz, albite, epidote, sphene, amphibole, chlorite and

some apatite. The MD<sub>3</sub> dykes are sometimes sheared along the strike. This is quite clear in the north-west part of the long, thin MD<sub>3</sub> dyke which runs from Nunaqarfia through the whole area. Therefore the SE fractures have possibly undergone a rejuvenation.

The third set of fractures in the present area is not very important. Nevertheless it again produced shearing and epidotisation. It is worth noting here that the TD dyke of the easternmost peninsula has exceptionally been sheared by a fracture belonging to this set.

### Conclusions

The first recognisable event consists of a layering or banding (S<sub>0</sub>) which is of pre-metamorphic origin. S-surfaces, which may be confused with S<sub>0</sub>, were formed by early metamorphic differentiation. This layering was deformed by ductile flow which formed the now intrafolial folds, of which there might be more than one phase. The first folding (F<sub>2</sub>) recognisable on both mesoscopic and a macroscopic scale is isoclinal and caused repetitions of the stratigraphy. The same deformation which produced the isoclinal folding was also responsible for a transposition of the earlier S<sub>0</sub> to a new foliation (S<sub>1</sub>) which lies parallel to the axial plane of the isoclinal folds. Metamorphic differentiation (P<sub>1</sub> pegmatites) and boudinage represent other features of this period. This folding is largely obliterated by the subsequent migmatisation and has a relic appearance. The subsequent F<sub>3</sub> folding indicates a highly plastic environment (p. 80) and was syntectonically accompanied by migmatisation. Large scale F<sub>3</sub> structures occur all over the area.

The environmental conditions during the following F<sub>4</sub> phase were characterised by a lower degree of plasticity, as is suggested by the shape of these folds, though the F<sub>4</sub> deformation still has a plastic character. Large scale F<sub>4</sub> folds enter some of the largest structures of the eastern field (p. 86), while they are little represented to the west. In adjacent areas (S. B. JENSEN, private communication) field examination has shown that a significant break, during which sediments were deposited, intervened between the F<sub>3</sub> and F<sub>4</sub> phases.

The F<sub>5</sub> folding took place in relatively brittle conditions, as is shown by the fracturing along the axial plane. The most prominent structures occurring all over the area are thought to be of F<sub>5</sub> age.

Subsequently basic dykes were intruded and fractures formed. These last produced zones of weakness easily eroded by the glacial activity and thus largely influenced the actual topography of the area.

## RELATIONSHIPS BETWEEN METAMORPHISM, PETROGENESIS AND STRUCTURE

The correlations thought to exist between metamorphic and structural events are summarised in Table 16.

It has already been stated that the area most probably consisted of a geosynclinal series which at first underwent a period of increasing metamorphism, as shown by a few relic mineral assemblages. This early period of metamorphism, during which an early metamorphic differentiation occurred (p. 78) may possibly be related to the  $F_1$  phase (or phases) of folding, but not enough is known about this period to establish definite relationships.

The effects of the  $F_2$  deformation (transposition of the  $S_0$  to the  $S_1$  gneiss foliation, formation of  $P_1$  pegmatites, boudinage) indicate a possible relationship to the period when migmatisation began, i.e. in a metamorphic grade of the low granulite facies (almandine-cordierite-biotite subfacies, p. 71).

The period of 'quartz (or grano-) dioritisation', where the effects of migmatisation are most marked, is clearly synchronous with the  $F_3$  deformation, as is shown by the relationships between  $F_3$  folds and  $P_2$  pegmatites and by the recrystallisation of hornblende elongated along the  $F_3$  fold axis. Also the increase in the water pressure (p. 71) is in good agreement with the highly plastic style of the  $F_3$  folds. Therefore  $F_3$  must have occurred in the high almandine-amphibolite facies.

Because of its relationships with the  $P_3$  pegmatites (p. 82) the  $F_4$  deformation can be related to the period of microcline porphyroblast formation i.e. with a medium to low amphibolite facies metamorphism. The style of  $F_4$  is less plastic than that of  $F_3$  and this is in agreement with a lower metamorphic grade. The transformation found in a rusty schist with an anorthite-anthophyllite assemblage into a hornblende-plagioclase assemblage, with hornblende aligned along the  $F_4$  fold axis, again indicates that the  $F_4$  deformation occurred with decreasing metamorphic conditions, but within the almandine-amphibole facies.

The  $F_5$  phase does not seem to have been associated with important metamorphism, but some low-grade transformations, quartz mobilisations (with consequent quartz veining) are, however, ascribable to this phase,

Table 16. Relationships supposed to exist between metamorphic and structural events.

| Metamorphic facies                                     | Petrogenesis  | Tectonism   |
|--|---|---|
| greenschist facies                                     | intrusion of doleritic dykes  | brittle deformations  |
| epidote-amphibolite facies                             | local retrogression connected with faulting                           | $F_4$ semi-brittle folding.                                     |
| medium to low almandine-amphibolite facies             | weak general retrogression. $P_4$ quartz mobilisates.                 | $F_4$ folding. $P_3$ pegmatites parallel to the axial plane     |
| high almandine-amphibolite facies (si-al-mu subfacies) | microcline blastesis. $P_3$ mobilisates. Biotitisation of hornblende. | $F_3$ synmagmatic folding.                                      |
| decreasing metamorphism                                | migmatisation   | Scattered presence of $S_1$ .                                   |
| low hornblende-granulite facies (al-co-bi subfacies)   | $P_2$ mobilisates   | $F_2$ folds. Axial plane structures.                            |
| ?  | quartz (grano-) dioritisation   | Formation of the regional $S_1$ .                               |
| increasing metamorphism                                | relic assemblages (?anatexis)   | Boudinages  |
| staurolite-almandine subfacies                         | $P_1$ mobilisates   | $F_1$ intrafolial folds, including probably more than one phase |
| ?  | ?   | geosynclinal series   |
|  |   | $S_0$   |

and its semi-brittle conditions correspond well with a low metamorphic grade. The scattered sericitisation and epidotisation, even of oligoclase, and the chloritisation of the mafics are its most evident effects. The further evolution linked with brittle deformation formed only mylonite zones or local retrogressions associated with laminations and the occurrence of some hydrothermal activity.

Although no absolute age determination has been carried out on the rocks of our area, the metamorphic and tectonic history previously described is considered to be pre-Ketilidian (except for the last swarms of basic dykes, see p. 73).

It is known that the undeformed MD dykes belong to those generations of dykes which in the Ivigtut area, 125 km to the south, were transformed during the Ketilidian deformations. The gneiss complex in the area is therefore pre-Ketilidian and is at least 1800 m.y. but may be even as much as 2500 or 3000 m.y. old, as these ages have been recorded from similar rocks in the Neria and Fiskenæsset areas, 80 km to the south and 100 km to the north respectively. (LARSEN, 1966; LARSEN & MØLLER, 1968; LAMBERT & SIMONS, 1969).

## ACKNOWLEDGEMENTS

The authors wish to express their gratitude to the Director of Grønlands Geologiske Undersøgelse, mag. scient. K. ELLITSGAARD-RASMUSSEN for his permission to publish this paper, and to cand. mag. S. B. JENSEN, who was the leader of the field parties in the Mellembygd area.

Thanks are due to Mr. S. B. JENSEN and Mr. T. C. R. PULVERTAFT for their very valuable criticism of the first draft of this paper. Many of their suggestions have been taken into account, but responsibility for the final draft rests with the authors.

It is with pleasure that we extend our thanks to the whole research staff of the 'Istituto di Mineralogia e Petrologia' of Modena University, and in particular to the Director, Professor GLAUCO GOTTARDI, and to Professor BERTOLANI, Dr. SIGHINOLFI and Dr. CAPEDRI for their valuable advice and help.

Mr. S. ENGELSTOFT and Mr. W. THOMSEN, who assisted us during the field work, the helicopter pilots Mr. F. CARDINAUX, Mr. M. BURKHARD and Mr. J. P. FÜLLEMANN, and Mr. L. H. NORRIS who helped with the correction of the English manuscript, are also gratefully acknowledged.

Mrs. M. ALBERTI is thanked for her typing of the manuscript.

Istituto di Mineralogia e Petrologia  
Via S. Eufemia 19  
41100 Modena  
Italia

## APPENDIX

### Analytical methods of chemical and mineral analyses

The following methods of chemical analyses were used.

Si, Al, Ti were determined by silicate rapid analysis according to the methods of SHAPIRO & BRANNOCK (1956).

Mn, total Fe, Mg, Ca, Na, K were determined by atomic absorption spectroscopy using a Perkin Elmer 303 spectrophotometer. The procedure is described by ALTHAUS (1966).

Fe<sup>2+</sup> was determined by semi-microanalysis according to MEYROWITZ (1963).

The modal analyses were carried out using a J. S. Swift & Son point counter.

Plagioclase compositions were determined either by means of the universal stage, using the curves published by EMMONS (1959), or by measuring the refractive index by means of an Emmons' stage with monochromatic sodium light and variable temperature.

The optical characters of pyroxenes and amphiboles were determined by means of the universal stage.

Olivine composition was determined by X-ray diffractometry following the method and using the curves published by YODER & SAHAMA (1957).

### Sample localities

Maps showing the position of the samples mentioned in the text are available for inspection in the archives of the Geological Survey of Greenland.

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## PLATES

### **Plate 1**

Plate 1a. Saw-like absorption contacts between quartz and plagioclase (sample 57724),  
Crossed nicols,  $\times 45$ .

Plate 1b. Substitution of plagioclase by microcline (sample 57873). Crossed nicols,  
 $\times 30$ .



1 a

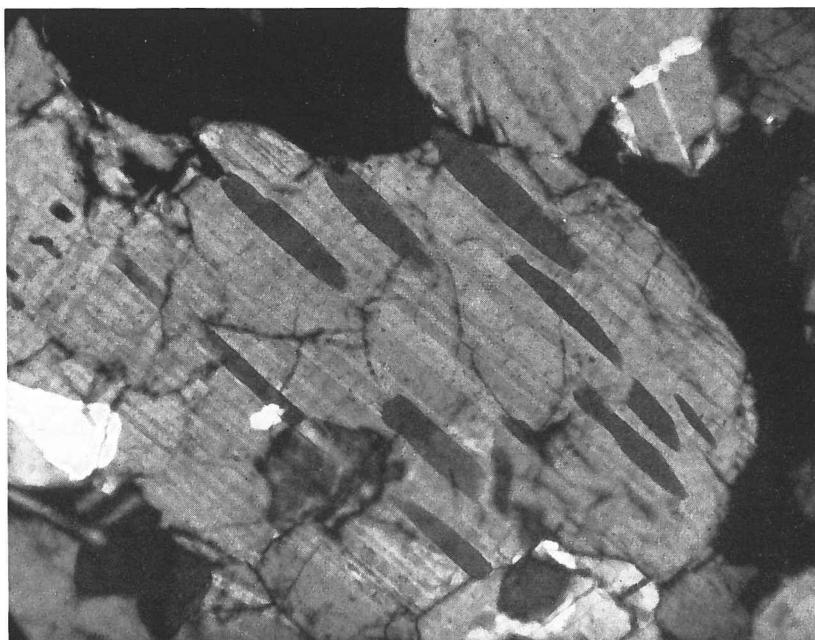


1 b

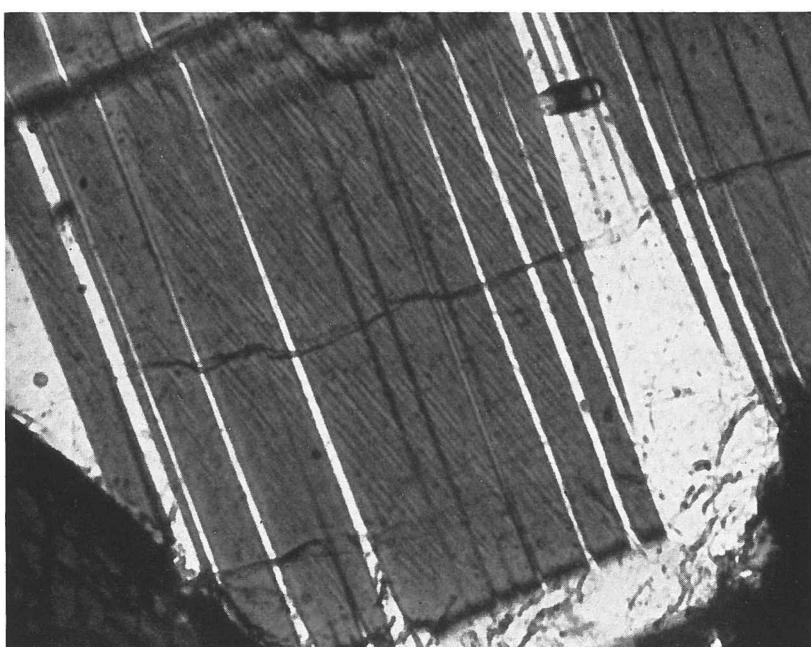
## **Plate 2**

Plate 2 a. Antiperthites in the homogeneous gneiss west of mountain 780 m (sample 57756). Crossed nicols,  $\times 45$ .

Plate 2 b. Microantiperthites in calcic plagioclase from an amphibolite (sample 73858). Crossed nicols,  $\times 120$ .



2 a

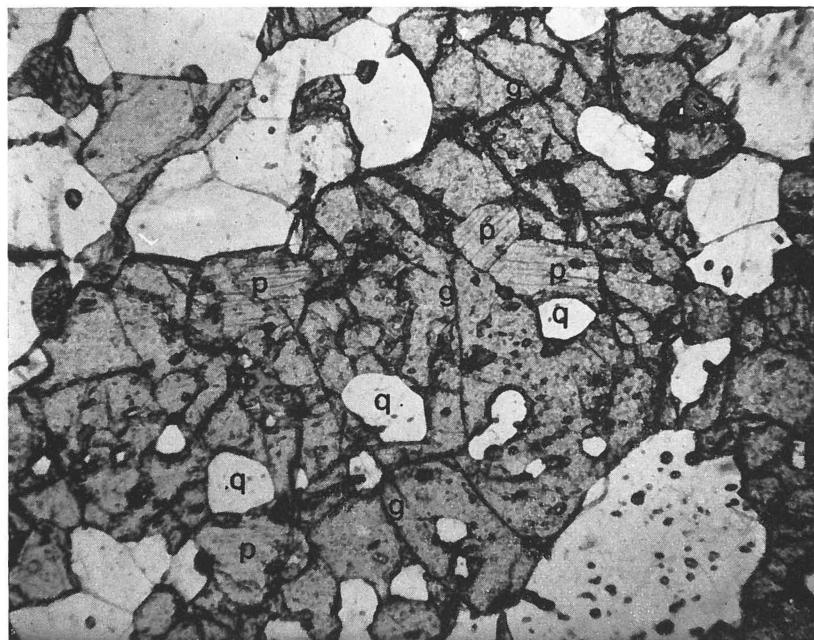


2 b

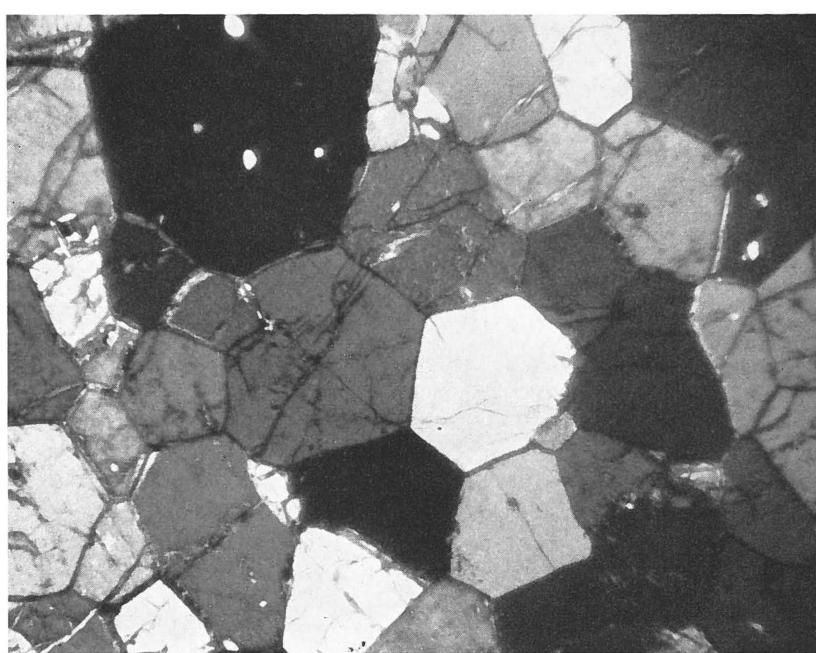
### Plate 3

Plate 3a. Garnet (g) replacing clinopyroxene (p). Sphene (s), hornblende (h) and quartz (q) are also formed in the reaction (sample 73819). Plane polarised light,  $\times 30$ .

Plate 3b. Polygonal equilibrium texture in a peridotite (sample 73566). Crossed nicols,  $\times 15$ .



3 a

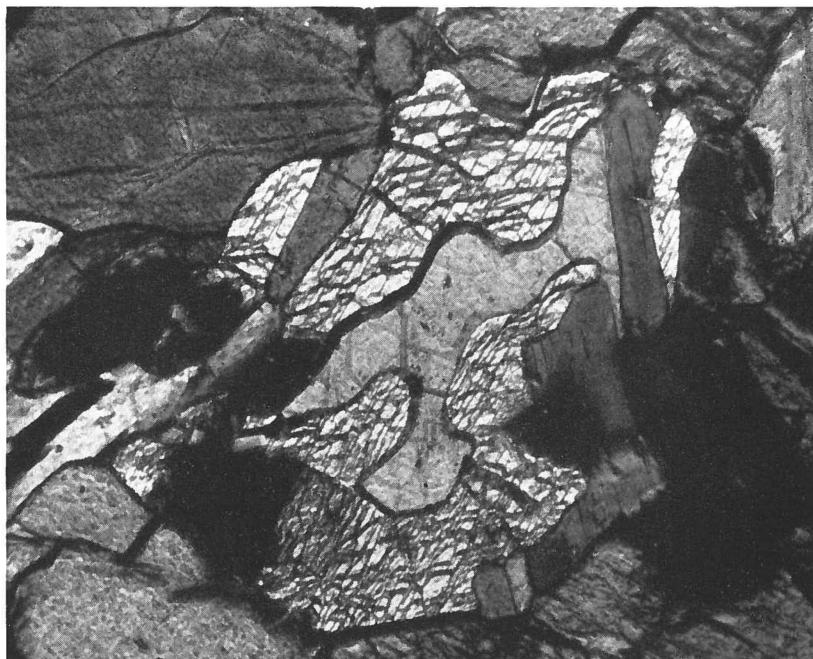


3 b

#### **Plate 4**

Plate 4a. Amphibole partially altered to carbonate in an ultramafic lens (sample 73839). Crossed nicols,  $\times 40$ .

Plate 4b. Irregular zoning in a plagioclase from a rusty schist. Some zones are microantiperthitic and have  $An_{80}$ , others are not microperthitic and have  $An_{60}$  (sample 75898). Crossed nicols,  $\times 90$ .



4 a

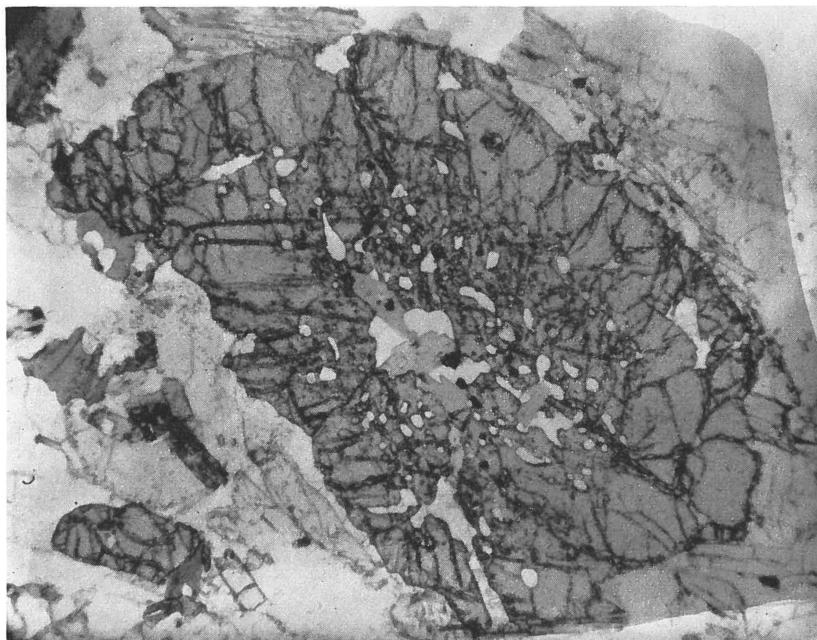


4 b

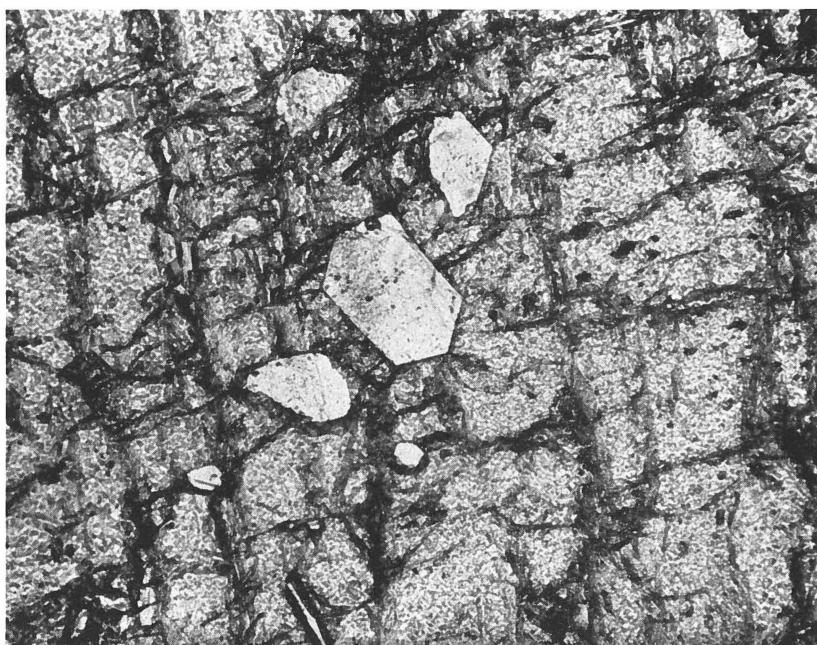
### **Plate 5**

Plate 5a. Atoll structure in garnet. Note the poikiloblastic centre and the compact outer zone (sample 58884). Plane polarised light,  $\times 20$ .

Plate 5b. Pseudo-idioblastic feldspar in garnet (sample 73590). Plane polarised light,  $\times 45$ .



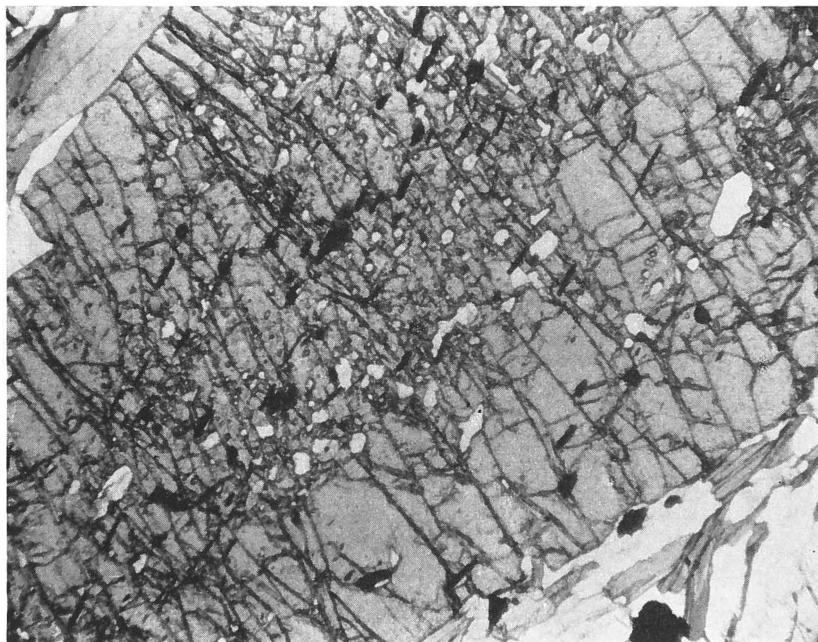
5 a



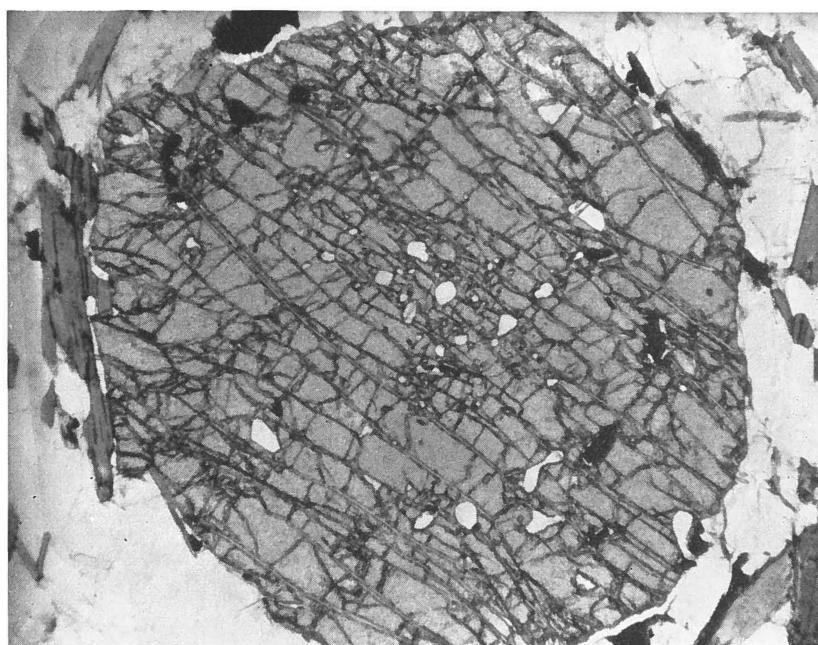
5 b

### **Plate 6**

Plate 6 a & 6b. Ilmenite in garnet: plate 6a) aligned parallel with the foliation of the rock; plate 6b) with concentric zonal distribution (sample 57820). Plane polarised light,  $\times 30$  and  $\times 20$ .



6 a

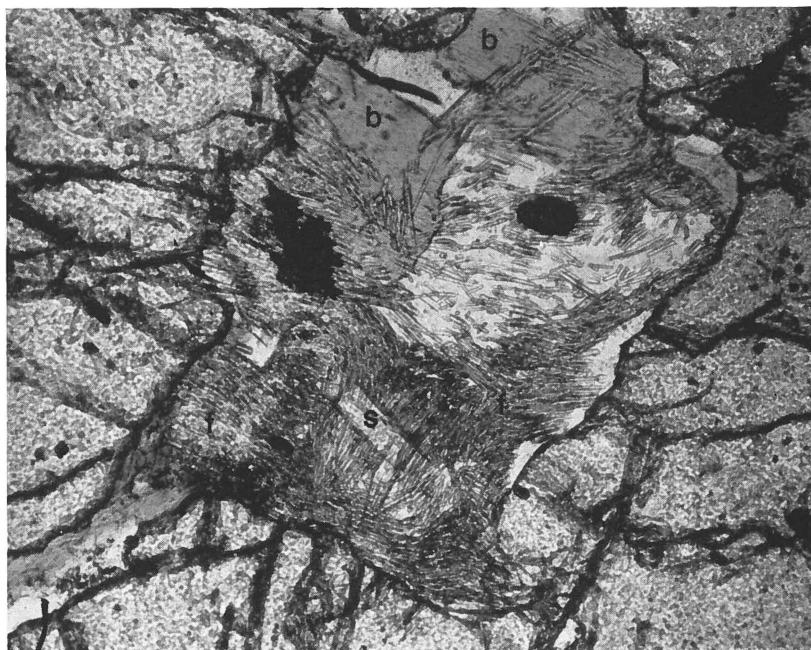


6 b

### Plate 7

Plate 7a. Feldspar, sillimanite and fibrolite occurring as the inner part of an atoll garnet. Fibrolite crystallisation post-dates sillimanite. (s = sillimanite, f = fibrolite, b = biotite; sample 73590). Plane polarised light,  $\times 45$ .

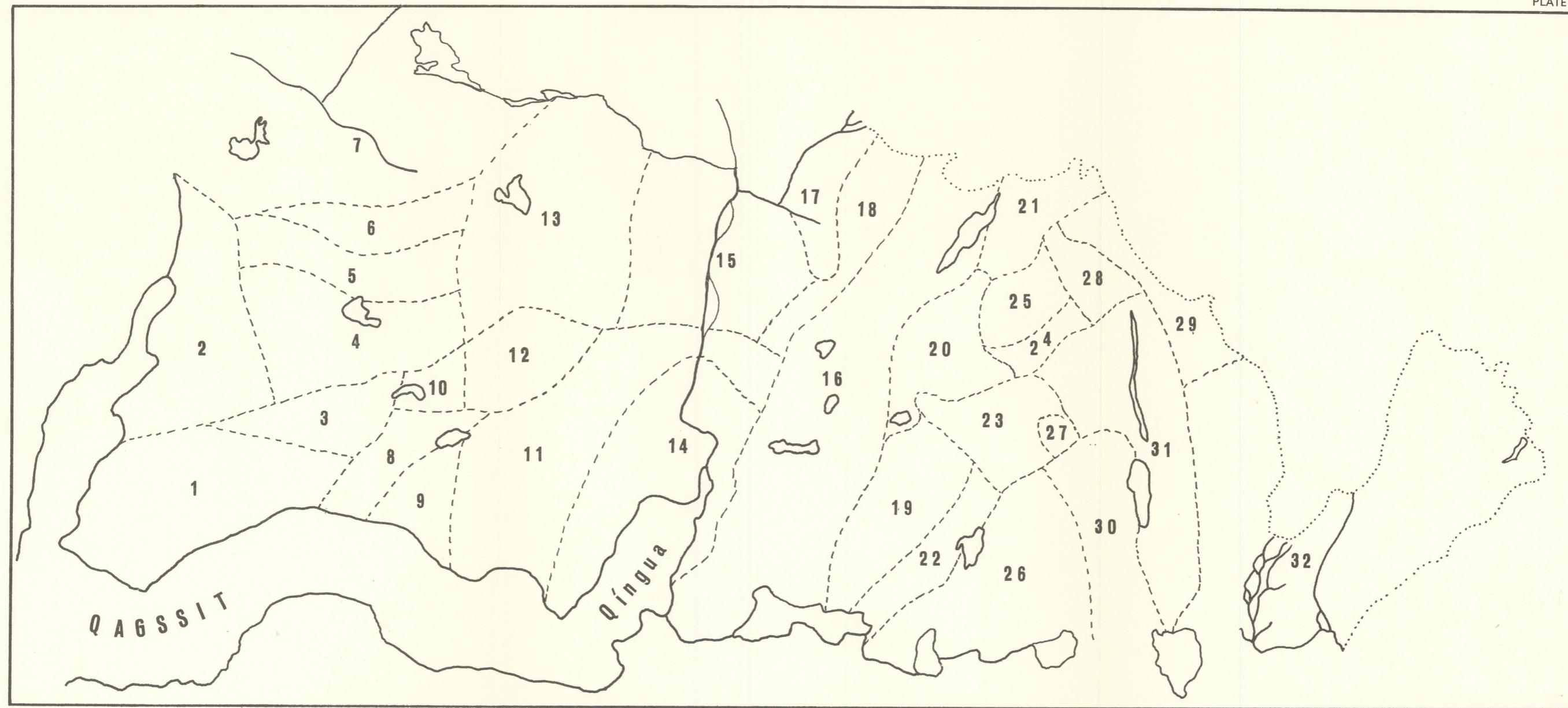
Plate 7b. Quartz-epidote symplectites at the contact between hornblende and plagioclase in an amphibolite inclusion in the gneisses (sample 57721). Crossed nicols,  $\times 20$ .



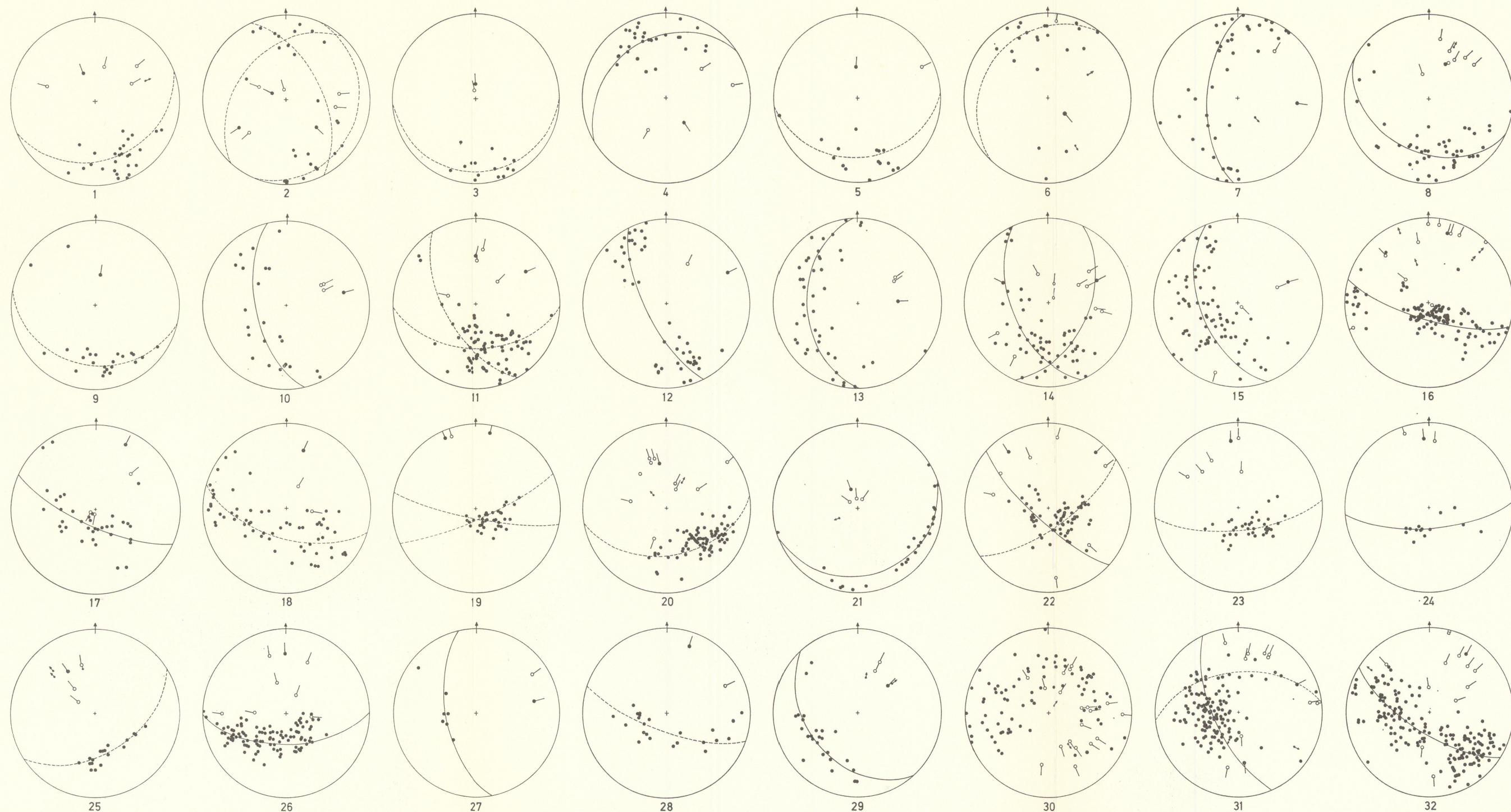
7 a



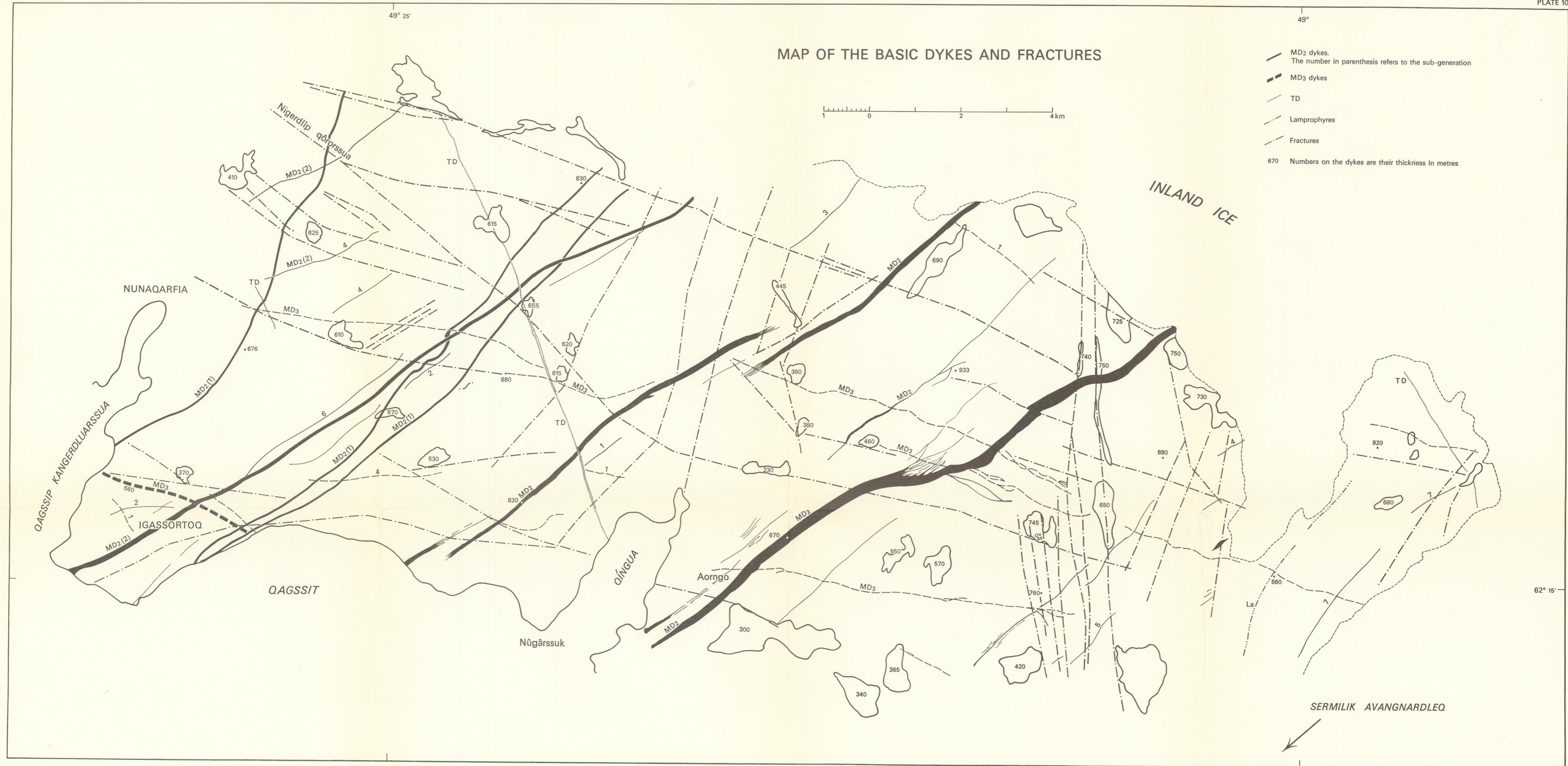
7 b



The sub-area fields into which the area has been divided.



Plots for each sub-area of the normal to foliation planes (dots), lineations (arrows), mesoscopic fold axes (circles with a tail), constructed  $\beta$  axes (dots with a tail).  
Lower hemisphere, Lambert's equal-area projection.



GRØNLANDS GEOLOGISKE UNDERSØGELSE

THE GEOLOGICAL SURVEY OF GREENLAND

MEDDELELSE OM GRØNLAND BD. 192, NR. 5 (G. RIVALENTI AND A. ROSSI).

PLATE 11

